# GEOLOGICAL SURVEY OF KENTUCKY.

JNO. R. PROCTER, DIRECTOR.

# CHEMICAL REPORT

OF THE

# COALS. SOILS. CLAYS, PETROLEUM,

MINERAL WATERS ETC., ETC.,
OF KENTUCKY.

BY ROBERT PETER, M. D., ETC., ETC.,

CHEMIST TO THE SURVEY,

ASSISTED BY ALFRED M. PETER. S. M.

THE SEVENTH CHEMICAL REPORT IN THE NEW SERIES. AND THE ELEVENTH SINCE
THE BEGINNING OF THE SURVEY

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## INTRODUCTORY LETTER.

CHEMICAL LABORATORY,
KENTUCKY GEOLOGICAL SURVEY,
LEXINGTON, KY., April 17, 1888.

JNO. R. PROCTER, Esq.,

Director of Geological Survey of Kentucky:

DEAR SIR: I have the pleasure herewith to send you for publication, my report of the chemical work done in this Laboratory for the Geological Survey since the publication of my last report.

Very respectfully,

ROBERT-PETER, M. D., ETC., Chemist to Kentucky Geological Survey. This page in the original text is blank.

# CHEMICAL REPORT.

Of the more than 273 analyses reported in the following pages, 137 are of coals from 14 counties in the eastern coal field of Kentucky; the greater number of them being from Bell, Breathitt, Clay, Harlan, Leslie, Perry and Whitley counties; a comparatively few samples are from Johnson, Knott, Knox, Letcher, Laurel and Martin counties. In the Appendix, West Virginia is represented in the analyses of seven samples, for comparison with Kentucky coals, together with the analyses of many other coals received since this report was made.

Most of these coals would come under the head of semibituminous, splint, or block coal; some few are, no doubt, entitled to the name "bituminous coal," and some of them are cannel coals. With them are reported some with such high ash percentage as to cause them to be classed with the bituminous shales. These, however, are comparatively few in number, and most of them could be used for fuel in their immediate vicinity. Generally speaking, these coals are of very good quality, and many of them would answer admirably for the manufacture of coke, while some of the splint or block coals could, probably, be used without coking for smelting iron ores. In short, the coals of this, one of the two immense coal fields of Kentucky, will generally compare favorably with those of any other region in the world.

Comparing the ash percentages of the coals of the seven counties which furnished the greatest number of samples, we find the following results, viz.

Counties.	No. of samples analyzed.	Highest ash per- centage.	Lowest ash per- centage.	Average ash of the whole.	Average exclusive of highest ash per- centage.
Whitley	18	7.90	1.54	3.56	none excluded.
Bell	11	8.40	1.80	4.28	none excluded.
Harlan	44	47.48	2.20	7.76	5.12 excluding 4 highest
Leslie	11	16.00	4.00	8.19	7.40 excluding 1 highest
Breathitt .	10	24.20	5 28	9.32	8.08 excluding 1 highest
Clay	9	32.00	5.80	9.79	7.00 excluding 1 highest
Perry	13	16.80	2.56	9.88	8.48 excluding 1 highest

Of course such a comparison as this, of a limited number of samples, might not apply to the products of the whole of the several counties in question, as other samples from coal beds not here represented might alter these averages materially, and it is proper to state that those coals or bituminous shales which give very large ash, should be excluded from such calculations. Moreover, it is generally found that cannel coals—which variety abounds in Breathitt county particularly—very generally give more ash than the softer coals.

The West Virginia coals gave the following results, viz.: The 7 samples gave—highest ash, 12.76 per cent.; the lowest, 2.40 per cent; the general average of their ash, 4.94 per cent.

In all these coals the sulphur ranges from 6.042 per cent. in what is really bituminous shale, of Clay county; 5.436 in this kind of shale from Harlan county, and 5.078 in cannel coal of Bell county, down to 0.420 in Harlan county coal, and 0.418 per cent. in coal of West Virginia. Generally the percentage of sulphur is comparatively moderate or small in these coals.

The thirty-nine Cokes analyzed were principally from Bell, Harlan and Knox counties, there being but one sample each from Laurel, Letcher and Whitley counties. West Virginia furnished 8 samples.

The 20 samples of coke from Bell county gave an average ash percentage of 7.83, but without the highest four, which would be excluded from this manufacture, the ash was only 6.97 per cent, the highest being 12.2 and lowest only 4.00; which compares favorably with the best cokes in the market. The percentage of carbon in the coke varied from 95.80 in F. Barner's to 85.30 in Jas. Meyers'.

The average ash of the 4 Harlan county cokes was 9.10 per cent. But excluding the highest, 17.9, the ash average is only 6.20 per cent. The 4 Knox county cokes gave an average of 5.20 per cent. of ash, including the highest, 8.40. The lowest being only 3.20. The 8 West Virginia cokes gave a general average of 6.47 per cent. of ash, including one at 9.10 per cent.

The sulphur is moderate or small in all these cokes which do not contain too much ash, ranging from 1.975 to 0.217 per cent.

Of the 51 soils, the analyses of which are herewith reported, 12 are from Ballard county; 15 from Calloway; 2 from Graves; 11 from Marshall, and 10 from McCracken; in all, 50 from the so-called Jackson Purchase; one only is from Jefferson county. Five other soils, from Clinton county, are reported in the Appendix.

The Jackson Purchase soils and subsoils, with the exception of those from the bottom lands of the Ohio, Obion and Tennessee rivers and Shannon creek, and the Oak and Hickory Flats, all contain but a small or moderate proportion of organic matters or humus. This is, probably, due to the circumstance that they, with the exceptions above noted, generally contain quite a large quantity of sand and insoluble silicates in proportion to their alumina. The exceptional soil, No. 2622, from the first bottom of Tennessee river, Calloway county, gave but 77.719 per cent. of siliceous residue; while the red sandy soil of Sandy Ridge, No. 2565, gave 93.385 per cent., the general range being from about 81. to 91. per cent.

This rather large proportion of sand, although it is very fine sand generally, and moderate proportion of alumina, cause these soils to be light and porous, easily worked or drained under favorable conditions.

Lime and phosphoric acid are small, or only moderate in quantity, in several of them, which may render necessary the use of calcareous and phosphatic fertilizers, but the potash is usually in good proportions; a few exceptions only being noticeable. This fact, as well as their light porous condition, allowing free penetration of the atmospheric agencies, with the absence of gravel or coarse sand, aids in making these soils favorable to tobacco culture.

All of the 26 clays, etc., reported, are from the Jackson Purchase. Six from Ballard county; 8 from Calloway; 5 from Graves; 1 from Hickman; 6 from Marshall, and 5 from McCracken county; mostly from the Tertiary and Quaternary formations. A few of the most siliceous or sandy of them are derived from decomposing chert or hornstone of the Lower Carboniferous siliceous group.

Many of them are highly refractory, and would make good fire-brick or furnace linings, or, possibly, glass pots. Some

are white enough for fine pottery-ware. Some, composed mainly of very fine sandy material, would answer to mix with tough clays, or probably for glazing material or glass, or for the manufacture of water cement. Several of these siliceous clays have such a fine texture that they may be used as scouring or polishing material.

Some of them, containing much iron peroxide, may find their applications as cheap paints, for terra-cotta work or cheap pottery-ware.

The 3 iron ores from Bath county belong to the Clinton group, and present the general characteristics of the so-called Clinton Iron Ore.

# BALLARD COUNTY. MINERAL WATERS.

No. 2554—MINERAL WATER, from the Kilgore Spring at Blandville: temperature (in August) 58° F. Slightly chalybeate. Collected by R. H. Loughridge, August 19th, 1885. Sample in stone-ware gallon jug. Cork sealed.

Evaporated to dryness, it left 0.0674 per 1000 of saline matters, which lost 0.0030 of organic matter on ignition.

The saline matters consisted of chlorides, carbonates and sulphates of calcium, magnesium, potassium, sodium and iron, silica and a trace of Lithium. They are slightly alkaline in reaction.

The water is a good saline chalybeate.

No. 2555—MINERAL WATER, from the McGee Spring, Blandville. Strongly chalybeate. Temperature 60° F. (in August). Collected by R. H. Loughridge, August, 1885.

# COMPOSITION IN 1,000 PARTS.

Carbonate of iron .																					
Carbonate of lime			8			٠														. [	.0108
Carbonate of magne	eie	Ł	1																	. 1	.0019
Chloride of magnes	iur	n																		.	. 0099
																					.0053
Carbonate of soda				 																.	. 0316
Sulphate of potash					•	•														.	.0072
Silica		•	•			٠	٠	•		•	٠	•	•8	•	٠	•	٠	٠	•		.0156
Total solid matt	er	s	- 54		٠		8 3											ė			0.1067

It contains also traces of organic matters and ammonia. It appears to be a good alkaline saline chalybeate water.

#### BALLARD COUNTY.

#### Soils and Subsoils.

No. 2556—Soil. Crawfishy, on west fork of Mayfield creek, taken to the depth of 10 inches, near the bridge on the road from Blandville to Milburn. Collected by R. H. Loughridge, 1884.

The dried soil contained small friable clods, is of a buff-grey color. The coarse sieve\* removed only a few small fragments of stone. Its siliceous residue, after digestion in acids, all passed through the fine sieve, + except 16.5 per cent. of fine white quartz sand.

No. 2557—Virgin Soil. Ohio river bottom. Taken 8 inches deep from Clear Lake Ridge, elevated a few feet above the cypress swamps, and having a growth of white oak, hickory, gum, prickley ash, walnut and black locust. Undergrowth of cane. Four miles west of Barlow. Geological formation: Quaternary brown loam table lands. Collected by R. H. Loughridge, 1884.

Dried soil in small friable clods of a dark grey-brown color. All passed through the coarse sieve, except a small quantity of vegetable debris. Its siliceous residue, from digestion in acids, all passed through the fine sieve, except a small quantity of fine white sand.

No. 2558—Virgin Soil. Obion bottom. South of Arlington. Taken to the depth of 10 inches. Collected by R. H. Loughridge, 1884.

Dried soil in friable clods of a dark grey-brown color. All passed through the coarse sieve except a little vegetable debris. Its silicious residue all passed through the fine sieve.

No. 2559—Virgin Upland Soil. Brown loam from the Barrens or tobacco lands, one mile south-east of Hazlewood post-office. Taken to the depth of 10 inches. Timber: red

<sup>\*</sup> With 64 meshes to the centimeter square.

<sup>†</sup> With 1600 meshes to the centimeter square.

and black-jack oaks. Geological formation: Quaternary brown loam table lands. Collected by R. H. Loughridge, 1884.

Dried soil contains friable clods of a grey-brown color. All passed through the coarse sieve except a very small quantity of small shot-iron ore. Its siliceous residue all passed through the fine sieve.

No. 2560—Subsoil of the next preceding, taken to depth of 6 to 14 inches below the surface. Collected by R. H. Loughridge, 1884.

Dried subsoil in friable clods of a brownish-buff color. All passed through the coarse sieve except a few particles of shot-iron ore. Its silicious residue all passed through the fine sieve.

No. 2561—Virgin Soil. Flatwood loam soil. Bandana postoffice. Taken to the depth of 10 inches. Timber: red, post
and white oaks, with some hickory, gum, persimmons, sycamore and hazel brush. Collected by R. H. Loughridge,
February 25th, 1884.

Dried soil with friable clods of a light-yellowish umber color. The coarse sieve separated from it some little shot-iron ore and a few small quartz pebbles All its silicious residue passed through the fine sieve.

No. 2562—Subsoil of the next preceding, taken at from 10 to 14 inches below the surface. Collected by R. H. Loughridge, The dried subsoil is in moderately firm clods of a yellowish-grey color. All passed through the coarse sieve except very few particles of shot-iron ore. Its silicious residue all passed through the fine sieve.

No. 2563—VIRGIN SOIL. Dark loam from Barlow. Taken to the depth of 8 inches. Timber: white oak, poplar, gum and sassafras. Geological formation: Quaternary brown loam lands. Collected by R. H. Loughridge.

The dried soil mostly in powder with some small friable clods of a brownish-umber color. All passed through the coarse sieve except a little shot-iron ore. Its silicious residue all passed through the fine sieve.

No. 2564—Subsoil of the next preceding, taken from 8 to 10 inches below the surface. Collected by R. H. Loughridge.

The dried subsoil is in pretty firm small clods, of a dirty yellowish color. The coarse sieve removed from it only a little vegetable debris. All its silicious residue passed through the fine sieve.

No 2565—RED SANDY SOIL of Sandy Ridge, 3 miles west of Ogden's store. Sample taken 10 inches deep. Timber: chiefly red oaks. Collected by R. H. Loughridge.

Dried soil of a light snuff-brown color. No clods. All passed through the coarse sieve except a small quantity of shot-iron ore. The fine sieve removed from it 44.65 per cent. of fine sand in rounded grains.

No. 2566—Virgin Dark, Bluff Loam Soil, near mouth of Mayfield creek, at Steam Shovel. Taken 8 inches deep. Growth: poplar, white oak and hickory. Collected by R. H. Loughridge.

Dried soil of a grey-brown color, containing but a few friable clods. Its silicious residue all passed through the fine sieve.

No. 2567—Subsoil of the next preceding bluff loam, taken from 8 to 12 inches below the surface.

The dried subsoil, of a brownish-buff color, is mostly in the form of small friable clods. All passed through the coarse sieve except a little vegetable debris. Silicious residue all passed through the fine sieve.

COMPOSITION OF THESE BALLARD COUNTY SOILS, ETC.

Calculated Dried at 212° F.

NUMBER IN REPORT.	2556	2657	2558	2559	2560	2561	2562	2563	2564	2565	2566	2567
Organic and volatile matters. Alumina and manganese oxide Iron peroxide Lime carbonate Magnesia Phosphoric acid (P <sub>2</sub> O <sub>5</sub> ) Potash extracted by acids Soda extracted by acids Water expelled at 380-400° Sand and insoluble silicates.	3.306 2.946 1.860 .196 .196 .008 .477 .389 .566	6.772 6.876 4.124 210 .273 .226 .728 .728 .145 1.263 80.999	4.635 4.217 1.915 1.463 267 .133 504 not est. 941	3 096 3.065 1.907 096 .134 .062 .420 not est. .571	2.079 4.613 2.478 .071 .202 .062 .062 .836 not est. .425	3.642 4.639 2.863 .071 .198 .103 .248 .842 .842	2.114 4.523 3.077 046 .844 .115 .115 .136 .648	3.510 8.906 2.112 .146 .289 .100 .419 .260 1.038 88 394	2.864 4.602 2.870 .086 .222 .126 .674 not est. .698 .88.766	1.961 1.907 1.403 .071 .094 .094 .260 .286 .563	3.783 2.130 1.519 .349 .175 .096 .265 .067 1.604	1.281 3.268 2.224 2.224 .146 .284 .078 .028 .958
Totals	100.534 100.611		100.981	100.041	100.975	100 353	100 291	100.174	100.408	99 984	100.280	99.888
Hygroscopic moisture Potash in silicious residue Soda in silicious residue Sand	1.100 1.497 .884 16.500	2.29U 1.069 .277 not est.	1.625	1.175 1.476 not est.	1.180	1.480	1.375 1.691 1.551	1.300	1.235 1.389 not est.	0.600 .997 .667 44.650	1.300	0.900 2.081 .927

The virgin soil, No. 2557, of the Ohio river bottom, is the richest in essential elements of fertility of all these soils, and ought to be very productive under good culture, provided it is well drained. Its proportion of available potash, 0.773 per cent., is extraordinary. It is exceeded in this particular, however, by No. 2560, which has 0.836 per cent. All of these soils contain more than the usual average of potash, except this subsoil, No. 2567, which has only 0.109 per cent., which, other conditions being favorable, fits them for the cultivation of tobacco, hay or other green crops. Nos. 2559-60-1-2-4-5, are more or less deficient in lime, and Nos. 2556-9-60-5-6-7, also contain too little phosphoric acid, and would be benefited by top dressings of lime and of commercial phosphatic fertilizers. 2556 is especially deficient in phosphoric acid, while it contains the other essential elements in good or large proportions. It would require only phosphatic manures, such as ground-bone, superphosphate, guano, etc., to make it quite productive; provided drainage and other physical conditions are present. of these soils are of fine texture; none contain coarse sand or gravel. They all have potash in more than average proportions, except No. 2567, which, indeed, contains what is generally considered a fair average quantity.

## CLAYS OF BALLARD COUNTY.

No. 2568-SANDY CLAY. Tertiary. Half a mile north-west of Blandville. Collected by Jno. R. Procter.

Nearly white. Quite plastic. Contains no appreciable coarse sand. Infusible before the blow-pipe. Calcines white.

No. 2569—CLAY. Three miles east of Blandville. A nearly white clay. Collected by R. H. Loughridge.

Quite plastic. Contains no coarse sand. Of very difficult fusion before blow-pipe. Calcines white.

No. 2570—CLAY. North side of west fork of Mayfield creek, 4 miles north-east of Milburn. Collected by R. H. Loughridge.

Of a very light buff-grey color. Quite plastic. Contains no coarse sand. Infusible before the blow-pipe. Calcines light-grey.

No. 2571—CLAY. Two miles north of Wickliffe. Collected by R. H. Loughridge.

Nearly white. Contains much fine white opaque sand. Quite plastic. Infusible before the blow-pipe. Calcines white.

No. 2572—Yellow Ochreous Clay. Wickliffe, Ballard county. Collected by R. H. Loughridge.

Of a handsome yellow ochre color. Washed in water it left a very small proportion of very fine sand. Before the blowpipe it fuses into a blackish slag or glass. Calcines of a handsome red color.

No. 2573—FIRE-CLAY. Wickliffe, Ballard county. Collected by R. H. Loughridge.

Of a dark-grey color when dried. Contains fine sand, brownish, mixed with a few small specks of mica. Before the blow-pipe, of very difficult fusion. Calcines white.

COMPOSITION OF THESE BALLARD COUNTY CLAYS.

Air Dried.

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Number in Report.	2568	2569	2570	2571	2572	2573
Silica	74.840· 16 580	71 180 20.800	76 540 14.820	63.840 26.040	44.840 22.830	73.240 15.760
Iron peroxide .	1.400	1.780	.960	.740	20.350	1.920
Lime	.269	trace.	trace.	trace.	.101	.325
Magnesia	. 209	. 101	.331	.137	. 138	519
Potash	1.293	.247	.926	.714	not est.	1.467
Soda	. 283	.291	. 229	. 207	not est.	.147
Water, etc	5.126	5 601	6.194	8.322	11.741	6.622
Totals	100.000	100.000	100.000	100.000	100.000	100.000
Sand	not est.	not est.	not est.	44.000	not est.	53.490

All of these Ballard county clays are quite refractory except the ochreous clays or yellow ochre, and could be used as fire-clay in its various applications. They all contain considerable proportions of fine sand. Some of them, having similar composition with the celebrated German glass pot clays,

deserve a trial for this use. The ochreous clay, No. 2572, could be used, after washing to remove sand, as a cheap pigment, either in its natural condition or after calcination.

No. 2574-GREEN SAND. Bluff at Caledonia opposite to Ballard county. Tertiary. Collected by R. H. Loughridge.

A dark-colored, nearly black, friable material. Powder of dark greenish-grey color.

#### COMPOSITION. (Air Dried.)

Silica			٠					•					•							59.940
Alumina																				13.370
Iron peroxide .																• •				10.210
Lime																				.549
Magnesia	60																			2.010
Potash																				3 243
Soda																-		112		078
Phosphoric acid					-						20	-		24		200		94		a trace.
Water								ì	×					8		92				10.600
																				100.000

This contains less potash than is usually found in Glauconite (green sand), but it has, probably, been somewhat altered by weathering.

#### BATH COUNTY.

#### IRON ORES.

No. 2575—Iron Ore from Carnel Rice's, property of Captain W. G. Allen. Clinton group of Upper Silurian formation. Collected by W. M. Linney, June, 1885.

Ore of a yellowish-brown color of the usual structure of the Clinton iron ore.

- No. 2576—Iron Ore from the Purvis lands in Bath county. Clinton group. Collected by W. M. Linney, June, 1885. Resembles the preceding, but is reddish in color.
- No. 2577—IRON ORE. Average sample from the lands of Wm. Warren, near head of Rose Run, Bath county. Clinton Group. Collected by W. M. Linney, June 1st, 1885.

  Ore of the structure of Clinton ore, of a reddish-brown color.

# COMPOSITION OF THESE BATH COUNTY CLINTON IRON ORES. (Air Dried.)

Number	in	Be	po	rt.					2575	2576	2577
Iron peroxide			_						47.630	51.430	58.570
Iron peroxide								. 1	5.468	5.132	3.720
Lime carbonate								. 1	16.560	13.080	15.160
Magnesia carbonate								. 1	9.974	9.444	4.528
Phosphoric acid (P2 O5)								. 1	1.202	1.138	1.010
Bilica								. i	7.160	7.800	6.960
Water, expelled at 212° F	٠							. 1	1.143	.693	1.607
Carbonic acid, water, etc.			•			•			10.863	11.283	8.445
Totals			•	٠		<u>.</u>	·	_	100.000	100.000	100.000
Percentage of iron									33.341	36.001	40.999

Although the iron in the analysis is estimated as peroxide, some of it, in all these ores, is in the form of ferrous carbonate. They also contain a little organic matter not estimated.

These ores have considerable proportions of lime and magnesia, which will aid in fluxing them, and a notable quantity of phosphoric acid, which will not prevent their profitable use in the production of merchantable iron.

### BELL COUNTY.

#### COALS.

No. 2578—Coal. Browning's cannel coal. Three miles above Pineville, on Cumberland river; 276 feet above drainage. Collected by R. C. B. Thruston. A 23-inch layer.

A pure-looking cannel coal. Fracture large, irregular conchoidal. Bird's-eye fracture in parts. Some shining pyrites, but no fibrous coal apparent.

No. 2579—COAL. Myers' bank. Low Branch of Yellow creek. Collected by R. C. B. Thruston. Sample from the 12-inch, 16-inch and 15-inch layers of the bed.

Generally pitch-black, breaking irregularly cuboidal with shining surfaces. Some fibrous coal and pyrites in some of the pieces. No. 2580—Coal. Caney Branch of Yellow creek. Collected by R. C. B. Thruston. Sample from the 20-inch and 8-inch layers of the bed.

Generally pitch-black, breaking irregularly, partly cuboidal, with shining surfaces. No fibrous coal or pyrites apparent in the sample.

- No. 2581—Coal. James Barnett's. Clear Fork of Yellow creek of Cumberland river. Average sample of the coal tested for coking. Collected by R. C. B. Thruston.
- A pure-looking sample, generally of what seems to be bituminous, with some little splint-coal.
- No. 2582—Coal. F. R. Barner's. Yellow creek. Average sample from the barrel-full tested for coking. Collected by R. C. B. Thruston.
- No. 2583—Coal. Myers'. Yellow creek. Average sample from the barrel-full tested for coking. Collected by R. C. B. Thruston.
- No. 2584—Coal. Alf. McTee's. Yellow creek of Cumberland river. Average sample from the coal tested for coking. Collected by R. C. B. Thruston.
- No. 2585—Coal. Dean seam; lower 36 inches of the bed. Greasy creek of Cumberland river. Average sample of the coal collected for coking by R. C. B. Thruston.
- No. 2586—Coal. Dean's seam. Average sample collected for coking two years ago by R. C. B. Thruston.
- No. 2587—Coal. W. D. King's. Yellow creek, Cumberland river. Average sample collected for coking by R. C. B. Thruston.
- No. 2588—Coal. Dorton Coal Bank, half a mile above the mouth of Straight creek, near Pineville. Bed 32 inches thick. Collected by R. C. B. Thruston.

A bright, pitch-black coal. Structure imperfectly laminated; fracture generally cuboidal. Very little fibrous coal and no pyrites apparent.

GEOL. SUR -2.

COMPOSITION OF THESE BELL COUNTY COALS.

(Air Dried.)

Number in the Report.	2578	2579	2680	2681	2682	2583	2584	2685	2586	2587	2688
Hygroscopic moisture Volatile combustible matters Coke	0.86 41.54 57.80	1.00 34 90 64.10	0.88 37.72 61.40	0.80 33.94 65.26	1.20 37.90 60.90	1.16 36.84 62.00	1.40 88 60 60.00	2.00 35.20 62.80	1.50 \$2.90 65.60	98.98 83.08	2.20 88.94 68.86
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100 00
Total volatile matters Fixed carbon in the coke	42.40 50 60 7.00	35.90 55.70 8.40	38.60 57.40 4.00	34.74 58.86 6.40	39.10 67.78 3.12	38.00 57.50 4.50	40.00 57.30 2.70	37.20 58.80 4.00	34.40 60.00 5.60	37.00 68 00 5.00	36.14 62.06 1.80
Percentage of sulphur	5.078 light spongy. grey brown.	1.329 light spongy. grey grey purple.	1.783 light spongy. 't.grey brown.	1.398 light spongy. lt. purple grey.	1.030 light spongy. dark-groy.	1.535 spongy. light grey.	0.629 light spongy. light brown.	0.637 dense spongy. light brown.	0.692 light spongy. whitish.	1.343 spongy. It. grey brown.	0.491 light spongy. lt.brown grey.

Coal No. 2578 is remarkable as being a coking cannel coal. It also contains an exceptional proportion of sulphur, viz.: 5.078 per cent. This, however, may have been an accident of the sampling. This coal and No. 2579 are the only ones of the lot which give an ash percentage above a good average, viz.: 7.00 and 8.40 per cent. severally. All the other coals leave only a moderate or small proportion of ash; No. 2588 giving only 1.80 per cent.

They are all good coals, most of them very good coking coals, as is shown in the results quoted further on.

### BELL COUNTY COKES.

- No. 2589—Coke. Forty-two-hours coke, made at Birmingham, Ala., by G. D. Fitzhugh, Esq., from coal taken from the bank of Mr Frederick Barner, on Yellow creek, Bell county, Ky. (see No. 2582), December, 1885.
- No. 2590—Coke (No. 3a). F. R. Barner's, Yellow creek, Bell county Ky. Forty-two-hours coke, made at Quinnimont, West Virginia, March 22-25, 1886. R. C. B. Thruston.
- No. 2591—Coke (No. 3). F. R. Barner's, Yellow creek, Bell county. Seventy two-hours coke, made at Quinnimont, West Virginia. R. C. B. Thruston.
- No. 2592—Coke. Forty-two-hours coke, made at Birming-ham, Alabama, by G. H. Fitzhugh, from coal taken from bank of James Barnett, on Clear creek, fork of Yellow creek, Bell county, Ky., December, 1885. R. C. B. Thruston. (See No. 2581 for the coal.)
- No. 2593—Coke (No. 6). James Barnett's, Clear Fork of Yellow creek, Bell county, Ky. Forty-eight-hours coke, made at Quinnimont, West Virginia. R. C. B. Thruston.
- No. 2594—Coke (No. 6). James Barnett's, Clear Fork of Yellow creek, Bell county, Ky. Seventy-two-hours coke. made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.

- No. 2595—Coke (No. 6). James Barnett's, of Clear Fork of Yellow creek, Bell county, Ky. Forty-eight-hours coke, made at Quinnimont, West Virginia. R. C. B. Thruston.
- No. 2596—Coke. Forty-two-hours coke, made at Birming-ham, Alabama, by G. D. Fitzhugh, from coal taken from the bank of James M. Myers, on Yellow creek, Bell county, Ky., December, 1885. R. C. B. Thruston.
- No. 2597—Coke (No. 4). James Myers, Yellow creek, Bell county, Ky. Seventy-two-hours coke, made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.
- No. 2598—Coke (No. 4a). James Myers, Yellow creek, Bell county, Ky. Forty-eight-hours coke, made at Quinnimont, West Virginia, Murch 20-23, 1886. R. C. B. Thruston.

COMPOSITION OF THESE BELL COUNTY COKES. (Air Dried.)

Number in Report.	2689	2590	2591	2692	2693	2594	2595	2596	2597	2598
Moist. expel'd at red heat. Carbon in the coke	98.90	0.20 95.80 4.00	0.50 94.50 5.00	1.80 87.20 11.00	0.60 87.80 11.60	0.40 89.20 10.40	1.30 88.90 9.80	2.40 85.30 12.30	0.40 90.80 8 80	0.10 91.50 8 40
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur Color of the ash {	1.480 dark brown.	1.710 dark brown.	0 959 dark brown.	1.343 light brown.	1.590 reddish brown.	1 590 greyish brown.	1.846 brownish grey.	1.548 chocolate brown.	0 547 light grey brown.	0.794 light grey brown.
Hours of coking	42	84	72	42	48	72	48	45	72	48
Owner of the coal	Fre	Frederick Barner.	er.		James Barnett	Barnett.			James Myers	gi

## Bell County Cokes-Continued.

- No. 2599—Coke. Forty-two-hours coke, made at Birming-ham, Alabama, by G. D. Fitzhugh, from coal taken from the bank of Daniel Howard, on Left-hand Fork of Straight creek, Bell county, Ky., December, 1885. R. C. B. Thruston.
- No. 2600—Coke. Forty-two-hours coke, made at Birmingham, Alabama, by G. D. Fitzhugh, from coal obtained from bank of Mr. Alf. McTee, on Yellow creek, Bell county, Ky., December, 1885. (The sample of coal from which this coke was made had some earthy matter mechanically mixed.)
- No. 2601—Coke (No. 5). McTee, Bell county, Ky. (The smaller sample.)
- No. 2602—Coke (No. 5). Alf. McTee, Yellow creek, Bell county, Ky. Seventy-two-hours coke, made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.
- No. 2603—Coke. Forty-two-hours coke, made at Birming-ham, Alabama, by G. D. Fitzhugh, from coal taken from bank of W. D. King, near Yellow creek, Bell county, Ky., December, 1885. R. C. B. Thruston.
- No. 2604—Coke (No. 7). W. D. King, Yellow creek, Bell County, Ky. Forty-eight-hours coke, made at Quinnimont, West Virginia, March 22-24, 1886. R. C. B. Thruston.
- No. 2605—Coke (No. 7). W. D. King, Yellow creek, Bell county, Ky. Seventy-two-hours coke, made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.
- No. 2606—Coke. Forty-two hours, made at Birmingham, Alabama, by G. D. Fitzhugh, from coal taken from bank of Moses Dorton, at mouth of Straight creek, near Pineville, Bell county, Ky., December, 1885. R. C. B. Thruston.
- No. 2607—Coke (No. 16). Seventy-two-hours coke; upper bench of Dean seam, made at Quinnimont, West Virginia, March 20-23, 1886. Bell and Knox counties, Ky. R. C. B. Thruston.
- No. 2608—Coke (No. 13).\* Of lower bench of Dean bank, Still-house Branch of Greasy creek, Bell county, Ky. Seventy-two hours coke, made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.

<sup>•</sup> These numbers in parentheses were attached to the samples by Mr. Thruston.

COMPOSITION OF BELL COUNTY COKES—Continued. (Air Dried.)

Number in Report.	2599	2600	2601	2602	2603	2604	2606	2606	2607	2608
Moist. expel'd at red heat. Carbon in the coke	94.20 5.60	1.60 89 60 8.80	0 70 92.70 6.60	0.40 94.00 5.60	0.26 92.94 6.80	90.66 90.54 8.80	0.20 92.20 7 60	1.00 92.80 6.20	0.90 92.70 6.40	1.80 90.50 7.70
Totals	100.00	100,00	105.00	100.00	100.00	100.00	100.00	100.00	100 00	100.00
Percentage of sulphur Color of the ash {	0 601 brown.	0.491 light brown. 42	0.450 brownish grey.	0.629 brownish grey. 72	1.033 light brown.	1.150 lt. reddish brown.	0.491 brown grey. 72	0.603 light brown 42	1.975 nearly white.	0 684 lt. brown- ish-grey.
Owner of the coal {	Daniel Howard.		Alf. McTee.			W. D. King.		Moses Dorton.	Dean Seam. Up'r b'nch, Lw'r b'nch	Seam. Lw'r b'nch.

These Bell county cokes are generally good, and some of them are very good.

The percentage of carbon in them varies from 95.80 per cent. in No. 2590, to 85.30 per cent. in No. 2596.

The percentage of ash in them varies from 4.00 per cent. in No 2590, to 12.30 per cent. in No. 2596.

The percentage of sulphur in them varies from 0.450 per cent. in No. 2601, to 1.975 per cent. in No. 2607.

#### BOYLE COUNTY.

No. 2609—LIMESTONE from quarry of H. Olmstead, near Danville, Boyle county. Sent by Mr. Procter, June 16, 1885. Will it make good lime?

A dull-looking, dark-grey, fine granular lime-rock, containing fragments of fossils.

#### COMPOSITION .- Air Dried.

Carbonate of lime	82.840 equal to 46.39 per cent. of lime
Carbonate of magnesia	3.511
Alumina and iron peroxide	3,861
Phosphoric acid (P2 O5)	819
Siliceous residue	5.620
Moisture, organic matter, loss, etc	3.349
	100.000

It would calcine into good lime fit for all ordinary uses in building, etc., and the phosphoric acid it contains would add to its value, when applied to poor soils as a top-dressing.

#### BREATHITT COUNTY.

#### COALS.

No. 2610—Coal. Eighty-eight inches. Wolf creek, Breathitt county. Collected by J. M. Hodge.

A weathered sample of splint coal.

No. 2611—Coal. Fifty-nine inches. Berry Turner's. Long's creek, Breathitt county. Sample from the lower thirty inches. Collected by J. M. Hodge.

A pure-looking, pitch-black coal; fracture irregular with shining surfaces. No pyrites apparent, and very little fibrous coal.

- No. 2612—COAL. Gouch & Co.; John Little's Branch, North Fork of Kentucky river, Breathitt county. Middle of the 46-inch seam. Collected by J. M. Hodge.
  - A weathered sample of what appears to be a splint coal.
- No. 2613—Coal. Forty-two inches; splint and bituminous. Green Taulbee's. Collected by J. M. Hodge.

  A weathered sample.
- No. 2614—COAL. Sixty inches. L. H. Nobles, Leatherwood Branch of Lost creek, Breathitt county. Collected by J. M. Hodge.
  - A much weathered sample of what seems to be a splint coal.
- No. 2615—Coal. Forty-three inches. Head of Leatherwood Branch of Lost creek. Collected by J. M. Hodge.

A pure-looking, pitch-black coal. No pyrites apparent, and but little fibrous coal.

No. 2616—Coal. Bituminous coal, 33 inches; cannel coal, 9 inches. Collected by J. M. Hodge.

A weathered sample of the lower 27 inches of the bituminous coal.

- No. 2617—Ooal. Sample of the cannel coal of the next preceding. Collected by J. M. Hodge. A much weathered sample.
- No. 2618—Coal. Gouch & Co.; John Little Branch, North Fork of Kentucky river. Bituminous coal, 35 inches; cannel coal, 11 inches. Breathitt county. Selected sample of the cannel coal. Collected by J. M. Hodge.
- No. 2619—Cannel Coal. Picked sample. Crawford's. Beginning Branch of Middle Fork of Kentucky river. Breathitt county. Collected by J. M. Hodge.

COMPOSITION OF THESE BREATHITT COUNTY COALS.

(Air Dried.)

Number in Report.	2610	2611	2612	2613	2614	2616	2616	2617	2618	2619
Specific gravity	1.861	1.276	1.410	1.310	not est.	1.384	not est.	not est.	1.177	1.274
Hygroscopic moisture Volatile combustible mat. Coke	2.80 33.60 63.60	2.00 35.36 62.64	7.40 30.20 62.40	3.80 34.40 61.80	9.60 29.46 60.94	2.80 31.16 66.04	3.80 32.30 63.90	0.80 41.70 67.50	1.20 63.80 45.00	1.00 41.10 67.90
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters Fixed carbon in the coke. Ash	36.40 54.20 9.40	37.36 57.36 5.28	37.60 52.04 10.36	38.20 51.80 10.00	39.06 44.14 16.80	33.96 53 34 12.70	36.10 48 80 15.10	42.50 33.30 24.20	55.00 39.46 5.54	42.10 46.70 11.20
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur Character of the coke. { Calor of the ash {	0.695 dense brownish grey.	1.019 light spongy. white.	0 621 pulveru- lent. very light sulmon.	0.585 dense. grey brown.	0.478 pulveru- lent. lt. brown- ish-grey.	0.690 dense. lt. brown- ish-grey.	0.840 pulveru- lent. light reddish.	0 952 pulveru- lent. light pink.	0.722 dense. lt. brick colored.	1.120 dense. dark grey.

The cannel coals, Nos. 2617, 2618 and 2619 are distinguished by their large proportions of volatile combustible matters—41.70, 53.80 and 41.10 per cent. severally. The best of these, No. 2618, leaves only 5.54 per cent. of ash, and is quite a pure good cannel coal. The two others, containing 24.20 and 11.20 per cent. of ash material, are proportionately less valuable. The ash percentage in all these coals, except Nos. 2611 and 2618, is above the average of good coals, varying from 9.40 per cent. in No. 2610, and 10. per cent. in No. 2613, up to 24.20 per cent. in No. 2617.

The much weathered samples, Nos. 2612 and 2614, show very large proportions of hygroscopic moisture. Other weathered samples contain it in from 2.00 to 3.80 per cent. In these cases, the coal further in the bed, where it has not been exposed to the atmospheric agencies would probably contain very much less moisture.

#### BUTLER COUNTY.

No. 2620—MINERAL WATER, from Sunny Lane Spring, near Mr. L. B. Orange's, Sunny Lane, Butler county. Collected by C. C. Cohron, M. D., October, 1885

#### COMPOSITION IN 1000 PARTS OF THE WAI'ER.

Carbonate of iron			ij,	٠				×						34	.	0.1315	Held in solution
Carbonate of lime							30	52	٠	60			100			.0147	by carbonic
Carbonate of magne	si	A	8.											:	. ]	.0110	) acid.
Carbonate of soda																.0251	
Sulphate of lime													::		.	.0255	
Sulphate of potash		:						:		20	20	្		10	. 1	.0103	1
Chloride of sodium										2		2		0	. !	.0100	[
Silica																.0264	
Total saline mas	tte	rs	ir	1	00	0	of	th	ie	WB	ite	r	•			0.2545	

It appears to be a good saline chalybeate water.

#### CALLOWAY COUNTY.

Soils, Etc.

No. 2621—VIRGIN SOIL of Clark's river bottom. Taken 10 inches deep. Murray, Calloway county. Collected by R. H. Loughridge.

Dried soil in moderately firm clods of a dirty brown-grey color. It all passed through the coarse sieve.\* Its siliceous residue left a little sand on the fine sieve.†

No. 2622—Virgin Soil of the first bottom of Tennessee river.
W. E. Brown's place, south of Shannon creek. Sample taken to the depth of twelve inches. Growth: white, red, and water oaks, sweet gum, scaly-bark hickory, some poplar, sassafras, ash, papaw, spicewood and redbud. Collected by R. H. Loughridge.

Dried soil of grey-brown color. Clods firm. All passed through the coarse sieve except a very small quantity of shotiron ore. Its silicious residue, from digestion in acids, all passed through the fine sieve, except a very small quantity of fine quartz sand.

No. 2623—Virgin Soil of the Big Barrens, 10 miles northwest of Murray, Calloway county. Sample taken to the depth of 6 inches. Timber: red, post, black-jack and white oaks and hickory, all small. Collected by R. H. Loughridge, 1885.

Dried soil of a grey brown color. Clods friable. All passed through the coarse sieve, except a small quantity of small shotiron ore and vegetable debris. Its silicious residue all passed through the fine sieve, except a small quantity of fine white sand.

No. 2624—Subsoil of the next preceding, taken from 6 to 10 inches deep. Collected by R. H. Loughridge, 1885.

Dried subsoil of a lighter and more reddish grey-brown color than the preceding. Clods rather firm. All passed through the coarse sieve, except a small quantity of shot-iron ore and vegetable debris. Its silicious residue all passed through the fine sieve, except a small quantity of fine white sand.

No. 2625—VIRGIN SOIL OF UPLAND, 1 mile north of New Providence, Calloway county. Taken to the depth of 8 inches. Timber: red and post oaks and hickory. Collected by R. H. Loughridge.

<sup>\*</sup> Coarse sieve has 64 meshes to the centimeter square.

<sup>†</sup> Fine sieve has 1600 meshes to the centimeter square.

Dried soil of a light-grey-brown color. Clods friable. All passed through the coarse sieve, except a small quantity of vegetable debris and small shot-iron ore. Its silicious residue all passed through the fine sieve.

No. 2626—Subsoil of the next preceding, taken 8 to 12 inches deep. Collected by R. H. Loughridge.

Dried soil of a lighter color and more reddish than the preceding; clods more firm. All passed through the coarse sieve except a small quantity of vegetable debris. Its silicious residue all passed through the fine sieve.

No. 2627—Virgin Soil of the second bottom of Tennessee river; W. E. Brown's place, south of Shannon creek, Calloway county. Taken to the depth of 6 inches. Growth: Spanish, red and white oaks, poplar, gum, chestnut and black locust. Collected by R. H. Loughridge, 1885.

Dried soil of a dark grey-brown color. Clods friable. All passed through the coarse sieve except a small quantity of vegetable debris and small shot-iron ore.

No. 2628—Subsoil of the next preceding, taken to depth of 6 to 12 inches. Collected by R. H. Loughridge.

Dried soil of a light reddish-grey-brown color. Clods friable. All passed through the coarse sieve except a small quantity of vegetable debris. Its silicious residue all passed through the fine sieve, except a small quantity of fine white sand and a few small specks of mica.

No. 2629—VIRGIN UPLAND SOIL, 5 miles east of Murray, Calloway county. Taken to the depth of 10 inches. The soil is dark for one inch, and then a light-brown loam to the subsoil. Growth: red oak chiefly, some post oak and hickory. Collected by R. H. Loughridge.

Dried soil of a light yellowish-grey-brown color. Clods somewhat firm. All passed through the coarse sieve except a small quantity of vegetable debris and fine shot-iron ore. Its silicious residue all passed through the fine sieve except a small quantity of fine sand.

No 2630—Virgin Soil. Upland sandy loam of the "Coalings" of the south-east corner of Calloway county Taken to the depth of 6 inches. Original growth: red, black Spanish and post oaks and hickory. Collected by R. H. Loughridge.

Dried soil of a dirty brownish-buff color. Clods quite friable. All passed through the coarse sieve except a little small shot-iron ore. Its silicious residue all passed through the fine sieve except 2.8 per cent. of fine white quartz sand.

No. 2631—Subsoil of the next preceding. Sample from depth between 6 and 12 inches. Collected by R. H. Loughridge.

Dried subsoil of a brownish-yellow color. Clods moderately firm. All passed through the coarse sieve except a small quantity of small shot-iron ore. Its silicious residue all passed through the fine sieve except 2.5 per cent. of fine quartz sand.

No. 2632—VIRGIN SOIL of Post Oak Flatwoods, 2 miles east of the Murray and Paris road, near the Tennessee State line. Sample taken to the depth of 6 inches. Timber: post and red oaks. Collected by R. H. Loughridge.

Dried soil of a dirty grey-buff color. Clods friable. All passed through the coarse sieve except a small quantity of fine shot-iron ore. Its silicious residue left 5.8 per cent. of fine white sand on the fine sieve.

No. 2633—Subsoil of the next preceding. Sample from depth between 6 and 12 inches. Collected by R. H. Loughridge.

Dried subsoil of a light-yellowish-grey color. Clods somewhat firmer than of the preceding. Coarse sieve removed from it only a small quantity of small shot-iron ore. Fine sieve separated 5.05 per cent. of fine white sand from its silicious residue.

No. 2634—VIRGIN BOTTOM SOIL of Shannon creek. Taken to the depth of 6 inches. A dark-blackish loam. Growth: white and red oaks, poplar and some walnut. Collected by R. H. Loughridge.

The coarse sieve separated a considerable quantity of shotiron ore. Fine sieve removed 2.05 per cent. of fine sand from silicious residue.

No. 2635—Subsoil of next preceding. Taken at depth of 6 to 12 inches. A stiff yellowish clay. Collected by R. H. Loughridge.

Dried subsoil of a light-grey yellow color. Clods quite firm. Coarse sieve removed a small quantity of shot-iron ore. Fine sieve separated 2.2 per cent. of fine white sand.

COMPOSITION OF THESE CALLOWAY COUNTY SOILS AND SUBSOILS.

(Calculated Dried at 212° F.)

Number in Report.	2621	2622	2623	3624	2625	3626	2627	3628	6292	2630	2631	2632	2633	2634	2635
Organic and vo'atile matters	3.475	6.080	3.540	2.632	2.862	2.569	5.482	2 627	3.80	2.497	2.999	2.650	2.188	5.848	3.500
Alumina and manganese oxide	3.612	171 8	3.18	5.239	3.321	4.920	88	5.333	3.711	2.020	4.725	2.564	2 . 2 . 8 . 8 . 8 . 8 . 8	2.455	5.575
Lime carbonate	20.0	8	960.	80	trace.	940	141	28	941	131	9.5	. 65°	8 :	378	mo.
Phosphoric acid (P2 O5)	8.8	218	6.8	8	.00	9	72	124	8	8	8	0,00	8:	546	8.3
Potash extracted by acids Soda extracted by acids	310 not est.	not est.	<b>*</b> 611.	not est.	300	2.8	not est.	not est.	.029	not est.	not est.	not est.	not est.	not est.	not est.
Water expelled at 380° F Sand and insoluble silicates	88.432	2.135	88.883	.860 86.1r6	89.353	86.342	1.401	86.300	88.122	92.631	86.522	91.345	90.90	84.356	84.426
To als	99.308	100.316	100.052	99.747	99.993	99.302	99.003	100.425	100.786		99.769	100.00	99.815	100.013	906 66
Hygroscopic moisture	1.450	2.650 1.487 253	1.150	1.378	0.950	1.125	1.500	3.5. 3.5.	1.307	0.875 1.605 424	1.650	1.150	1.30	388	1.380 5.88

The alluvial river bottom soils, Nos. 2621, 2622, 2627, 2628, 2634 and 2635, with the single exception of No. 2621, from Clark's river bottom, have the composition of good fertile soils, containing all the essential elements and organic matters in more than average proportions, and with good culture and sufficient drainage ought to be quite productive. No. 2621 is quite deficient in lime and phosphoric acid, and would, doubtless, respond favorably to applications of super-phosphates or other fertilizers containing these essential ingredients.

Lime is apparently quite deficient, not only in this but in soils Nos. 2625, 2626 and 2631, and is below a good average in Nos. 2623, 2624 and 2633. Top-dressings of slacked lime or ground limestone would, no doubt, be profitable to all these soils. Phosphoric acid is very deficient in soils Nos. 2621, 2625 and 2626, and in proportions below a good average in Nos. 2623, 2624, 2629, 2630, 2631 and 2632. Potash is very deficient in Nos. 2630 and 2632, and below a good average proportion in No. 2623. No. 2630 appears, by its chemical composition, to be the poorest of all these soils; but this, as well as all the others, if well drained and properly cultivated, with the use of suitable manures or fertilizers, may be made to produce good crops.

#### SANDS OF CALLOWAY COUNTY.

No. 2636—Fine Micaceous Sand, slightly calcareous. Cretaceous formation, 2 miles north-east of New Concord. Collected by R. H. Loughridge.

A very fine-grained white sand, containing minute specks of mica. All passed through the coarse sieve. Its silicious residue, from digestion in acids, all passed through the fine sieve except 4.45 per cent. of fine white quartz grains with small specks of mica.

No. 2637—Sand apparently similar to preceding, but slightly colored brownish with iron peroxide. Collected by R. H. Loughridge.

All passed through the coarse sieve. The fine sieve separated from its silicious residue only 6.75 per cent. of small quartz grains with mica scales.

# COMPOSITION OF THESE CALLOWAY COUNTY SANDS. (Calculated Dried at 212° F.)

Numb	er	in	Re	po	rt	•										2636	2637
Organic and volatile matters													•0		.	1.815	0.69
Organic and volatile matters Alumina, iron peroxide and	P	, (	0,													. 605	,50
Lime carbonate		•													.	.090	.09
Magnesia															.	.232	trace.
Potash removed by acids .		en e														.044	not est.
Soda removed by acids															. :	not est.	not est.
Water expelled at 600° F.																	
Lime carbonate		e i						•	٠					×		97.395	98 14
Total			•			•	٠	•			•			•	٠.	100.181	99 43
Hygroscopic moisture											٠			•	. !	0.075	0 02
l'otal potash		92 9			•		•	•	•	•	3.	•	•	•	353	0.514	0.28

These sands might be useful in glass-making or to mix with tough clays for pottery, etc.

## CALLOWAY COUNTY.

CLAYS, ETC.

No. 2638—CLAY. Siliceous earth (or decomposed chert) from Brandon's Mill. Lower Carboniferous siliceous group. Calloway county. Collected by R. H. Loughridge.

A very fine-grained, friable, white mass, containing small pieces of undecomposed chert. Powder harsh to the feel.

No. 2639—CLAY. Tertiary, half a mile south of Wyatt's school-house, north-west part of Calloway county. Collected by R. H. Loughridge, June 17, 1885.

A white clay. Powder feels soft. Contains about 21 per cent. of fine white sand.

No. 2640—White Pipe-clay. Below Quaternary gravel. R. Morris', east of New Providence. Used for pottery. Collected by R. H. Loughridge.

Color, light-buff-grey, with some darker streaks. Powder soft, contains a small proportion of very fine sand.

No. 2641—BLACK PYRITOUS PLASTIC CLAY. Tertiary; below the white pipe-clay. R. Morris', east of New Providence.

Indurated, imperfectly laminated, fine-grained. Powder soft. Light slate color. Contains a few small grains of pyrites.

No. 2642—Black Joint-Clay or so called soapstone. Lignitic Tertiary. Murray, Calloway county. Collected by R. H. Loughridge.

Fine grained, indurated clay, showing minute specks of mica. Tough. Fracture irregular, with imperfect lamination. Said not to glaze in pottery. Washed out 39.78 per cent. of light-brown, fine sand, containing a few small mica scales.

No. 2643—CLAY from Russell's Pottery, 6 miles east of Murray. Collected by R. H. Loughridge.

Nearly white, with a few small brownish ochreous veins.

No. 2644—DARK CLAY. (Bluish when wet.) Mahan's, 6 miles east of Murray. Collected by R. H. Loughridge.

Color, purplish-slate-grey. Contains minute mica scales, and about 39 per cent. of fine white sand, with minute mica scales.

No. 2645—Black Plastic Clay underlying white clay at Rufus Morris', east of New Providence. Tertiary. Collected by R. H. Loughridge.

Resembles the next preceding. Imperfectly laminated.

No. 2646—Yellow Clay or ochre. Wadesboro. Collected by R. H. Loughridge.

Of a handsome yellow ochre color. Calcines of a handsome red color. Fuses before the blow-pipe. Resembles No. 2572, of Ballard county (which see). Laminated, Laminæ varying in depth of tint.

COMPOSITION OF THESE CALLOWAY COUNTY CLAYS. (Air Dried.)

		4	Number in Report.	a p	<u></u>	- L	de)	HOL	ئب				2638	<b>ac</b>	2639	2640	2641	2642	2643	2644	2645
Silica							:		! :				87.8	300	46.020	61.680	56.680	66.380	57.840	64.140	57 400
Alumina .		: 10 : 13		•	•	٠	•			•	٠	<u>~</u>	10 4	180	38 980	28.500	29, 700	16.480	30.340	32.140	29 440
I ron peroxide	100		•	•	*3	•				•	٠	•			-	_	084	3.500	2:	1 040	134
Magnesin .					• •			. :		•	• •	 		287	3 %	136	28.	261	000	033	245
Potash			•		- 11		16						· •.	888	808	-	1 004	928	.618	.965	. 522
Soda		•	٠	•	: .*					•		:	•	500	.172		+72.	.228	616.	.468	. 437
Water, etc.		· •	•	3 50 3 50	1					•			•;	262	13.610	ū	10.581	11 774	6 45	11.204	10.48
Totals	*				•	•				•	•		100.000	8	100.000	100.000	100 000	100 000	100 000	100.000	100.000
Sand	0	•	8.5		•				:	•	2		not est.	est.	21.000	not est.	not est.	39.780	not est.	39.000	not est

The decomposed chert No. 2638, is mainly very fine and pure siliceous sand. It would serve well to mix with tough clay to reduce its shrinkage, etc. It might answer to make what are called Bath bricks, used for scouring purposes or polishing, and possibly could be used as an ingredient of white glass.

These are mostly good refractory clays. Those which contain the least of fluxing materials, such as iron oxide, lime, potash, soda or magnesia, are the most refractory, and would make good fire-brick or linings for furnaces, etc. Many of them would make good white pottery ware or terra-cotta.

No good reason appears from the analysis why No. 2642 should not glaze,\* as, on testing it for sulphate of lime, very little was found in its composition.

The least refractory of these clays is, undoubtedly, No. 2642, which contains 3.500 per cent. of iron peroxide; 0.928 per cent. of potash, etc. The large proportions of these fluxing materials is, however, partly neutralized by its large proportion of silica.

The ochreous clay—yellow ochre—might be used as a cheap pigment, either raw or calcined. (See remarks on Ballard county clays, etc.)

## CLAY COUNTY.

#### COALS.

No. 2647—Coal. Fifty-one inches. Isaac Jackson's, left Fork of Goose creek, Clay county. Collected by G. M. Hodge, December, 1885.

A pure-looking coal. No apparent pyrites and but little fibrous coal. Ferruginous stains on some of the pieces.

No. 2648—COAL. Thirty-one inches. T. T. Garrard's mine, east side of Goose creek, at its forks. Sample from 200 feet under. Collected by G. M. Hodge, December, 1885.

A pure-looking, pitch-black coal, with very little fibrous coal and only a few specks of pyrites.

<sup>•</sup> With common salt alone.

No. 2649—Coal. Thirty-nine inches. Mrs. S. A. White's, Salt Works mine. Sample from twenty feet under. Collected by G. M. Hodge.

Resembles the preceding. No pyrites apparent.

No. 2650—Coal. Forty-five inches. G. L. Hornsby's, Laurel creek, Clay county. Collected by G. M. Hodge.

Apparently a good splint coal. No apparent pyrites, but some ferruginous stains. Seems to be a somewhat weathered sample.

- No. 2651—Coal. Forty-seven inches. G. M. Jones', Beech creek, Clay county. Sample of the upper 32 inches. Collected by G. M. Hodge.

  Somewhat weathered.
- No. 2652—Coal. Forty-seven inches. G. M. Jones', Beech creek. Sample of the lower 15 inches. Collected by G. M. Hodge.

A much weathered sample.

No. 2653—Coal. Bituminous, 30 inches; cannel coal, 5 inches. J. T. Smith's, Tom's Branch, Left Fork of Goose creek. Sample of the bituminous coal from 15 inches under. Collected by G. M. Hodge. (See No. 2647.)

A much weathered sample, with reddish, ferruginous incrustation.

No. 2654—COAL. Thirty-eight inches. Alvis Hubbard's, Katy's creek. Red Bird creek, Clay county. Collected by G. M. Hodge.

A somewhat weathered sample.

No. 2655—Cannel Coal. J. T. Smith's, Tom's Branch Goose creek. (See No. 2653.) Bituminous coal, 30 inches; cannel coal, 5 inches.

COMPOSITION OF THESE CLAY COUNTY COALS. (Air Dried.)

Number in Report.	2647	2648	2619	2650	2651	2652	2653	5654	2655
Specific gravity	1.288	1.287	1.278	1.292	1.813	not est.	not est.	1.290	1.160
Hygroscopie moisture Volatile combustible matters	35.6 <b>6</b> 63.30	38.10 60.70	1.48 35 92 62.60	1.46 34.84 63.70	0.92 87 54 61 54	0.42 82.38 67.20	2.80 29.40 67.80	1.60 34.28 64.12	0.30 44 16 55 54
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100 00	100.00	100.00
Total volutile matters	36.70 56.90 6.40	39.30 5.80 5.80	37.40 64.70 7.90	36.30 57.70 6.00	38.46 53.44 8.10	06.98 27.98 27.98	32 20 67.00 10 80	35 88 54 82 9 30	44. 1.84. 1.80.
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100 00	100.00	100.00
Percentage of sulphur	0 885 light spongy. It. brown- ish-grey.	1.793 spongy. lilac-grey.	0.885 spongy. light lilac-grey.	0.531 spougy. nearly white.	1 (801 spongy. It. brown- ish-grey.	5 042 dense, dark brown.	1.178 deuse friable. light brown.	1.766 dense spongy. purplish brown.	1 244 dense. dark-grey.

The large proportion of ash of No. 2652, viz.: 32 per cent., greatly reduces its value, especially for transportation, and its more than 6 per cent. of sulphur would prevent its profitable use in the working of iron. Nos. 2655, 2654 and 2653, containing severally 11.80, 9.30 and 16.80 per cent. of ash, are proportionately reduced in value, especially for cokemaking. The lowest ash proportions are found in Nos. 2648, 2650 and 2647, being 5.80, 6.00 and 6.40 severally. The large amount of ash in several of these Clay county coals would not prevent their profitable use as fuel in the vicinity of the several beds.

#### CLINTON COUNTY.

#### MINERAL WATER.

No. 2656-MINERAL WATER from Sewell's Mountain, near Albany, Clinton county.

Near the summit of Poplar or Sewell's Mountain. Issues from beneath a heavy ledge of carboniferous sandstone or conglomerate. Flow small. Temperature 52° F. Collected by R. H. Loughridge.

#### COMPOSITION IN 1000 PARTS OF THE WATER.

Iron carbonate															•		0.320 ) Held in sol
Lime carbonate																	.011 } tion by ca
Magnesia carbon	at	e													•3	. '	.021 ) bonic acid.
Silica																	
Potash and soda	sa	lts	i	n	ni	nu	te	qı	18	nti	ty				•		not est.

A weak chalybeate water. Good of its kind.

### FLEMING COUNTY.

#### MINERAL WATERS.

No. 2657—CHALYBEATE WATER from Fox Spring. Geological position: near the top of the corniferous group. Collected by W. M. Linney.

No. 2658—WHITE SULPHUR WATER. Fox Springs. Geological position: Black slate. Collected by W. M. Linney.

COMPOSITION	IN	1000	PARTS	OF	THE	WATER
-------------	----	------	-------	----	-----	-------

1	Nu	m	bei	r i	n .	Re	po	rt												2657	2658
Carbonate of iron																		-		0.0130	0.0016
Carbonate of lime	- 13	•	•	88			-	-			•	30	•		10	1072	50		١.		.098:
Carbonate of magnesia	- 50	-	·	:5	1		53	- 10		· ·		1	5	15	85	2000	5,0	50			.0430
Sulphate of iron		-	•			•		•		•		•	•	•			•		٠.	.0175	
Sulphate of potash .												*0							1	.0244	.0149
Sulphate of soda																				. 0424	.0460
Sulphate of lime																	200		ĺ	. 0483	
Sulphate of magnesia	2	23													94	7720	20			.0485	
Chloride of sodium .			2	82			-	20	0	34	201	25		00	860		20			trace.	.042
Carbonate of soda			į									·		i							.0538
Bilica								0			2								100	.0328	.0238
Organic matters and lo	ess	3		•				•	្វ				•	•						.0121	. 080
Total									2.4					-2			*		-	0.2390	0 4038

No. 2657 is a good saline chalybeate water. No. 2658 is an alkaline saline chalybeate, containing much less of salts of iron than the other.

# FULTON COUNTY.

No. 2659—Grey Silt or Loess. Near the top of the bluff at Hickman, Fulton county. Collected by R. H. Loughridge.

Dried silt of greyish-cinnamon color. Clods friable. All passed through the coarse sieve, except a few small rock fragments. Its siliceous residue, from digestion in acids, all passed through the fine sieve.

#### COMPOSITION .- (Air Dried)

Alumina and manganese oxid	e																.`)	4.062
Iron peroxide																		2.125
Lime carbonate														•	200		1	14.900
magnesia caroonate			1.0														a. [	7 670
rnosphoric acid (P. U.)				 			-			-	0.0			200	-			.173
totasii extracted by acids		100			-		1.20		-								.50	.521
Soun extracted by acids														2.1		8.00	84	1.082
and and insoluble silicates .					-						100			0		12		67 293
Moisture, organic matters, etc					*	•				·	•			è	÷	9		2.17:
Total								٠						•			. '	100 000
																	*	
Potash in silicious residue	٠				•				•			•	•	•			.	1.410
Soda in siliceous residue				•				•						•			.	1.268

(See No. 2142 in previous report for analysis of the lower portion of this bed.)

#### GRAVES COUNTY.

SOILS.

No. 2660—VIRGIN SOIL of black-jack oak barren, three miles north-west of Mayfield. Taken to the depth of 10 inches. Growth: chiefly black-jack oak and some young red oaks. Collected by R. H. Loughridge.

Dried soil of a grey-brown color. Clods friable. All passed through the coarse sieve.\* Its siliceous residue, from digestion in acids, all passed through the fine sieve.†

No. 2661—Subsoil of the next preceding. Taken from 6 to 12 inches deep. Collected by R. H. Loughridge.

Dried subsoil of a grey-buff color. Clods comparatively firm. All passed through the coarse sieve. Its siliceous residue, from digestion in acids, all passed through the fine sieve.

# COMPOSITION OF THESE GRAVES COUNTY SOILS.

(Calculated Dried at 212° F.)

Numb	er	u	1	Re	)x	rt	•									2660	2661
Organic and volatile matters									-	i i i						2.639	2.187
Alumina and manganese ox	ide					83							:			2.636	3.511
Iron peroxide											्					1.756	2.48
Iron peroxide											٠					.096	. 14
Magnesia																.159	. 238
Phosphoric acid (P. O.)																.084	. 049
Potash extracted by acids .			•		•						*					.149	.110
Soda extracted by acids.					*3						**		18			not est.	. 039
Water expelled at 380-400°	F.					30	300				*		3		*	.696	1.150
Sand and insoluble silicates					•	٠			•	٠	*	٠	37	٠		91 295	89 813
Total			•							ě	v					99.510	99.73
Hygroscopic moisture																0.915	0.57
Hygroscopic moisture Potash in siliceous residue																1.732	1.82
Soda in siliceous residue .			•	•												1 461	. 870
Character of the soil					٠			•	٠				•	•	{	virgin surface.	subsoil.

These may be characterized as light soils, containing moderate, or small proportions of organic matters, lime and phosphoric acid, with an average amount of potash Manures would be required to keep up fertility.

<sup>\*</sup> With 64 meshes to the centimeter square.

<sup>†</sup> With 1600 meshes to the centimeter square.

# GRAVES COUNTY.

#### CLAYS.

No. 2662—Arenaceous Clay (from decomposing chert?).
Railroad cut, 3 miles north of Boaz station. Quaternary.
Collected by R. H. Loughridge.

Lump of the hardness of chalk, of a fine granular texture; light-brownish-grey color. A portion of a small, whitish, rounded chert pebble in the sample. Somewhat plastic, very difficult to fuse before the blow-pipe.

No. 2663—CLAY, 3 miles west of Linnville. Lagrange Tertiary group. Collected by R. H. Loughridge.

Dried clay of a light-brownish-grey color. Fine textured. Infusible before the blow-pipe.

No. 2664—Whitish Pipe-clay. South of Guill Hill. Railroad cut, 3 miles south of Wingo. Tertiary. Collected by R. H. Loughridge.

Dried clay of a light-brownish-grey color; nearly white. Of very difficult fusion before the blow-pipe.

No. 2665-CLAY. Boaz Station. Collected by R. H. Lough-ridge.

Dried clay, nearly white (light-brownish-grey). Fine textured. Very difficult to fuse before the blow-pipe.

No. 2666-CLAY from Howard's pottery, at Bell City. Collected by R. H. Loughridge.

Dried clay, nearly white (of a light-brownish-grey color), with some ferruginous stains. Fine textured. Infusible before the blow-pipe.

COMPOSITION	OF	THESE	GRAVES	COUNTY	CLAYS.
		(Air	Dried.)		

Number in Report.	2662	2668	2664	2665	2666
Silica	76.780	62.680	75.120	61.920	56.980
Alumina	14.740	25.880	15.960	30.060	32.160
Iron peroxide	1.640	2.900	1.420	.300	2 160
Lime	trace.	trace.	trace.	truce.	trace.
Magnesia	.389	.319	.317	.064	. 209
Potash	1.440	1.147	1.351	1.602	.838
Soda	.117	928	.245	.239	.111
Water, etc	4.894	6.146	5.587	5.815	7.542
Total	100.000	100.000	100.000	100.000	100.000
Fine sand	2 400	not est.	not est.	not est.	16.44

The arenaceous clay, No. 2662, contains enough alumina to make it plastic enough for the manufacture of good firebrick, etc. No. 2664 has an analogous composition, and had, probably, the same origin. The other clays are quite plastic, very refractory, and good for the manufacture of light-colored or white pottery-ware, fire-brick, etc., and terra-cotta.

# HARLAN COUNTY.

COALS.

No. 2667—Coal. Sixty-three inches. Dale Bledsoe's, Reuben Branch of Beech Fork of Middle Fork of Kentucky river. Collected by G. M. Hodge.

A somewhat weathered sample. Has no apparent pyrites.

No. 2668—Coal (sample 4). Three and a quarter miles above the mouth of Clover Lick creek of Poor Fork of Cumberland river. Outcrop sample from the exposure in the bed of the creek. Forty-one inches thick. A seam 75 to 100 feet below sample 3. Collected by R. C. B. Thruston.

A slightly weathered sample.

No. 2669—Coal. Six feet five inches thick. Right bank of Looney creek, about 2 miles above the mouth. Collected by T. H. Morgan, August 24, 1885.

A much weathered sample, containing a few pieces of bright hard coal.

No 2670-COAL on the land of E. Dickson, on Tantrough Branch of Poor Fork of Cumberland river, 20 miles from Mount Pleasant. Coal No. 1.? Collected by R. C. B. Thruston.

A much weathered sample.

No. 2671—COAL a quarter of a mile up Tyrey's Branch of Clover Lick creek (3 miles above its mouth) of Poor Fork of Cumberland river. Collected by R. C. B. Thruston. The coal shows, across Tyrey's Branch, 41 inches thick.

A weathered sample of the outcrop.

No. 2672—Coal in Fickle's Cove of Looney creek, on J. Jenkins' land. Coal 6 feet 4 inches thick.

A weathered sample.

No. 2673—Coal. Right bank of Looney's creek, about 2 miles above the mouth. Coal 4 feet thick. Collected by T. H. Morgan, August 24th, 1885.

A much weathered sample.

- No. 2674—COAL (sample No. 5). Garner's Hollow, 15 miles above the mouth of Poor Fork of Cumberland river. Cannel coal 42 inches, on land of John L. Cornett, beneath cannel shale. Collected by R. C. B. Thruston.
- No. 2675—COAL (sample No. 6). Thirty-six inches thick. Garner's Spring, 15 miles above the mouth of Poor Fork. One hundred and forty feet above black limestone.

A much weathered and dirty sample. (Dust was sifted out before analyzing.)

No. 2676—Coal (sample No. 7). Head of Island Branch of Poor Fork of Cumberland river, 16 miles from Mount Pleasant; 180 feet above black limestone. Coal 51 inches thick, on land of John L. Cornett. Collected by R. C. B. Thruston.

A much weathered sample. The dust was sifted out before analyzing.

No. 2677—COAL (sample No. 9). Head of Island Branch of Poor Fork of Cumberland river. Soft coal below sample 8, 23 inches thick. Collected by R. C. B. Thruston.

Somewhat weathered. The sample contained some slaty pieces, which were retained when it was analyzed.

COMPOSITION OF THE ABOVE DESCRIBED HARLAN COUNTY COALS.

(Air Dried.)

Number in Report.	2667	2668	5669	2670	2671	2672	2673	2674	2675	2676	2677
Hygroscopic moisture Volatile combustible matters	1.60 33.30 65.10	1.12 33 68 65.20	3.40 31.08 65 52	1 60 35.00 63.40	1 40 32.20 66.40	2.30 38.30 64.40	2.50 34.30 63.20	1.00 29.50 69.50	4.62 32.98 62.40	6.08 32.72 61.20	1.90 36.00 62.10
Totals	100.00	100.00	100 00	100.00	100 00	100 00	100.00	100.00	100.00	100.00	100.00
Total volatile matters Carbon in the coke	34.90 49.70 15.40	34.80 63.10 2.10	34.48 62.52 3.00	36.60 67.40 6.00	33.60 57.20 9.20	35.60 57.20 7.20	36.80 58.70 4.50	30.50 31.60 37.90	37.60 58.30 4.10	38.83 3.20 3.00	87.90 56.10 6.00
Totals	100.00	100.00	100 00	100.00	100.00	100.00	100.00	106.00	100.00	100.00	100.00
Percentage of sulphur	spongy. spongy liluc light grey. brown?	0.420 spongy. light brown'h.	0.365 friable. lt. purple grey.	0.558 spongy. reddish brown.	O.742 dense. lilac grey.	0.502 dense. brown.	0.502 dense friable. brown'h.	0.613 pulver- ulent. brown.	0.475 pulver- ulent. very lt. brown.	0 311 pulver- ulent. light brown.	o.461 spongy. light brown.

No. 2674 contains so much ash material, 37.9 per cent., as to cause it to be unprofitable for transportation or the manufacture of coke. The profitable use of its 61.1 per cent. of combustible materials could only be made near the bed. No. 2667 also contains more than a good average of ash. Several of these coals, especially those with little ash and sulphur, would make good coke. No. 2669, which gives only 3 per cent. of ash, presents the apparent anomaly of yielding a friable coke. This can only be explained by the fact that it was a much weathered sample. Very probably the coal, deeper in the bed where it has not been altered by the atmospheric agencies, will be found to give a spongy coke, like its neighboring coal in this schedule, No. 2668, which it resembles in composition.

# HARLAN COUNTY COALS, ETC.—Continued.

No. 2678—Coal (sample 10), from near the mouth of Island Branch of Poor Fork of Cumberland river, 16 miles from the Court-house of Harlan county. Coal 36 inches, with two thin partings of fibrous coal. Land of John L. Cornett. Collected by R. C. B. Thruston.

A slightly weathered sample, containing some slaty pieces. Part of it is bright fat coal.

No. 2679—Coal (sample 13). George Turner's bank, 1 mile above Mount Pleasant, in Big Black Mountain, on Clover Fork of Cumberland river. Coal 48 inches. No parting. Collected by R. C. B. Thruston.

A bright pitch-black coal, fracture generally cuboidal and irregular. Very little fibrous coal apparent, but some granular pyrites.

No. 2680—Coal (sample 14). Mr. Silas Woodson Kelly's, Yocum's creek, Clover Fork of Cumberland river. Sample from 2 seams, 22 and 17 inches thick severally, separated by 8 inches of shale. Lower 17-inch layer is very hard and rich. Collected by R. C. B. Thruston.

Apparently a pretty good, firm splint coal, not much fibrous coal between the laminæ, and no apparent pyrites.

- No. 2681—BITUMINOUS SHALE (sample 15). Sharpe's creek of Yocum's creek of Clover Fork of Cumberland river. Land of John Farley. Shale 15 inches thick, above coal 25 inches thick. Collected by R. C. B. Thruston.
- No. 2682—Coal (sample No. 16). Twenty-five inches of the bed above described. Collected by R. C. B. Thruston.

A bright pitch-black coal, breaking generally irregular cuboidal. Very little fibrous coal and no pyrites apparent in the sample.

No. 2683—COAL (sample 17). Right Hand Fork of Yocum's creek of Clover Fork of Cumberland river. Land of Wright Winn. Collected by R. C. B Thruston. Coal 31½ inches in all, with a clay parting.

Portions breaking somewhat irregularly; pitch-black, with shining surfaces. Other portions dull black, breaking in irregular laminæ, with but little fibrous coal between them, and no apparent pyrites.

No. 2684—Coal (sample 18). Turkey-pen Branch of Poor Fork, 14 miles above Harlan Court-house. Land of Jonathan Cornett. Coal 39 inches thick. Collected by R. C. B. Thruston. Sample from the lower 39-inch seam.

A somewhat weathered sample of what appears to be a splint coal, much of the sample in a powdered condition. Some fibrous coal and softer, brighter lumps in the sample.

No. 2685—Coal (sample 19). On Isaac Creech's land on Mud-Lick Fork of Seagreaves creek of Clover Fork of Cumberland river. Collected by R. C. B. Thruston. Sample from the 23, 11 and 5-inch seams, which are parted by thin shale layers.

A pitch-black coal, breaking generally into irregular laminar with shining surfaces. Some pieces more dull. Some bright scales of pyrites and fibrous coal apparent.

No. 2686—COAL (sample 20). Child's creek of Clover Fork of Cumberland river. Collected by R. C. B. Thruston. Sample of 40 inches of the 44-inch seam near the bottom of the bed.

Sample from the outcrop, badly weathered. The bed has a soil roof and is opened at the side of a bridle path.

No. 2687—Coal (sample 21). Child's creek of Clover Fork of Cumberland river. Three hundred feet below sample 20. Collected by R. C. B. Thruston. Sample from the 21, 5, and 11-inch seams. One shale parting. Sandstone roof.

A hard splint coal, dull black, with but little fibrous coal and no apparent pyrites.

No. 2688—COAL (sample 22). On Buck Branch of Child's creek of Clover Fork. Sample includes all the 3 seams of the bed, 10, 13 and 11 inches severally, which are parted by thin layers of shale. Collected by R. C. B. Thruston.

Generally splint coal. But little fibrous coal and no pyrites apparent. Some ferruginous incrustation. Portions breaking irregularly, with pitch-black, irregular surfaces.

COMPOSITION OF THE ABOVE DESCRIBED HARLAN COUNTY COALS.

(Air Dried.)

Number in Report.	2678	2679	7680	2681	2682	2683	2684	2685	2686	2687	2688
Hygroscopic moisture.	2.02 32.98 65.00	1.76 34.64 63.60	1.60 37.70 60.70	1.04 21.88 77.08	1.50 85.30 63.20	1.52 33.00 65.48	31.16 67.04	1.58 31.82 66.60	6.18 34.24 59.58	2.20 33.52 64.28	1.80 35.00 63.20
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	35.00 52.70 12.30	86.40 60.50 3.10	39.30 54.84 5.86	22.92 29.60 47.48	36.80 60.24 2.96	34.62 51.96 13.52	32.96 62.28 4.76	33.40 63.20 3.40	40.42 62.88 6.70	85.72 55.72 8.56	36.80 57.30 5.90
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Character of the coke	0.546 spongy. light salmon.	0.904 light spongy. very lt. grey.	0.968 light spongy. light brown.	5.486 pulverulent. purplish brown.	1.041 dense spongy. brown.	0.684 dense spongy. light grey.	0.794 dense spongy. light buff.	1.288 spongy. brown.	0.519 pulver- ulent. light buff.	0.656 dense friable lt.yel'b- grey.	0.986 dense spongy. It grey'h brown.

The bituminous shale, No. 2681, with its 47.48 per cent. of ash, yet contains 51.48 per cent. of combustible matters, volatile and fixed, including 5.436 per cent. of sulphur. It would not bear the cost of distant transportation, nor could it well be used for fuel for most manufacturing purposes, yet it may find profitable application in the close vicinity of the bed.

Several of these coals above described are very good coking coals, especially those which leave but little ash and contain a small proportion of sulphur. Coal No. 2686, which contained so much moisture—6.18 per cent., and left only a pulverulent coke—being a much weathered sample, may be found a good coking coal deeper in the bed, where it has been protected from the atmospheric agencies.

### HARLAN COUNTY COALS-Continued.

No. 2689—Coal (sample 23). Buck Branch of Child's creek of Clover Fork. Three hundred and twenty-five feet above sample 22. Collected by R. C. B. Thruston. Sample from all the coals of the bed.

A mixed sample. Some splint coal, with fibrous coal between the laminæ, but no apparent pyrites; other portions breaking irregularly, with pitch-black, shining surfaces.

No. 2690—Coal (sample 24). Baily Hollow of Clover Fork, 20½ miles above Mount Pleasant; 135 feet above drainage. Collected by R. C. B. Thruston.

Sample from 36 and 8-inch layers. The 36-inch coal is a fat coking coal, with four bands of fibrous coal. The 8-inch coal below is harder, with some little splint coal.

The bed has five several seams of coal, in all 65 inches, separated by five layers of shale, in all 41 inches. With sandstone roof.

No. 2691—Coal (sample 25). Head of Child's creek of Clover Fork, within 20 yards of the State line, 360 feet above sample 20. Sample from the upper 24-inch coal.

The bed contains four seams of coal, separated by three layers of shale, with a conglomerate sandstone roof.

The sample seems to have been weathered.

No. 2692—Coal (sample 26), of the 24-inch splint coal in the bed described under the next preceding number. Collected by R. C. B. Thruston.

Contains some fibrous coal, but no apparent pyrites. Some pieces break irregularly, with bright shining, pitch-black, irregular surfaces.

No. 2693—Coal (sample 27). Near head of Child's creek of Clover Fork, 380 feet below sample 26. (The same as sample 20 (No. 2686), but from a better opening. This sample not as much weathered as that.)

Mostly splint coal, showing but little fibrous coal and no apparent pyrites. Some portions softer coal, breaking somewhat irregular, with bright irregular pitch-black surfaces.

No. 2694—COAL (sample 28). Right-hand Fork of Breeding Creek of Clover Fork. Collected by R. C. B. Thruston. (Splint coal 6 inches, shop coal 26 inches.)

Generally hard, dull-black, breaking into irregular laminæ with but little fibrous coal, and no apparent pyrites.

No. 2695—Coal (sample 29). Three feet six inches. Elkaner Winn. Branch of Clover Fork. The upper 15 inches is splint coal, showing some pyrites and fibrous coal. The rest is a fat coal, with more pyrites, sometimes in quantity together. Collected by R. C. B. Thruston.

Generally a pure-looking coal, mostly breaking with bright irregular surfaces, pitch-black. Some portions more dull, breaking into laminæ, with fibrous coal between. Some bright pyrites apparent.

No. 2696—Coal (sample 30). Low Gap Branch of Child's creek of Clover Fork, 16 feet above sample 20. Sample from the 16 and 3-inch layers in the bed. (So-called shop coal.) Collected by R. C. B. Thruston.

The bed shows six layers of coal, alternating with seams of shale. Sandstone roof.

No. 2697—Coal (sample 31, page 20). Head of Laurel Fork of Seagreaves creek of Clover Fork. Two hundred feet above

fossiliferous limestone. Sample from the upper 70-inch layer. Collected by R. C. B. Thruston.

The bed, with soil roof, has its 70-inch upper layer separated by 26 inches of shale from a lower 6-inch coal seam. Mostly splint coal badly weathered, with some clay intermixed.

No. 2698—Coal (sample 31, page 13). Low-gap Branch of Child's creek of Clover Fork of Cumberland river. Fifteen feet above sample 20. Sample from the 36-inch seam. Splint coal. Collected by R. C. B. Thruston.

Bed with five seams of coal, separated by four layers of shale, etc. Sandstone roof. Sample shows but little fibrous coal and no pyrites.

No. 2699—COAL. Green Jones', near mouth of Martin's Fork. Bed contains two seams of coal, 49½ and 13 inches severally. A pure-looking pitch-black coal, breaking irregularly cuboidal with irregular shining surfaces. A very little bright pyrites, but no fibrous coal apparent.

COMPOSITION OF HARLAN COUNTY COALS—Continued.

(Air Dried.)

Number in Report.	5689	2690	1697	2692	2693	2694	2692	2696	2697	2698	2699
Hygroscopic moisture.	1.80 33.60 64.60	2.46 31.94 65.60	4.50 33.00 62.50	2.50 38.60 58.90	2.00 36.80 61.20	2.06 37.34 60.60	1.80 87.10 61.10	35.00 63.00	3.10 36.10 60.80	2.16 34.14 63.70	1.68 35.72 62.60
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters.	35.40 9.20 9.40	34.40 61.20 4.40	87.50 58.90 3.60	41.10 53.04 5.86	38.80 55.86 5.34	39.40 57.70 2.90	38.90 56.70 4.40	37.00 62.20 10.80	89.20 56.60 4.20	36.30 <b>55</b> .50 8.20	37.40 60.20 2.40
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	0.821 spongy.	0.588 dense.	0.628 pulver ulent.	1.425 spongy.	0.958 light	0.574 spongy.	0.964 light spongy.	0.959 dense.	0.794 dense.	0.684 dense.	0.574 dense.
Color of the ash	(purp'h) nearly white.	light brown.	salmon	greyish brown.	purplish grey.	purplish grey.	lt. pur'h grey.	lt. pur'h grey.	greyish brown.	light grey.	light brown'h grey.

All of these coals are good, and most of them very good, they containing generally but small proportions of ash and sulphur. The best of them are Nos. 2699, 2694, 2691, 2697, 2690, 2695, 2693 and 2692, arranged in the order of their ash percentage.

The relative ash percentage, however, does not always show their exact comparative value as fuel, as may be seen in the following table:

Number.	Ash Percentage.	Volatile Combus- tible Matters.	Fixed Carbon.	Total Combusti- ble Matters.
2699	2.40	35.72	60.20	95.92
2694	2,90	37.34	57.70	95.04
2691	3.60	33.00	58.90	91.90
2697	4.20	36.10	56.60	92.70
2690	4.40	31.94	61.20	93.14
2695	4.40	37.10	56.70	93.80
2693	5.84	36.80	55.86	92.66
2992	5.86	38.60	53.04	91.64

No. 2692 contains the largest proportion of sulphur, viz.: 1.425 per cent., to be deducted from the total combustible matters; but it also contains the largest proportion of volatile combustible matters, viz.: 38.60 per cent.

Nos. 2696, 2689 and 2698, containing more ash materials than those above tabulated, yet are good fuels, containing large proportions of combustible matters, as shown below:

Number.	Ash Percentage.	Volatile Combus- tible Matters.	Fixed Carbon.	Total Combusti- ble Matters.
2696	10.80	35.00	52,20	87.20
2689	9.40	33,60	55.20	88.80
2698	8.20	34,14	55.50	89.64

The weathered samples, with their large proportions of hygroscopic moisture, of course show smaller proportions of combustible matters than may be found in the same coals deeper in the bed.

HARLAN COUNTY COALS—Concluded.

No. 2700—COAL (sample 33). Seagreaves creek of Clover Fork. Sample from the whole 39 inches. Collected by R. C. B.

Thruston. (Considerable pyrites and some fibrous coal low down in the series.)

The sample contains some splint coal, but is mostly coal which breaks irregularly cuboidal, with shining, irregular surfaces, having little or no pyrites or fibrous coal. (The upper 3 inches of the bed is splint coal.)

No. 2701—Coal (sample 34). Seagreaves creek of Clover Fork of Cumberland river, 400 feet below fossiliferous limestone. Sample from the 39 inches, the upper layer, 28 inches, being of shop-coal; the lower 11 inches, separated from this by 6 inches of indurated clay, is splint coal. Collected by R. C. B. Thruston.

A badly weathered sample, showing some fibrous coal but no pyrites.

No. 2702—Coal (sample 35). Fugit creek of Clover Fork. Sample from the main 41-inch layer of the bed The coal is mainly splint, containing very little fibrous coal or pyrites. Collected by R. C. B. Thruston.

A firm splint or block coal. Some pieces breaking irregularly with shining surfaces.

No. 2703—Coal (sample 36). Steep Hollow or Coal Branch of Clover Fork, on land of John M. Smith, 16.82 miles above Mount Pleasant. Sample from the 37-inch cannel coal. Collected by R. C. B. Thruston. This layer is at the bottom of the bed.

No. 2704—Coal (sample 37). White Oak Branch of Yocum's creek of Clover Fork. Sample from the 50-inch layer, which is nearly all shop-coal and hard. Collected by R. C. B. Thruston.

Seems to be generally a good firm splint or block coal.

- No. 2705—COAL (sample 38). Sample from the 25-inch layer, mainly rich shop-coal. Bottom of the same bed as the next preceding sample. Collected by R. C. B. Thruston.

  A good sample of firm semi-cannel or block coal.
- No. 2706—COAL (sample 39). Gray's Branch of Martin's Fork, 5 miles from Mount Pleasant. Sample from the 28

and 26-inch layers. Fat coking coal, except 3 inches of splint coal at the bottom of the 26-inch layer. Collected by R. C. B. Thruston.

A badly weathered sample. The bed contains four layers of coal separated by seams of shale.

No. 2707—COAL (sample 45). Frank's Branch of Yocum's creek of Clover Fork. Sample from the 48-inch seam. Collected by R. C. B. Thruston.

Some fibrous coal, but no pyrites apparent. A somewhat weathered sample.

No. 2708—Coal. John L. Cornett's, Poor Fork of Cumberland river. Average sample of coal sent for coking. By R. C. B. Thruston, January, 1886.

A somewhat dull-looking sample of splint coal, breaking into thin irregular laminæ. Some fibrous coal between, but no apparent pyrites.

- No. 2709—Coal. Milton Hensley's. Wallen's creek of Cumberland river. Average sample of the coal tested for coking. R. C. B. Thruston.
- No. 2710—Coal. S. S. Branson, Clover Lick creek of Poor Fork of Cumberland river. Average sample of coal tested for coking. R. C. B. Thruston.

COMPOSITION OF HARLAN COUNTY COALS-Concluded.

(Air Dried.)

Number in Report.	2700	2701	2702	2708	2704	2706	2706	2707	2708	2709	2710
Hygroscopic moisture.	1.52 34.88 68.60	4.64 29.16 66.20	1.56 35.04 63.40	1.56 42.64 55.80	1.74 31.76 66.50	1.56 35.24 63.20	2.98 32.02 65.00	1.44 33.56 65.00	1.26 29.20 69.54	2.20 36.70 61.10	1.80 87.10 61.60
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	36.40 60.70 2.90	33 80 55.20 11.00	36.60 2.80 60.80	44.20 46.48 9.32	33.50 67.10 9.40	36.80 58.00 5.20	35.00 55.30 9.70	35.00 62.80 2.20	30.46 56.68 12.86	38.90 58.86 2.24	38.40 58.24 3.36
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	o.615 0.986 dense pulveruspongy. ulent. brown'h brown'h grey.	0.986 pulveru- ulent. very lt. brown'h grey.	0.835 light spongy. light brown'h grey.	0.574 dense. grey'h brown.	0.656 dense spongy. light grey.	0.931 friable. very light grey.	0.692 pulver- ulent. light grey.	0.849 dense. light brown grey.	0.848 dense spongy. grey.	0.277 light spongy. light brown.	1.290 dense spongy. light brown. grey.

Several of these coals, especially those which give small proportions of ash, would yield very good coke, and even those with the largest percentage of ash material would be very good fuel for all ordinary purposes. Calculating their proportions of combustible materials, according to the method used in the table next preceding the above, even No. 2708, with its 12.86 per cent. of ash, contains 85.88 per cent. of combustible matters, including its 0.848 per cent. of sulphur, and No. 2706, with its 9.70 per cent. of ash, contains 87.32 per cent. of combustible matters, including 0.692 per cent. of sulphur. This, as well as Nos. 2701 and 2709, are weathered samples, and No. 2703 is a cannel coal, as is indicated by its large percentage of volatile combustible matters.

The very best of these coals are Nos. 2707, 2709, 2702 and 2700, arranged in the order of their ash percentage, the first having only 2.20 and the last named 2.90 per cent. These contain from 96.36 in No. 2707 to 95.58 per cent. in No 2700 of total combustible matters. The others fall in between these two extremes.

## HARLAN COUNTY COKES.

- No. 2711—Coke (No. 11). Made of Wallen's creek coal, Harlan county, Kentucky. Forty-eight hour coke made at Quinnimont, West Virginia, March 22 and 24, 1886. R. C. B. Thruston.
- No. 2712—(Coke No. 10). Made from coal of John L. Cornett, Island Branch of Poor Fork, Harlan county. Seventy-two hour coke made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.
- No. 2713—Coke (No. 9). Made from coal of S. S. Branson, Clover Lick creek of Poor Fork, of Cumberland river, Harlan county. Seventy-two hour coke, made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.
- No. 2714 (No. 11). Made of Whallen's creek coal, Harlan county. Seventy-two hour coke, made at Quinnimont, West Virginia, March 20-23, 1886. R. C. B. Thruston.

# COMPOSITION OF THESE HABLAN COUNTY COKES.

(Air Dried.)

Number in Raport.	2711	2712	2713	2714
Moisture, etc., expelled at red heat Fixed carbon	0,60 98,10 6,30	1.34 80.76 17.90	0.40 93.60 6.00	0.90 92.90 6.20
Totals	100.00	100.00	100.00	100,00
Percentage of sulphur	· 0.546 brownish grey.	0.648 light grey.	1.068 light grey brown.	0.368 grey. brown.
Hours of coking	48	72	72	72

With the exception of No. 2712, which contains too much ash material, these have the composition of very good cokes.

Their degree of porosity, or their power to resist pressure, were not tested, but they all appear to be good firm, dense cokes. It is a fact, however, known to most manufacturers of coke, that its density or porosity depends somewhat on the pressure which is applied to it in the oven when in its soft condition in the act of coking.

Hence, the lower portion of a bed of coke, which has been measurably coked by the heat of the oven before much weight of coal is applied, is generally more porous and spongy than that which is subsequently formed under the load of the full charge of coal. No doubt different coals, under similar conditions, will give cokes of very different degrees of density, and possibly, also, the length of time given to the process, and the mode of application of the heat, whether gradual or sudden, may affect the porosity and density of the coke.

### HICKMAN COUNTY.

#### CLAYS.

No. 2715—FIRE-CLAY. Bluff above Columbus, sixty-five feet above low water. Below gravel. Collected by John R. Procter.

A fine sandy clay, of a light grey color. Infusible before the blow-pipe.

## COMPOSITION .- (Air Dried.)

Silica	•		٠		•	٠	•	•			•	٠	•	•		٠			•				•		- 1	85.180
Alumina		•					•					•		•				•			 •	•			· 1	10.260
Iron peroxide												•			٠										. 1	1.120
Iron peroxide Lime																										trace.
Magnesia Potash																	80								. 1	.064
Potash																										. 954
Soda																									. 1	.146
Soda				•	•	•		•	•	•	•		•	•	•			•		•						2,276
																										100,000
Fine sand (per	 en	t.)						٠				2	ু											93	.	70,100

This so-called clay owes its plastic properties to the state of very fine division of the large quantity of siliceous sand which it contains, its 10.260 per cent. of alumina being equivalent to only 25.920 per cent. of kaolin, the basis of true clay.

It appears to be quite refractory in the fire, notwithstanding its nearly one per cent. of potash. No doubt it could find profitable applications: in the ceramic art; as a scouring material; in glass manufacture; or for fire-proof linings, etc.

No. 2715 (a). RED OCHREOUS CLAY. Below gravel. Columbus Bluff. Collected by John R. Procter.

Of a dull pink color. Calcines of a purplish-brown color. Quite plastic. Washed in water, it yielded 36.69 per cent. of fine sand.

It owes its color to hydrated peroxide of iron, like the ochres described under Ballard and Calloway counties, Nos. 2572 and 2651 (which see). It could no doubt be used as a cheap pigment (after washing it), either in its raw state or after calcination, or applied in terra-cotta works.

#### JEFFERSON COUNTY SOIL.

No. 2716—Soil (Surface). Collected by George Crum from a farm near Louisville. Characterized as a mucky soil, which fails to produce crops. Appears to belong to the Devonian black shale formation. The sample was sent to Prof. M. A. Scovell, Director of Agricultural Experiment Station (No. 25), at the State College of Kentucky, for analysis.

When received in a bottle it was wet and plastic like clay, but it dried readily when exposed in a dry atmosphere. The dried soil is of dirty drab color. Clods friable.

All passed through the coarse sieve except 5.7 per cent. of shot-iron ore. Dried at 212° F., it lost only 1.7 per cent. of hygroscopic moisture.

COMPOSITION .- (Dried at 212° F.)

						50-3				20-0										Per Cent.
Organic and volatile matters. Alumina Iron peroxide Carbonate of lime	•	•						•						ं			12		_	5.100
Alumina																		0		10,524
Iron peroxide	٠	•				1	•						•	0			ì	Ō	3	3.795
Carbonate of lime																				.140
Magnesia			•										•		•		-	•		.458
Phosphoric acid (P, O,)																	•	•		.176
Potash extracted by acids															•				. 1	.191
Soda extracted by acids																				.201
Water expelled at 380° F	•																			1,000
Magnesia	es		•	•	٠	•	•	•	•	•	•	•	•		•		•	٠		78,950
																				100.535
Hygroscopic moisture Potash in the siliceous residue Soda in the siliceous residue .																				1.700
Potash in the siliceous residue																				1.855
Soda in the siliceous residue.	•																	•		.229

The chemical composition of this soil is such as characterizes productive soils generally, with the exception that the alumina is in large proportion, and the lime somewhat below a good average; but its physical condition, causing it to be quite plastic and adhesive when wet, and no doubt its local position, where it is subjected to an excess of water, are unfavorable to productiveness.

Thorough draining to carry off the excess of water, and the free use of lime, would very probably make it productive.

# JOHNSON COUNTY.

COALS.

No. 2717—Coal. Walter Fletcher's, Little Paint creek. Cannel coal, from the upper twenty-two inches of the forty-three inch bed. Sample taken from the specimen at the Exposition.\* By R. C. B. Thruston. Collected by G. M. Hodge.

<sup>\*</sup>At New Orleans.

A pure-looking cannel coal. No apparent pyrites or fibrous coal.

No. 2718—COAL from same bed. Sample of the lower twenty-one inches. Bituminous or splint coal. By R. C. B. Thruston. Collected by G. M. Hodge.

A very pure-looking coal.

# COMPOSITION OF THESE COALS .- (Air Dried.)

Number	r i	n	Re	po	rt											2717	2718
Specific gravity					•		•			•						1.242	1.275
Hygroscopic moisture Volatile combustible matters Coke				•		•	•		•	•	•				:	1.44 50.22 48.34	2.56 39.94 57.50
Totals	•	•	٠	•	•	٠	•	•	•	٠	•	•	٠	•	•	100.00	100.00
Total volatile matters Fixed carbon in the coke	:	:	:	:	:	:	:	•	•	•	:	:	:	•	:	51.66 40.74 7.60	42.50 54.10 3.40
																0,837	1,030
Percentage of sulphur Character of the coke Color of the ash	:	:	:	•	:	•	:	:	:	:	:	:	:	•	•	dense. light buff.	spongy. light buff.

Both very good coals. The cannel coal is distinguished by its larger proportions of volatile combustible matters and ash, as well as by its lower specific gravity, notwithstanding its much larger ash percentage.

# KNOTT COUNTY COALS.

No. 2719—Coal on John Amburgy's land, Wolf Run Branch of Amburgy's Branch of Carr Fork of Kentucky river. Sampled for analysis from the specimen at the New Orleans Exposition (1885) by R. C. B. Thruston. Collected by G. M. Hodge. The upper portion of the bed bituminous coal. Apparently a much weathered sample. Friable, and soiled with earthy matter, which will increase its moisture, etc.

No 2720—Coal of the same bed, etc., the lower twenty-five inches. Cannel coal, etc.

A pure-looking cannel coal.

## COMPOSITION OF THESE KNOTT COUNTY COALS.

(Air Dried.)

Number	r i	n	Re	p	ort											2719	2720
Specific gravity												٠				1.355	1.206
Hygroscopic moisture Volatile combustible matters Coke											•					6.48 29.78 63.74	0.72 44.40 54.88
Totals	•	•	•	٠	•	1		•		•		ı.	٠	•	٠	100.00	100.00
Total volatile matters Fixed carbon in the coke					•		•				•		٠		•	36.26 60.64 3.10	45.12 47.00 7.88
Totals	•				•		•	٠		•		•				100,00	100.00
Percentage of sulphur Character of the coke Color of the ash	•	•	•	•			•	•	•	•		•		•	{	0.598 pulveru- lent. salmon colored.	0.753 dense. light buff.

Very good coals, exemplifying the distinctive characters of cannel and bituminous coals, with this qualification, that the bituminous coal, having been much more altered by "weathering" than the cannel coal, contains much more hygroscopic moisture, and gives a pulverulent coke instead of a spongy one, which it would no doubt yield if in the unweathered condition. Deeper in the bed, this layer of the coal would probably be found much better in quality.

### KNOX COUNTY COALS.—Continued.

No. 2721—COAL. Noah Wiggins', Fighting creek of Cumberland river. Average sample from the coal sent for coking. Collected by R. C. B. Thruston.

A bright splint coal, breaking with irregular shining surfaces. Some fibrous coal between the irregular laminæ.

No. 2722—Coal. Sandy Branch, near Flat Lick. A sample of the coal sent for coking. Collected two years ago. R. C. B. Thruston.

No. 2723—COAL. O. P. Ely's, Flat Lick. Average sample of the coal collected two years ago. Collected by R. C. B. Thruston.

No. 2724—Coal. Pursifull seam, near Flat Lick. Average sample of coal collected two years ago. R. C. B. Thruston.

#### COMPOSITION OF THESE KNOX COUNTY COALS.

(Air Dried.)

Number in Report.	2721	2722	2723	2724
Hygroscopic moisture	1.20 38.80 60.00	1.80 36.60 61.60	2.00 35.30 62.70	1.20 35.06 63.72
Totals	100.00	100.00	100.00	100.00
Total volatile matters	40:00 58.86 1.14	38.40 58.12 3.48	37.30 61.90 0.80	36,28 58,92 4,80
Totals	100.00	100.00	100.00	100.00
Percentage of sulphur	0.615 spongy. dark sal- mon col'd.	0.884 spongy. lt. brown grey.	0.766 dense. spongy. light brown.	0.766 light spongy. very light brown.

These are all remarkably good coals, which would yield good cokes. The exceptional small ash percentage of No. 2723 was verified by repetition with the same sample.

#### KNOX COUNTY COKES.

- No. 2725—Coke from Noah Wiggins' coal, Fighting creek, Cumberland river. Seventy-two hours' coke. Made at Quinnimont, West Virginia. R. C. B. Thruston.
- No. 2726—Coke from coal of Sandy Branch, near Flat Lick.

  A forty-eight hour coke. Made at Quinnimont, West Virginia. R. C. B. Thruston.
- No. 2727—Coke from O. P. Ely's coal, Flat Lick. Seventy-two hours' coke. Made at Quinnimont. R. C. B. Thruston.

No. 2728—Coke from coal of Pursifull seam, near Flat Lick. Seventy two hour coke. Made at Quinnimont. R. C. B. Thruston.

# COMPOSITION OF THESE KNOX COUNTY COKES. (Air Dried.)

	2725	2726	2727	2728
	0.90	0.40	0.50	0.90
	95.90	93,70	96.00	90.70
•	3.20	5.90	3.50	8.40
	100,00	100.00	100,00	100.00
. ;	0.546 reddish brown.	0.478 brownish grey.	0.217 brown.	0.546 brownish grey.
		0.90 95.90 3.20 100.00 0.546 reddish	0.90 0.40 95.90 93.70 3.20 5.90 100.00 100.00 0.546 0.478 brownish	0.90 0.40 0.50 95.90 93.70 96.00 3.20 5.90 3.50 100.00 100.00 100.00 0.546 0.478 0.217 reddish brownish

These are all remarkably good cokes, so far as chemical composition can determine.

### LAUREL COUNTY COALS.

No. 2729—Cannel Coal of Frederick Wiedmer, one mile from London, Laurel county. Sample brought to the laboratory by John R. Procter.

Somewhat laminated. Portions resembling bituminous shale. Showing some minute specks of mica.

No. 2730—Coal of the Pitman Coal Company, Pittsburg, Laurel county. Sample from the coal selected two years ago for coking. By R. C. B. Thruston.

GEOL. SUB.—5.

# COMPOSITION OF THESE LAUREL COUNTY COALS. (Air Dried.)

Numbe	er i	αi	Re	po	ort	•										2729	2730
Hygroscopic moisture Volatile combustible matters Coke	•	•	:	:	•	:	:	•	:	•	:	•	:	•	•	0.60 31.66 67.74	2.56 84.56 62.88
Totals	•	•	•	•	•	•		•	•	•		•	٠	•		100.00	100.00
Total volatile matters Fixed carbon in the coke Ash		•		•	٠	•	•	•		٠	•	•				32.26 45.24 22.50	37.12 59.58 3.30
Totals	•				•	•	•	•								100.00	100.00
Percentage of sulphur Character of the coke Color of the ash															· {	not est. dense. dark pur- plish-grey.	0.895 spongy. light grey.

No. 2730 is a very good, rich coal, well suited for the manufacture of coke, as the report below demonstrates.

No. 2729 may answer well for fuel in its own vicinity, but its high ash percentage might preclude distant transportation.

No. 2730a—Coke of the Pitman Coal Company's coal (No. 2730). Seventy-two hour coke. Made at Quinnimont West Virginia, March 20-27, 1886. R. C. B. Thruston.

### COMPOSITION. (Air Dried.)

Moisture, etc.,	expe	lec	1 8	ıt	T	<b>30</b>	п	eat				•		•	•				•		•			•	•	0.50
rixed carbon .			6 3																							92.60
Moisture, etc., e Fixed carbon . Brownish-grey	ash		0 1	•	٠	•	•	٠	•	•	•	•	•	•	•	•	•	٠	•	•	•		•	•	•	6.90
																										100.00
Percentage of a	ulph	ur				•		22							٠	્										0.61

This appears to be a very good coke. Much of the sulphur of the coal was burnt out in coking.

#### LAWRENCE COUNTY.

#### PETROLEUM.

No. 2731—Petroleum from the F. F. well, on Blain creek, land of the Vinson Oil Company, Lawrence county. Sample sent by John R. Procter. (Received January 9, 1885.)

A thick petroleum, of the consistence of molasses; of a brown color, from admixture of earthy matters, colored with iron peroxide and water, of which some separates on standing. It is an emulsion of dense petroleum with water containing earthy matters, colored with ferric oxide.

It was submitted to fractional distillation with following results:

The distillation was not completed because of the breaking of the flask and the limited quantity of the sample.

No. 2732—Petroleum from Lower Laurel well, 320 to 330 feet deep. Yields about 100 gallons per day. Well No. 1, on Lower Laurel creek, about 1 mile from the mouth of the creek. Sample sent to Prof. Crandall by W. R. M. Broas, of Ashland, November 15, 1885.

A brownish-black, rather thick fluid, of specific gravity 0.853, containing very little water.

It was submitted to distillation in an iron retort until 91.4 per cent. had passed over. This, called the *first distillate*, was submitted to fractional distillation, with results stated below, under the head *second distillation*. The residue in the iron retort, of specific gravity 0.860, is noted there as No. 7, or *first distillate*.

This first distillate, submitted to second distillation, gave the following products:

These products would fall nearly under the following commercial heads, viz:

No. 1. Mostly Gasoline and Naphtha.

No. 2. A. Naphtha.

No. 3. Kerosene, or so-called coal-oil.

No. 4. Mineral sperm oil

No. 5. Fine lubricating oil; spindle oil.

Nos. 6 and 7. Paraffine oils, from which heavy lubricating oils and solid paraffine may be obtained.

### LESLIE COUNTY.

#### COALS.

No. 2733-COAL. McClellan Schell's. Bed 73 inches thick. Upper Double Branch of Greasy creek. Collected by G. M. Hodge.

A weathered and somewhat soiled sample of what seems to be a good coal.

No. 2734—Coal. Nicholas Schell's. Bed 64 inches thick. Upper Double Branch of Greasy creek. Collected by G. M. Hodge.

Seems to be a splint coal. Sample somewhat weathered. Some little fibrous coal, but no pyrites apparent.

No. 2735—COAL. Bed 44 inches thick. Lewis' Branch of Greasy creek. The lower 7 inches was not sampled. Collected by G. M. Hodge.

A somewhat weathered sample of what seems to be a good splint coal.

No. 2736—Coal. Bed 83 inches thick. Head of Pace Trace, White Oak creek. Sample from the lower 53 inches. Collected by G. M. Hodge.

A weathered sample of what seems to be a good splint coal.

No. 2737—COAL. Bed 69 inches thick. William Sisemore's, Rockhouse creek. The bottom 6 inches not sampled. Collected by G. M. Hodge.

A pure-looking, firm coal, generally breaking irregularly, with irregular shining surfaces. A portion with lamellar fracture, and some fibrous coal; no pyrites apparent.

- No. 2738—COAL. Jesse Morgan's. Sixty-five inches thick. On Middle Fork. Collected by G. M. Hodge. Resembles the next preceding. Seems to have more splint
- coal.
- No. 2739—CANNEL COAL. Jerry Ledington's. Thirty-eight inches thick. Beech Fork, mouth of Oldham Branch. Collected by G. M. Hodge.
- No. 2740—Coal. Sixty one inches thick. Wesley McFadden's. Big creek. Red Bird creek. Collected by G. M. Hodge.
  - No pyrites apparent, and but little fibrous coal.
- No. 2741—Coal. Bed 57 inches thick. Richard Collins'. Hal's Fork. Big creek. Red Bird creek. Collected by G. M. Hodge.
  - No apparent pyrites, and but little fibrous coal.
- No. 2742—Coal. Bed 53 inches thick. Asper's Bank, Middle Fork of Kentucky river, two miles below Hyden. Collected by G. M. Hodge.
- Some fibrous coal between the laminæ, but no apparent pyrites.
- No. 2743—Coal. Bed 46 inches thick. Silas Nantz's, Oldhouse Branch, Beech Fork, Middle Fork of Kentucky river. Collected by G. M. Hodge.
- Seems to be somewhat weathered. Ferruginous incrustation on some pieces. Some fibrous coal apparent, but no pyrites.

COMPOSITION OF THESE LESLIE COUNTY COALS. (Air Dried.)

Number in Report.	2733	2784	2735	2736	2737	2738	2739	2740	2741	2742	2748
Specific gravity	1.342	1.363	1.251	1.509	1.279	1,291	not est.	1.822	1.286	1.321	1.602
Hygrometric moisture	3.20 29.70 67.10	. 35.68 62.60	1.72 35.02 63.26	9.40 32.20 58.40	98.20 63.20	0.70 84.70 64.60	1.10 44.20 54.70	1.60 84.94 63.46	1.40 35.68 62.92	1.80 34.14 64.06	1.80 82.86 66.34
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100,00
Total volatile matters	82.90 67.60 9.60	37.40 51.20 11.40	36.74 57.60 5.66	41.60 48.80 9.60	36.80 54.00 9.20	35.40 55.20 9.40	45.30 43.70 11.00	36.54 55.46 8.00	37.08 58.92 4.00	85.94 67.86 6.20	88.66 50.84 16.00
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	0.626 dense. lt. brn'h grey.	1.367 light spongy. lt. pur'h grey.	0.599 spongy. lt. brn'h grey.	0.433 pulverulent. nearly white.	1.307 spongy greyish- brown.	o.983 spongy. lt. pur'h grey.	0.690 dense. lt. grey- brown.	1.066 spongy. lilac- grey.	0.667 light spongy. lt.red'h grey.	0.613 dense. lt. lilac- grey.	1.409 dense spongy. lilac-grey.

Except Nos. 2741, 2735 and 2742, these Leslie county coals contain more than a good average proportion of ash material, these three, the best of all, giving, severally, only 4.00, 5.66 and 620 per cent. of ash; contain, severally, 94.60, 92.62 and 92.00 per cent. of total combustible matters, including small proportions of sulphur. The ash proportion in the other eight coals varies from 8 per cent. in No. 2740 up to 16 per cent. in No. 2743. By adding together the volatile combustible matters and fixed carbon in the coke, it will be seen that the total combustible matters in these eight coals varies from 81.00 in No. 2736 to 90.40 per cent. in No. 2740. So that the worst of all contains 81 per cent. of combustible matters, including 0.433 per cent. of sulphur. It may be stated, however, that the sample of this particular coal was from the very much weathered outcrop of the bed, and contained as much as 9.40 per cent. of moisture, and that this, as well as the other coals which had been weathered, will be found much better taken deeper in the bed.

It will be noticed that those coals which have the greatest proportions of ash materials have also the largest percentage of sulphur, generally speaking. The cannel coal No. 2739 is to be distinguished by its largest proportion of volatile combustible matters. Coals with large proportions of ash material are not available for the manufacture of good coke; but cannel coals, although containing large proportions of ash materials, may yet be available in gas manufacture because of their large proportions of volatile combustible matters.

#### LETCHER COUNTY.

COALS.

No. 2744—COAL. Right Hand Branch of Right Hand Fork of Collier's creek, 4 miles above its mouth. Two hundred and sixty feet above a limestone. Forty-one inches of coal on 2 inches of cannel shale. Collected by R. C. B Thruston. A badly weathered outcrop sample; a somewhat mixed splint coal.

No. 2745—Coal. Wilson Lewis', Roland's Branch of Poor Fork of Cumberland river. Average sample collected for coking. By R. C. B. Thruston.

# COMPOSITION OF THESE LETCHER COUNTY COALS. (Air Dried.)

Number in Report.											2744	2745					
Hygroscopic moisture																2.40	1.40
Volatile combustible matters																32.60	33.40
Coke	•	•	٠	•	•	•	•	•	•	•	•	•	•	٠	•	65.00	65,20
Totals	•	•	٠	•	•	•		•	•	•						100.00	100.00
Total volatile matters																35,00	34.80
Fixed carbon in the coke																62.44	59.0
A.e.h															•	2,56	6.15
Totals	•	٠	٠	٠	•	•	•	•	•	•	•	•	٠	٠		100.00	100.00
Percentage of sulphur																0.492	1.426
Character of the coke																dense friable.	spongy.
Color of the ash		•						•				•	•		{	lt. salmon- colored.	light grey.

Both good coals. The weathered sample does not probably represent the entire character of the coal deeper in the bed, where it would contain less hygroscopic moisture and more sulphur. The analysis of the coke made from coal No. 2745 is given below.

No. 2746—Coke (No. 22). From Wilson Lewis' coal (No. 2745), Roland's Branch of Poor Fork. Seventy-two hour coke. Made at Quinnimont, West Virginia. R. C. B. Thruston.

#### COMPOSITION. (Air Dried.)

Moisture expelled at red heat					0.00	•											•		2.5	0.70
Fixed carbon																				89.30
Fixed carbon	•	•	٠	٠		•	•	٠	٠	•	•	٠	•	•	•	٠	•	٠	٠	10.00
																				100,00
Percentage of sulphur				•															ু	0.863

The percentage of ash is somewhat above good average, but not excessive.

#### MARSHALL COUNTY.

Soils.

No. 2747.-VIBGIN SOIL of Crawfish Flats of Tennessee River Valley, 2 miles below the mouth of Jonathan creek. Sample taken to the depth of 8 inches Growth: hickory and red and post oaks and sassafras. Collected by R. H. Loughridge.

Dried soil in friable clods of a dirty light grey-buff color, with dark infiltrations. The coarse sieve\* separated from it about 8.2 per cent. of fine shot-iron ore. Its siliceous residue, from digestion in acids, all passed through the fine sieve† except 0.15 per cent. of fine hyaline sand.

No. 2748—Subsoil of the next preceding soil, etc.

Color somewhat lighter than that. Clods firmer. All passed through the coarse sieve except about 3.9 per cent. of shotiron ore. Its siliceous residue passed through the fine sieve, except 3.15 per cent. of fine white sand.

No. 2749—VIRGIN SOIL from bottom land of East Fork of Clark's river, near Benton. Sample taken 10 inches deep. Growth: mostly white oak. Soil: crawfishy. Collected by R. H. Loughridge.

The dried soil, in friable clods, is of a dark buff-grey color. The coarse sieve separated from it about 5.8 per cent. of shot-iron ore. Its siliceous residue left 3.725 per cent. of fine white sand on the fine sieve.

No. 2750—VIRGIN SOIL of Post Oak Flats of Clark's river, north of Stringtown. Sample taken to the depth of 12 inches. Crawfishy in character. Growth: post oak chiefly. The coarse sieve removed from it about 4 per cent. of shotiron ore. Its siliceous residue left 1.05 per cent. of fine white sand on the fine sieve.

No. 2751—Virgin Soil. Dark sandy loam from low ridge of Tennessee River Valley, 2 miles below the mouth of Jonathan creek. Sample taken to the depth of 12 inches.

<sup>\*</sup>Coarse sieve has 64 meshes to the centimeter square.

<sup>†</sup>Fine sieve has 1600 meshes to the centimeter square.

Growth: white and Spanish oaks and dogwood. Collected by R. H. Loughridge.

Dried soil in moderately firm clods of a light grey-brown color. All passed through the coarse sieve except about 0.4 per cent. of shot-iron ore. The fine sieve separated from its siliceous residue 1.5 per cent. of fine white sand.

No. 2752—VIRGIN, CRAWFISHY OR GLADY SOIL of Tennessee Valley, 1 mile north of Calvert City. Sample taken to the depth of 10 inches. Collected by R. H. Loughridge.

Dried soil, in pretty firm clods, of a grey-buff color. All passed through the coarse sieve except small fragments of sassafras roots and a little shot-iron ore. The fine sieve removed from its siliceous residue 1.825 per cent. of fine white sand.

No. 2753—VIRGIN LOAM SOIL of low ridge of Tennessee River Valley, 1 mile north of Birmingham. Sample taken to the depth of 12 inches. Growth: oak and sassafras. Collected by R. H. Loughridge.

Dried soil, in moderately firm clods, of a light grey-brown color. All passed through the coarse sieve except a small angular fragment of red jasper and a few grains of shot-iron ore. Only a few grains of fine sand were separated from the siliceous residue by the fine sieve.

No. 2754—FLATWOOD VIRGIN SOIL, four miles south-west of Benton, on the Harvey road. Sample taken to the depth of 8 inches. Growth: red and white oaks, hickory and sassafras. Collected by R. H. Loughridge.

Dried soil, with friable clods, of a light grey-brown color. All passed through the coarse sieve except a very small quantity of shot-iron ore and vegetable debris. Insoluble silicates all passed through the fine sieve.

No. 2755—Subsoil of the next preceding. Sample taken at the depth of from 8 to 14 inches. Collected by R. H. Loughridge.

Dried soil in moderately firm clods, of a light grey-brown color. All passed through the coarse sieve. Its siliceous residue all passed through the fine sieve.

No. 2756—WHITE SILTY SOIL. Middle Fork of Clark's river. Quaternary. Collected by R. H. Loughridge.

Dried soil of a light grey-buff color. All passed through the coarse sieve except about 0.44 per cent. of shot-iron ore and ferruginous sandstone. The fine sieve separated only a very small quantity of fine white sand from its siliceous residue.

No. 2757—Bottom Soil of West Fork of Clark's river, near Brewer's old mill. Sample taken to the depth of 10 inches. Growth: exclusively beech. Rich yellow loam soil. Collected by R. H. Loughridge.

Dried soil of a grey-brown color. Clods moderately firm. All passed through the coarse sieve except a very small quantity of shot-iron ore.

COMPOSITION OF THESE MARSHALL COUNTY SOILS.

(Calculated Dried at 212° F.)

Number in Report.	2747	2748	2749	2750	2751	2012	2763	2764	2765	2756	2757
Organic and volatile matters	3.443	2.878	3.757	3.845	8.467	3.506	4.411	2 949	0 880	9 199	708 8
Alumina and manganese oxide	8.303	5.021	3.706	4.623	4.876	8.080	4.448	8.671	1.087	3.875	8.248
Iron peroxide	1.955	. 5.680	1.360	3.089	2.887	2.087			7.667	2.980	1.764
Lime carbonate	020	truce.	970	920.	9.5 9.5	trace.			.046	.046	.146
Under House Committee Comm	196	1.00	148	. 18 1.	194	. 281			.827	8	28.
Potash extract d by acids.	.047	919	60.0	477	192	2 6	99.	55.	070	not est.	
Sodn extracted by acids	not est.	not est	210	189	not est.	046		-	.048	not est.	ne t est.
Water expelled at 380° F	.607	722.	.683	224	.581	609		974	182	.841	1.028
Sand and insoluble silicates	90.217	88.877	90.070	87.057	88.460	89.648	86.678	89 452	89,463	88.884	89.757
Totals	99.895	99.892	100.047	99.733	100.210	99,467	100.087	99.558	99.786	99.182	99.773
Bygroscupic moisture	1.300	1.100	1.620	1.800	1 226	1.900	1 400	1.300	1 800	1 860	7 300
Potash in siliceous residue	1.544	0.515	1.488	1.284	1.600	1 506	1.614	1.484	1.462	1.681	1.604
Sand	0.150	3,150	8.726	1 050		1.825	trace.	0	084.0	trace.	not est.
Character of the soil	Virgin	Subsoil	Virgin	Virgin	Virgin	Virgin	Virgin	Virgin	Subsoil	Sile	Rottom
						8		9			Topone.

All of these Marshall county soils are very deficient in lime except No. 2757, which contains only 0.146 per cent., which may be considered rather below a good average proportion. They would no doubt all be benefited by top-dressings of lime. Several of them only gave mere traces of this essential element of vegetable food.

Potash is deficient in Nos. 2747, 2749 and 2755, but is in quantity above a good average in Nos. 2748, 2750 and 2752, and in good average proportions in the rest of the soils.

Phosphoric acid is somewhat below average proportions in Nos. 2753 and 2757, and very deficient in Nos. 2754 and 2755.

All these deficiencies could be supplied by the application of appropriate commercial fertilizers or otherwise, and then even the worst of these soils could be made and kept fertile, with good husbandry, if they are sufficiently drained; they being all light and porous, made up of very finely divided earthy materials, without any coarse sand or gravel.

No. 2758—WHITE SAND. Cretaceous. North bank of Clark's river, 2 miles north of Benton, Marshall county. Collected by R. H. Loughridge.

A fine-grained sand; white, with a slight tinge of yellowish. Contains small mica specks. It all passed through the coarse sieve. Its insoluble residue, after digestion in acids, is mostly fine white hyaline quartz sand, containing some small mica scales.

COMPOSITION OF THIS MARSHALL COUNTY SAND. (Calculated Dried at 212° F.)

Organic and volatile matters.														22		•	•			0.39
Alumina, iron oxide, etc Lime carbonate		•					•				•	•				٠				.32
Lime carbonate		•	٠		20		•	•	•			•	٠		•		•			a trace
Magnesia		•					•				•		•				*			.01
Magnesia																				.09
Potash extracted by acids																				.189
Potash extracted by acids Soda extracted by acids																	2			l
Water expelled at 400° F		•																		.05
Sand, etc	•	ě		•	•	•	•	•	٠	÷		•	•				•	•		98.98
																			8	100.05
Tygroscopic moisture							•								•				3	0.05
Hygroscopic moisture Potash in siliceous residue Soda in siliceous residue																				.00
oda in siliceous residue							•00									•				.00

This sand might be made useful in glass-making and in empering tough clays for refractory pottery or fire-bricks.

#### MARSHALL COUNTY.

CLAYS.

No. 2759—CLAY. Burradell's. Five miles north of Benton. Collected by R. H. Loughridge.

Dried clay of a light grey color. Sandy. Infusible before the blow-pipe. Calcines hard. Washed in water, left 70.33 per cent. of fine whitish sand, containing small mica scales.

No. 2760—CLAY from Gray's place, near Scale, Marshall county. Collected by R. H. Loughridge.

A whitish clay, with a little yellow ochre in spots and layers. Before the blow-pipe it fused light-colored. Contains no sand.

No. 2761—Ash-colored Earth from a cistern on the old Winter's place, 8 miles south-east of Olive Post-office. (Of what value is it?) Collected by R. H. Loughridge.

Nearly white. Infusible before the blow-pipe. Calcines white and hard. Washed in water, left about 33.7 per cent. of light grey sand.

- No. 2762—CLAY. Banks of Tennessee river, at Highland. Geological position, Port Hudson. Collected by R. H. Loughridge.
- Of a light grey-brown color. Contains some small blue specks (iron phosphide), scantily diffused. Fuses before the blow-pipe. A similar clay, with blue particles of iron phosphide (*Vivianite*), is described under McCracken county, No. 2781, which gave decided reaction of phosphoric acid.
- No. 2763—PIPE-CLAY. J. T. Pugh's place, 2 miles east of Palma. Tertiary formation. Collected by R. H. Loughridge.
- Of very light brownish grey color. Contains minute mica specks. Before the blow-pipe it fuses with great difficulty.
- No. 2764—Decomposing White Chert of siliceous Subcarboniferous limestone, 4 miles west of Birmingham, on Mrs. Lou. Stone's place. Collected by R. H. Loughridge.

# COMPOSITION OF THESE MARSHALL COUNTY CLAYS.

(Air Dried.)

Number in Report.	2759	2760	2761	2762	2768	2764
Silica	84.580	52,580	91,580	60,980	62,920	93,700
Alumina	10.650	31.070 1.510	5.980 .220	18.480 7.500	29,880	3,580
Lime	.137	.137	.045	.780	trace.	trace.
Magnesia	. 101	.245	.055	1.128	.209	.127
Potash	.954	1.775	.094	2.264	1.564	.618
Soda	.292	.318	.032	.627	.172	.117
Water, etc	2.956	12.365	1.994	7.841	5.255	1.858
Totals	100.000	100,000	100.000	100,000	100,000	100.000
Sand	70,330	0	33,700	not est.	not est.	0

These clays are all quite refractory, except No. 2762, which melts comparatively easy before the blow-pipe, and No. 2760, which fuses with difficulty. This latter clay would serve for the manufacture of white pottery ware. The first named could be used for colored terra-cotta, etc. No 2763 is quite a refractory white clay, and Nos. 2761 and 2764, the latter designated as decomposing chert or hornstone, and the former being of a similar origin probably, are both remarkable because of their very large proportions of silica, which is so very finely divided that they possess the plastic property of clay, and calcine hard. They could probably be utilized as scouring materials, in the form called Bath bricks, or otherwise. They are very refractory. They are more properly designated siliceous earths than clays.

#### MARTIN COUNTY.

COALS.

No. 2765—COAL. Head of Scafford's Lick Branch of Rock-castle creek, on land of the Rockcastle Mining and Lumber Company. Sampled for analysis by R. C. B. Thruston from the section of the bed exhibited at the New Orleans Exposition. Collected by G. M. Hodge from above the 26-inch parting.

A somewhat weathered sample of pure-looking splint coal.

No. 2766—Coal from the same bed section. Collected by G. M. Hodge. Sampled from the part below the 26-inch parting. By R. C. B. Thruston.

Apparently more weathered than the preceding. Splint coal, showing fibrous coal, but no apparent pyrites, between its thin laminæ. Much of the sample is in a pulverulent condition.

#### COMPOSITION OF THESE MARTIN COUNTY COALS.

(Air Dried.)

Numbe	r i	n	Re	epo	ori											2765	2766
Specific gravity		•				•										1.344	1.451
Hygroscopic moisture Volatile combustible matters.	. :		:		:		:	:	•	:		:				2.92 84.98	2.62 27.98
Coke														•	:	62.10	69,40
Totals	٠	•	•	•	•	•		٠	•	•		•	٠			100.00	100.00
Total volatile matters Fixed carbon in the coke																37.90 55.30	30.60 50.40
Ash	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	6.80	19.00
Totals	•		•	•	•	•	•	•	•	•	•	•		•	•	100.00	100.00
Percentage of sulphur Character of the ccke																0.681 dense spongy.	0.736 friable.
Color of the ash											•						light grey.

The coal of the upper portion, with its 6.80 per cent. of ash, contains 90.28 per cent. of combustible matters, including its sulphur, while that of the lower seam contains only 78.38 per cent., with its 19 per cent. of ash. This latter coal would not bear long transportation. Both samples seem to show weathering, and it is probable that deeper in the bed the coal will be found to be somewhat purer.

#### MCCRACKEN COUNTY

Soils.

No. 2767—Soil (virgin). Oak and Hickory Flatlands. Albert Bradshaw's land, south-west of Paducah. Sample taken to

the depth of 8 inches. Timber: white and red oaks, hickory, dogwood and sassafras. Collected by R. H. Loughridge.

Dried soil of a brownish-grey color. Clods friable. All passed through the coarse sieve\* except a small quantity of vegetable debris. Its siliceous residue, from digestion in acids, all passed through the fine sieve† except a very small quantity of very fine hyaline sand.

No. 2768—Subsoil of the next preceding. Sample taken at the depth of 8 to 15 inches. Collected by R. H. Loughridge. Dried subsoil of a lighter color than the surface soil, and more yellowish. Its small clods were firm. All passed through the coarse sieve except a small quantity of vegetable debris. Its siliceous residue passed through the fine sieve except 2.75 per cent. of fine white quartz sand.

No. 2769—Virgin Soil. Brown loam. New Hope Church. South-west corner of McCracken county. Sample taken to the depth of 6 inches. Timber: white and red oaks, hickory, dogwood and sassafras. Collected by R. H. Loughridge.

Dried soil of a light grey-brown color. Clods friable. All passed the coarse sieve\* except a small portion of vegetable debris. Its siliceous residue, from digestion in acids, all passed through the fine sieve† except a few small grains of hyaline quartz sand.

No. 2770—Subsoil of the next preceding. Taken at the depth of from 6 to 12 inches. Collected by R. H. Loughridge.

Dried subsoil of a grey-buff color. Clods friable. All passed through the coarse sieve except a little shot-iron ore and a small quartz pebble. Its siliceous residue passed through the fine sieve.

No. 2771—VIRGIN SOIL of Red Oak Barrens, 4 miles east of Woodville. Sample taken to the depth of 8 inches. Growth: red oaks, 10 to 15 feet high. Collected by R. H. Loughridge. Dried soil of a grey-brown color. Clods friable. All passed

<sup>\*</sup> Of 64 meshes to the centimeter square.

<sup>†</sup> Of 1600 meshes to the centimeter squa e. GEOL. SUB.—6.

through the coarse sieve. The fine sieve separated a small quantity of fine hyaline quartz sand from its siliceous residue.

No. 2772—Subsoil of the next preceding. Taken from 8 to 12 inches deep. Collected by R. H. Loughridge.

Dried subsoil of a brownish-buff color. Lighter colored than the preceding. Clods friable. The coarse sieve removed from it 3.6 per cent. of shot-iron ore. All its siliceous residue passed through the fine sieve except a few small grains of quartz sand.

No. 2773—VIRGIN SOIL of Post Oak Flatwoods, 1 mile west of Paducah, on Hinkleville road. Taken 10 inches deep. Growth: chiefly post oak, a little red oak and hickory. Collected by R. H. Loughridge.

No. 2774—Subsoil of the next preceding. Sample taken from 10 to 14 inches deep. Collected by R. H. Loughridge.

Dried subsoil, yellowish-grey. Clods quite friable. The coarse sieve removed from it 3.1 per cent. of shot-iron ore. Its siliceous residue all passed through the fine sieve except a small quantity of fine white sand.

No. 2775—Virgin Soil. Fine silty ash-colored soil of the Post Oak Flatwoods, near the county line, north of Woodville. Sample taken to the depth of 8 inches. Growth: post and red oaks. Collected by R. H. Loughridge.

Dried soil of a light grey or ashy color. Clods friable. All passed through the coarse sieve except 7.1 per cent. of shotiron ore. Its siliceous residue all passed through the fine sieve.

No. 2776—Subsoil of the next preceding. Taken 6 to 12 inches below the surface. Collected by R. H. Loughridge.

Dried subsoil, very light grey. Its small clods are somewhat firm. The coarse sieve separated from it 5.4 per cent. of shot-iron ore. Its siliceous residue all passed through the fine sieve except a few grains of fine sand.

COMPOSITION OF THESE MCCRACKEN COUNTY SOILS.

# (Calculated Dried at 212° F.)

Number in Report.	2767	2768	2769	2770	2771	2772	2773	2774	2775	2776
Organic and volatile matter	4.885	8.781	8.549	2.365	8.868	2.885	2.807	2.309	2.088	1.637
Iron peroxide	. 8 . 8 . 8	8.416	3.160	1.880	4.089 2.768	4.941 2.883	5.336 3.086	8.982 8. <b>44</b> 1	2.818	8.295 1.950
Magnesia		25.	.197	121	. 134	.097	.127	.097	171.	.161
Phosphoric acid (P. O.) Potash extracted by acids	.110	.086	960.	.078	121.	360.	.095	.079	.062	88
Soda extracted by acids	119	88.	.583	not est.	805	168	.087	not est.	.146	not est.
Water expense at 400 f	88.288	26.78	90.050	91, 147	. 662	88.	.946	.761	.588	.463
Totals	99,689	809.66	100.519		100.877		100.258		100.416	100 588
Hygroscopic moisture .	2.750	2.820	1.265	1.100	1.655	1.500	1.500	1.700	0.875	1.060
Soda in siliceous residue	.480	.500	1.050	.758	.518	1.262	704	.718	1.446	1.566
Character of the soil	Virgin. Subsoil.	Subsoil.	Virgin.	Subsoil.	Virgin.	Subsoil.	Virgin.	Subsoil.	Virgin.	Subsoil.
				-						

No. 2767 contains an average proportion of organic and volatile matters, as well as good percentages of the other essential elements, with the one exception of magnesia. This is not likely, however, to prevent it from being a quite fertile soil, if the physical conditions, drainage, etc., are favorable. All the others, except No. 2771, contain less than a good average of phosphoric acid. Nos. 2775 and 2776 are especially deficient in this essential ingredient. Phosphatic fertilizers would be beneficial to all these, and especially to the latter mentioned soils. Lime is generally present in good average proportions, except in subsoils Nos. 2772 and 2774, and potash is in ample proportions in all. These soils are all composed of very finely divided materials, none of them containing any gravel or coarse sand. They ought all to be productive, with good management and economy of manures, if well drained, and the other physical conditions are favorable.

#### McCracken County Clays.

No. 2777—CLAY. Mr. Jones', 3 miles south of Paducah. Collected by R. H. Loughridge.

Dried clay of a light grey color. Fused with difficulty before the blow-pipe. Calcined white.

No. 2778—CLAY on John Mitchell's place, 3 miles east of Lovelaceville. Collected by R. H. Loughridge.

Dried clay of a light brownish-grey color. Fused grey before the blow-pipe.

No. 2779—GYPSEOUS CLAY from Mr. W. J. Hough's place, 4 miles west of Paducah. Collected by R. H. Loughridge.

Dried clay, nearly white, with ferruginous stains in spots. Contains a very small proportion of sand. Fuses slightly before the blow-pipe. Calcines white.

No. 2780—Blue Micaceous Clay. Tertiary; on Armstrong's place, 7 miles east of Paducah. Collected by R. H. Lough ridge.

Dried clay of a grey color. Contains minute specks of mica. Before the blow-pipe it fuses grey.

No. 2781—Blue Micaceous Clay, charged with blue particles of Vivianite. (See No. 2762 for a similar clay in Marshall county.) Collected by R. H. Loughridge.

Darker colored than the preceding; 56.60 per cent. of brownish sand, containing small mica scales, was washed out of it. The blue specks gave a marked reaction of phosphoric acid. Before the blow-pipe this clay fused grey.

COMPOSITION OF THESE MCCRACKEN COUNTY CLAYS.

(Air Dried.)

Number in Report.	2777	2778	2779	2780	2781
Silica	59,500	66,320	67.580	69.220	73,192
Alumina	24.960	22.930	20.040	17.540	16.540
Iron peroxide	.720	1.190	.540	1.440	1.840
Lime	. 325	. 437	1.743	. 437	. 369
Magnesia	.396	.209	.158	. 858	.461
Potash	1.934	1.107	1.340	2.452	1.969
Soda	.286	.470	.075	.472	.541
Water, etc	11.879	7.337	8.524	7.581	5,088
Totals	100,000	100,000	100.000	100,000	100.000
Sand	not est.	not est.	small pro-	50.580	56,600

Nos. 2777 and 2779 are good plastic clays, suitable for the manufacture of white or whitish pottery ware. They are also quite refractory. Nos. 2778 and 2780, containing more iron peroxide, are rather less refractory, and calcine somewhat colored.

The large proportions of fine sand in Nos. 2780 and 2781 somewhat diminishes their plasticity, and their considerable percentages of iron peroxide and potash renders them fusible at a high temperature. Possibly very fine sand exists in some of the others, not easily removed by washing.

The Vivianite in No. 2781 is in too small quantity to make it available for phosphatic manure. No. 2778 contains some of its lime in the form of sulphate, but not in large proportion.

#### OLDHAM COUNTY.

No. 2782—MINERAL WATER from Anita Springs, residence of Dr. G. T. Berry, 1 mile from Lagrange. Flows from the cavernous layer of the Clinton Group of the Upper Silurian. Collected by W. M. Linney, September, 1885.

The water is transparent, colorless and inodorous. On evaporation it left 0.274 of a gramme of saline matters to the litre (about 1,000 grammes) of the water. These consist mainly of carbonates of lime and magnesia, sulphates of potash and soda, chloride of magnesium and silica. It is a pure weak saline water.

#### PERRY COUNTY COALS.

No. 2783—COAL. Sixty-one inches. John Fields'. Sampled by John T. Proffit from the lower 36 inches. Collected by G. M. Hodge.

A weathered sample of splint coal. Some fibrous coal between the thin laminæ, but no appearance of pyrites. Some ferruginous incrustation.

No. 2784—Cannel Coal. Ten inches (in the same bed 24 inches of bituminous coal), 1 mile below Rush Branch, Mid dle Fork Kentucky river. Collected by G. M. Hodge.

A somewhat weathered sample. Ferruginous incrustation on some of the surfaces.

No. 2785—Coal. Forty-four inches. William Boling's, Rush Branch of Middle Fork of Kentucky river. Sample of the upper 20 inches. Collected by G. M. Hodge.

A pure-looking coal. No apparent pyrites. Some little fibrous coal.

- No 2786—COAL Sample of the lower 20 inches of the same bed as the above. Resembles that coal, but is somewhat brighter.
- No. 2787—COAL. Fifty-six inches. Abner Campbell's. Fish Trap Branch of North Fork of Kentucky river. Collected by G. M. Hodge.

Some portions dull, like cannel coal; others bright. Some fibrous coal between the laminæ, but no apparent pyrites.

No. 2788—COAL. Forty-five inches. Thomas J. Johnson's, Ebersole Branch of Middle Fork of Kentucky river. Collected by G. M. Hodge.

Apparently a splint coal, somewhat weathered. Some fibrous coal between the laminæ, but no apparent pyrites.

No. 2789—COAL. Fifty two inches. John Spencer's, Grapevine creek. Sample of the lower 48 inches. Collected by G. M. Hodge.

Generally dull black. Fibrous coal and some little granular pyrites between the laminæ. Some portions of bright pitch-black.

No. 2790—Coal. Forty-eight inches, mouth of Guy's creek, Middle Fork of Kentucky river. Sample of the upper 32 inches. Collected by G. M. Hodge.

Generally dull black splint coal. Some fine fibrous pyrites and fibrous coal. Portions shining pitch-black.

No. 2791—COAL. Forty inches. John Spencer, Grape-vine creek. Collected by G. M. Hodge.

A somewhat weathered sample of splint coal.

No. 2792—Coal. Thirty nine inches. Joseph Campbell's, mouth of Rock Lick Branch of North Fork of Kentucky river. Collected by G. M. Hodge.

A weathered sample of splint coal.

No. 2793—COAL. Thirty-nine inches. Alexander Combs', North Fork of Kentucky river, 5 miles below Hazard. Collected by G. M. Hodge.

Apparently good splint or semi-bituminous coal. No apparent pyrites.

No. 2794—COAL. Forty five inches. Samuel Whitaker's, Willard creek. Collected by G. M. Hodge

Portions of the sample dull splint coal. Some fibrous coal between the laminæ, but no apparent pyrites. Some pieces bright pitch-black.

No. 2795—Coal. Thirty six inches. Peter Gross'. Mine near the mouth of Squabble creek, Middle Fork of Kentucky river. Collected by G. M. Hodge.

Generally pitch-black coal, breaking irregularly with irregular shining surfaces. A few pieces dull and laminated. No pyrites apparent, and but very little fibrous coal.

COMPOSITION OF THESE PERRY COUNTY COALS.

(Air Dried.)

Number in Report.	2783	2784	<b>*</b> 785	3786	2787	2788	2789	2790	1672	2792	2793	279	2795
Specific gravity	1.333	not est.	1.279	1.300	1.359	1.334	99: -	1.366	not est.	not est.	1.390	1.390	8.259
Hygroscopio moisture	35.30	0 <del>1 2</del> 0 8 9 9 9 9 9	39.60	35.90	30.34	£ 4.20 8.00 8.00 8.00	55.36 65.34	31.86	3.32 3.32	8.0° 67.7° 8.8°	36.2 26.24 30.24 30.24	63.8.06 3.8.06	37.16 61.86
Total	100.00	80'00	108.8	100.00	100.00	100.00	10.00	100.00	130.8	100.00	180.00	18.8	18.8
Total volatile matters. Fixed carbon in the coke.	38.80 53.14 8.06	45.60 37.60 16.80	40.80 52.70 6.50	37.10 55.30	35.60	38.20 52.20 9.60	34.70	34.40 55.30	36.80 47.80 15.40	32.40 58.50 9.10	37.80 56.20 6.80	36 80 52.80 10.40	39.00 57.10 3.10
Total	100.00	100.00	100.00	100.08	100.00	100.00	100 00	100.00	100.00	100.00	100.00	90.00	100.00
Percentage of sulphur	1.035 dense. lt.brown- ish-grey	o.970 pulveru- lent. grey- brown	spongy. light lilac- grey.	o.654 light spongy. nearly white.	friable. It. purlp'h-	o. 763 dense. friable. purplish- grey.	0.450 friable. very light grey.	o. 557 friable. very light grey.	o. 491 pulveru- lent. very light grey.	o. 505 friable. purplish- grey	o. 557 light spongy. very light grey.	o. 722 friable. purplish-	o.749 spongy. It. purp'h- grey.

Of these Perry county coals only one, No. 2784, is a cannel coal, characterized in its composition by its larger percentage of volatile combustible matters than in the other samples, which are mostly splint or semi-bituminous coals. This cannel coal gives also a larger proportion of ash than the other coals, viz: 16.8 per cent.; yet the sum of its combustible materials is 82.4 per cent. The purest of all these samples is No 2795, which gives only 3.1 per cent. of ash, and its total combustible matters amount to 95.0 per cent. As can be seen, several of these coals contain more than a good percentage of ash, which, however, will not prevent them from being valuable for all ordinary purposes, although not so available for making good coke as others which contain less earthy material. The sulphur percentage in these coals is generally moderate.

#### PULASKI COUNTY COALS.

No. 2796—COAL. Barren Fork. The run of the mine. Sample of the coal taken for coking two years ago. By R. C. B. Thruston.

No. 2797—Coal of the Barren Fork Coal and Mining Company. Nut coal. Average sample of the coal taken for coking two years ago. By R. C. B. Thruston

COMPOSITION OF THESE PULASKI COUNTY COALS.
(Air Dried.)

Numbe	r i	n	Re	epe	ort	į										2796	2797
Hygroscopic moisture Volatile combustible matters																2.32	2.12
Coke		:		•	•	:	:		:	:	:	:	:		:	32.48 65.20	$\frac{31.56}{66.32}$
Totals	•		٠					2	٠	•	•			•	•	100.00	100.00
Total volatile matters Fixed carbon in the coke			•								•		•	•		34.80	33.68
Ash	:	:	:	:	•	:		•		•	•	:	:	:: ::		59.10 6.10	59.02 7.30
Totals	٠	•	•	٠	•	•	•		•							100.00	100.00
Percentage of sulphur Character of the coke			:	:		:										1.077 spongy.	1.879
Color of the ash																light grey.	

Pretty good coals, containing 91.58 and 90.58 per cent. of total combustible matters severally. The ash and sulphur percentages are somewhat above a very good average for coking purposes.

# WARREN COUNTY.

No. 2798—MARL from near the bottom of the Saint Louis limestone beds, near Bowling Green, about 100 feet above Barren river. Sanple from the top, middle and bottom of a 4 foot bed. Average sample. Collected by M. H. Crump, Civil Engineer, April 30, 1885.

Of a light grey color, portions with ferruginous stains. Plastic with water. Powder of a light buff color. Calcines nearly white, with a light reddish tint.

### COMPOSITION. (Air Dried.)

Silica										•				•		700	996		28,780
Alumina												::#		•30				0.00	11.631
fron peroxide						1.5			•	•			11-0	•	900				2.960
Lime carbonate .																		.	43.760
Magnesia carbona	te														2			.	6.320
Magnesia carbona Phosphoric acid (1	P.	0	.)														0	. 1	.249
Potash			٠.												0				2.124
Soda																-	2		not e∗t.
Soda																÷	-		4.176
																		- 1	
																		- 1	100,000

This marl would be beneficial as a top-dressing to light sandy soil which is deficient in lime. It could also be, no doubt, converted into water cement of the character of Portland cement, by proper admixture with more lime and calcination.

This marl may be said to contain a little more than 43 per cent. of clay (exclusive of moisture), and nearly 44 per cent. of carbonate of lime. Good artificial Portland cement is said to be made, in England, etc., from a mixture of 21 to 23 per cent. of clay, and 79 to 77 per cent. of chalk (which is nearly pure carbonate of lime). Ground limestone could be used, and the best proportions found by actual experiment.

No. 2798 a-Mineral Water, from a bored well 502 feet deep; beginning about the middle of the St. Louis lime.

stone. Bored by Captain Smallhouse. Sample sent by M. A. Crump, Civil Engineer, Bowling Green.

The water, contained in a stoneware jug, was turbid with a yellowish material. Smelt strongly of hydrogen sulphide and petroleum; a little of the latter substance floating on it.

Qualitatively examined, it was found to contain chlorides, sulphates and sulphides of magnesium, calcium, sodium, &c. Its total solid saline matters is 5.09 per cent., most of which is sodium chloride—common salt. A pretty strong saline sulphur water

#### WHITLEY COUNTY COALS.

No. 2799—Coal. Possum Branch of Wolf creek. Collected by A. R. Crandall. Three feet of coal, under a sandstone ledge; on farm of Joseph Cox.

A bright, pure-looking coal, showing very little fibrous coal, and no apparent pyrites.

No. 2800—Coal. Forks of Jellico creek. Jellico Creek Salt Works, at R. P. Crickmore's. Upper 38 inches of the main coal. Sample collected by A. R. Crandall.

A bright, pure-looking sample; fracture generally shining and somewhat irregular. Some few thin scales of bright pyrites, but very little fibrous coal apparent. Contains a few pieces of lamellar structure and dull appearance.

No. 2801—Coal. Right Fork of Bennett's creek. Coal 48 inches thick. Sample from the whole thickness, without parting.

A block coal or splint. Mostly bright, pure-looking coal, of irregular, shining fracture. A small portion more dull, and lamellar in structure. No pyrites or fibrous coal apparent.

No. 2802—Coal. Right Fork of Patterson creek. Pleasant Palley's coal. Fifty-two inches thick. Sample from the whole bed. Collected by A. R. Crandall.

A pure-looking coal, breaking generally with irregular fracture and shining surfaces; some pieces showing "bird's-eye" fracture with satiny lustre. A small portion lamellated. Hardly any fibrous coal and no pyrites apparent.

No. 2803—Coal. Briar creek. W. Jones' coal, 28 inches thick. Sample from the whole face of the bed. Collected by A. R. Crandall.

Apparently a pure splint or block coal. No pyrites, and but little fibrous coal apparent.

No. 2804—Coal. Briar creek. Henry Green's coal, 3 feet thick. Sample from the whole thickness, from 20 feet in the entry. Collected by A. R. Crandall.

A pure-looking, pitch-black coal, breaking generally with shining, irregular surfaces. But little fibrous coal and no pyrites apparent.

No. 2805—COAL. Briar creek. Richardson's coal, 36 inches thick without parting. Average sample from the whole face. Collected by A. R. Crandall.

Quite a pure-looking coal, of the character of the next preceding.

No. 2806—Coal. Head of Little Patterson creek. M. A. Jones' coal. Average thickness, 3 feet without parting. Collected by A. R. Crandall.

Quite a pure-looking, pitch-black coal, breaking with irregular shining surfaces. Portions lamellated with some fibrous coal, but no apparent pyrites.

No. 2807—Coal. Tackett's creek. Main coal. Terrel Siler's.

Two miles from the mouth, and half a mile above Mr.

Siler's house. Bed 41 inches thick, with a half-inch parting 16 inches from the bottom. Sample collected by A. R. Crandall.

Resembles the preceding.

No. 2808—COAL. Mahan Station. Average sample of the upper 23 inches. (A parting of 5 inches.) Sample collected by A. R. Crandall.

Coal resembles the preceding.

COMPOSITION OF THE TEN PRECEDING WHITLEY COUNTY COALS. (Air Dried.)

Number in Report.	2799	2800	2801	2802	2803	7804	2805	2806	2807	2808
Specific gravity	1.308	1.286	1.282	1.279	1.271	1.281	1.414	1.262	1.277	1.321
Hygroscopic moisture	1.84 83.84 64.32	2.00 81.30 66.70	2.40 35.80 61.80	2.14 39.76 58.10	36.10 62.00	1.90 37.40 60.70	1.94 38 92 59.14	2.14 36.06 61.80	2.40 86.60 61.00	2.20 33.10 64.70
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	35.68 59.96 4.86	33.30 62.94 8.76	38.20 58.96 2.84	41.90 55.06 3.04	38.00 59.50 2.50	39.30 58.60 2.10	40.86 56.84 2.30	38.20 59.20 2.60	39.00 57.70 3.30	35.30 56.80 7.90
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	2.180 light spongy. libac.	0.901 light spongy. white.	0.556 light spongy. It. br.n'h grey.	0.892 light spongy. It. br'n'h grey.	8 0.997 0.928 0.764 0.755 spongy. spongy. spongy. spongy. tt. br'n'h lt. br'n'h lt. br'n'h st. br'n'h grey. grey.	o.928 spongy. it. br'n'h grey.	spongy. It. br'n'h grey.	o.755 spongy. lt. br'n'h grey.	0.708 dense spongy. It. br'n'h grey.	4.234 light spongy. dark purplish.

Most of these are remarkably pure coals, their ash percentages extending from 2.10 per cent. in No. 2804 up to 7.90 in No. 2808. This latter is exceptional, not only in its much greater ash proportion, but in the extraordinary amount of sulphur it contains, viz: 4.234 per cent. This must have been present in the sample as very fine granular iron pyrites, as was rendered probable by the large amount of iron contained in its ash. Its total combustible matters, including the sulphur, is 89.9 per cent.

No. 2804 contains 96 per cent. of combustible matters, with its 2.1 per cent. of ash.

The ash proportions vary, in the other samples, from 3.76 down to 2.30 per cent. These are all, except No. 2808, remarkably good, pure coals, and no doubt would make very good coke. As several of them are somewhat soft coals, approaching so-called bituminous coals, their coke may be light and spongy unless made under pressure.

#### WHITLEY COUNTY COALS-Continued.

No. 2809—Coal. Mahan's coal. Mahan's Station; eighteen inches below the parting. Average sample. Collected by A. R. Crandall.

Resembles the next preceding.

No. 2810—Coal. W. M. Mahan's coal. Mahan's Station. Eighteen inches of the bottom; 13 inches of the top, with a 5-inch parting. Average sample of the whole bed. Collected by A. R. Crandall.

A pure looking, pitch-black coal; breaking with shining irregular surfaces. Very little fibrous coal, and no pyrites apparent.

No. 2811—Coal. Caddell's. Wolf creek. Bed 37 inches thick, with a slight parting. Sample from near the outcrop. Collected by A. R. Crandall.

A somewhat weathered sample of what appears to be a good coal.

No. 2812—COAL. Thomas' cannel coal. Head of Left Fork of Wolf creek. Splint coal, with some cannel at the bot-

tom. Twenty-six inches near the outcrop. Sample collected by A. R. Crandall.

A much weathered sample.

No. 2813—COAL. Lower 26 inches of the R. P. Crickmore coal. (Upper part previously sampled.) Collected by A. R. Crandall.

A bright, pure looking, pitch-black coal; breaking generally with irregular shining surfaces. Very little fibrous coal, and no pyrites apparent.

No. 2814—COAL. B. P. Shelby's. Upper 22 inches. Collected by A. R. Crandall.

A pure-looking sample of what seems to be a splint coal. But little fibrous coal, and no pyrites apparent.

No. 2815—Coal. Wagner Siler's. On Mud Creek. Forty inches thick, without parting. Average sample from the whole face of the bed. Collected by A. R. Crandall.

A pure-looking coal; generally breaking with irregular shining surfaces. Some portions lamellated, with some little fibrous coal, but no apparent pyrites between.

No. 2816—Coal. J. S. Berry's. Jellico seam. Sample from coal collected two years ago for coking. By R. C. B. Thruston.

COMPOSITION OF THESE WHITLEY COUNTY COALS. (Air Dried.)

Number in Report.	5808	2810	2811	2812	2813	2814	2816	2816
Specific gravity	1 271	1.811	1.369	1.887	1.268	1 841	1 298	•
Hygroscopic moisture	2.20	36.70	1.50	7.26	1.70	2.24	2,68	2.00
Ooke	62.94	61.50	67.94	68.90	60.90	62.10	64.30	63.18
Totals	100.00	100.00	100.00	100.00	100.00	. 100.00	100.00	100.00
		00	90 64	9. 1	0.00			
Total volatile matters	58.14	88.90	42.06 51.24	68.20	39.10 59.36	87.90 58.76	86.70 62.70	86.82
<b>A</b> sh	08· <del>*</del>	3.00	6.70	6.70	1.64	3,34	1.60	2.70
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	1.340	1.093	2.768	0.640	1.721	1.060	0.670	0.629
Character of the coke	spongy.	spongy.	dense spongy	•	spongy.	light spongy.	light spongy. light spongy.	light spongy.
Color of the ash	grey.	plish-grey.	purple.	grey-brown.	plish-grey.	salmon color.	salmon color. light brown.	brown-grey.

These are good, and some of them very good coals. The highest ash percentage—in No. 2811—is 6.70. This approaches cannel coal in composition, and contains 91.80 per cent. of combustible matters, including 2.768 per cent. of sulphur, a large proportion of the latter.

The lowest ash proportions in these coals is in Nos. 2813 and 2815, being only 1.54 and 1.60 per cent. severally.

No. 2813 contains 96.76 per cent. of combustible matters, including 1.721 per cent. of sulphur.

No. 2815 contains 95.82 per cent. of combustible matters, including 0.670 per cent. of sulphur.

No. 2816 contains 95.30 per cent. of combustible matters, including 0.629 per cent. of sulphur.

No. 2810 contains 95.20 per cent. of combustible matters, including 1.093 per cent. of sulphur.

No. 2814 contains 94.42 per cent. of combustible matters, including 1.060 per cent. of sulphur.

No. 2809 contains 93.00 per cent. of combustible matters, including 1.340 per cent. of sulphur.

All of these six coals could no doubt be made available in the manufacture of coke. Those would, of course, be preferred which contained the least sulphur and ash. An example of the coke made from one of these (No. 2816) is given below.

No. 2817—Coke. (No. 14.) Of J. S. Berry's coal. Jellico series. Five miles north of Williamsburg, Whitley county. Seventy-two hours' coke. Made at Quinnimont, West Virginia, March 20th-23d. R. C. B. Thruston.

# COMPOSITION .-- (Air Dried).

Moisture, etc., expelled at Fixed carbon		red	. 1	hea	t	•	Ť.	•	•						•									1.50
Fixed carbon	9 1		8		•			9	:	13		•	9a - 3	•		1	•	•	•	•	•		•	93.30
Fixed carbon Light brownish-grey ash	•	•	•	•	•	٠	٠	*	*	٠	•	*	•	•	÷	•	•	•	•	•	•		•	5.20
																								100 00
Percentage of sulphur .			٠			•	٠	٠				20			ı,		•					33		0.382

Comparing its sulphur percentage with that of the coal from which it was made, it will be seen that a large propor-

tion of it was removed in the process of coking. This, no doubt, is the case with most coals; the quantity removed varying according to the condition or form of combination of the sulphur in the coal.

# APPENDIX.

For comparison with Kentucky cokes and coking coals, a number of samples of both, of West Virginia, were collected by Mr. Thruston, the results of the analyses of which are given in the following pages:

## WEST VIRGINIA COALS.

- No. 2818—COAL. (Sample 11.) Three miles above the mouth of Preacher creek, or Callahan creek, of Rolling Fork of Powell's river, Wise county. Sample of the 27-inch cannel coal of the bed. Collected by R. C. B. Thruston. A dull-looking cannel coal.
- No. 2819—COAL. Quinnimont Mines, Fayette county. Thickness of bed where sampled, 34 inches. Collected by R. C. B. Thruston.
- A bright, pure-looking coal. Fracture irregular, with shining surfaces. No pyrites or fibrous coal apparent.
- No. 2820—COAL. Fire creek, Fayette county. Collected by R. C. B. Thruston.
- No. 2821—COAL. Stone Cliff, Fayette county. Upper seam, above the conglomerate; bed 4 feet 3 inches to 5 feet. Sample from the 4 feet 3 inches. Collected by R. C. B. Thruston.
- A pure-looking, pitch black coal. Somewhat laminated. Fracture generally irregular, with shining surfaces.

- No. 2822—Coal. Stone Cliff, Fayette county. Quinnimont seam. Coal 44 to 52 inches; no parting. Sample from where it measures 46 inches. Collected by R. C. B. Thruston.
- No. 2823—COAL. Sewell, Fayette county. On top of the conglomerate. From 34 to 42 inches thick. Collected by R. C. B. Thurston.

A pure-looking, soft coal. Irregular fracture, with shining surfaces. Fibrous coal, but no pyrites apparent.

No. 2824—COAL. Hawksnest, Fayette county, West Virginia. Sample exclusive of parting. Collected by R. C. B. Thruston.

A pure-looking, pitch-black coal. Fracture irregularly cuboidal, with shining surfaces. Some little fibrous coal apparent.

COMPOSITION OF THESE WEST VIRGINIA COALS.

(Air Dried.)

Number in Report.	2818	2819	7820	2821	2822	2823	2824
Hygroscopic moisture Volutile combustible matters. Coke	0.86 38.54 60 60	0.40 18.00 81.60	0.60 20.80 78.60	0.60 22 80 76.60	0 90 19.10 80.00	0.80 22.14 77.06	1.06 32.34 66.60
Totals	100.00	100.00	100.00	100 00	100.00	100.00	100.00
Total volatile matters	39 40 47.84 12.76	18.40 79.00 2.60	21.40 75.34 3.26	23 40 74.20 2.40	20.00 73.60 6.40	22.94 74 46 2.60	33.40 62.00 4.60
Totals	100 00	100 00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	0.711	0.491	0.418	0 491	0.848	0.464	0.629
Character of the coke	dense friable. chocolato brown.	much inflated. light brownish grey.	light spongy. light buff-grey.	light spongy. light buff-grey.	light spongy. light buff-grey.	much inflated. light brown- ish-grey.	spongy. light buff-grey.

With the exception of No. 2818, which is a cannel coal, and No. 2824, these West Virginia coals are quite remarkable because of their small proportions of volatile combustible matters, and generally small ash, as well as the very light or inflated cokes they leave when heated under the atmospheric pressure only. They also present the singular fact that their cokes are of very difficult incineration.

#### WEST VIRGINIA COKES.

- No. 2825—Coke. Forty-eight hours. Made at Quinnimont, of Quinnimont coal. March 22d-24th, 1886. R. C. B. Thruston.
- No. 2826—Coke. Of Hawksnest coal, Fayette county. Made in Solendorf Coppe oven, at Quinnimont. Collected by R. C. B. Thruston.
- No. 2827—Coke. Forty-eight hours. Of Fire creek coal, Fayette county. Made at Quinnimont. March 26th, 1886. Collected by R. C. B. Thruston.
- No. 2828—Coke. Forty-eight hours. Of Stone Cliff coal, Fayette county. Made at Quinnimont. Collected by R. C. B. Thruston.
- No. 2829—Coke. Forty-eight hours. Of Stone Cliff coal, Fayette county. Made at Quinnimont. March 16th, 1886. Collected by R. C. B. Thruston.
- No. 2830—Coke. Seventy-two hours. Of Stone Cliff coal, Fayette county. Made at Quinnimont. Collected by R. C. B. Thruston.
- No. 2831—Coke. Forty-eight hours. Of Sewell coal. Made at Quinnimont. Collected by R. C. B. Thruston.
- No. 2832—Coke. Seventy-two hours. Made of Sewell coal, at Quinnimont. Collected by R. C. B. Thruston.

COMPOSITION OF THESE WEST VIRGINIA COKES.

(Air Dried.)

Number in Report.	2825	2826	2827	7878	2829	2830	2831	2832
Moisture expelled at red heat. Fixed carbon	92.20 92.00 5.80	90.40 90.50	1.70 92.90 5.40	0.86 92.74 6.40	0 86 96.40 3.24	1.60 89.80 8.60	0.50 92.90 6 60	1.60 91 80 6.60
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100 00	100.00
Percentage of surphur	0.491	0.711	0.401	0.437	0.519	0.821	0.560	0 478
Color of the ash	brownish- grey.	light grey- brown.	brownish- grey.	light buff-grey.	light red- dish brown.	light grey-brown.	light brown- ish-grey.	light brown-grey.

These are, generally, quite good cokes, No. 2830 being the only one which presents ash and sulphur a little above good average proportions.

The variation in the proportions of moisture is no doubt accidental. The purest of them all is No. 2829, containing 96.40 per cent. of fixed carbon, and only 3.24 per cent. of ash, with 0.519 of sulphur.

#### BELL COUNTY.

#### COALS.

No. 2833—Coal. (Sample 1.) On John R. Slusher's land, Left Fork of Straight creek, Bell county. Collected by R. C. B. Thruston, June 2d, 1886.

A weathered sample, incrusted in part with ferruginous material. Sample from the upper two seams, 6 and 31 inches in thickness, severally, separated by 1 inch of shale. Another seam of 5 inches of coal, separated from these two beds by 5 inches of shale, was not sampled.

#### COMPOSITION .- (Air Dried.)

Per cent.	Per cent.
Hygroscopic moisture 1.40	Total volatile matters 38 50
Dense spongy coke 61.50	Fixed carbon in the coke
100.00	100.00
Percentage of sulphur 1.151	

A good semi-bituminous coal, containing rather more volatile combustible matters than the very best coking coal.

No. 2834—Coal. (Sample 2.) James Green's, on the divide between Browney's and Hansee's creeks, Bell county. Collected by R. C. B. Thruston, July 13th, 1886.

A pure-looking, pitch-black coal, generally of cuboidal structure; breaking with irregular, shining surfaces. Some portions somewhat lamellated, and of somewhat dull fracture. Very little fibrous coal, and no pyrites apparent.

This sample is of the upper seam of 38 inches, separated, in this bed, by a 20-feet space. from the lower two seams of coal.

No. 2835—COAL. (Sample 12.) From James Green's lower bank. Sample from the 26-inch seam of coal lying 20 feet below the coal from which sample 2 (No. 2834) was taken, and separated from coal of sample 13 (No. 2836) by a 2-inch parting of shale. Sample collected by R. C. B. Thruston, September 24th, 1886.

A pure-looking, pitch-black, hard coal, resembling the brighter, harder portions of the next preceding sample.

No. 2836—COAL. (Sample 13.) From James Green's bank. Sample from the 21-inch seam lying under coal of sample 12 (No. 2835), and separated from that by a 2-inch layer of shale. Collected by R. C. B. Thruston, September 24th, 1886.

A bright, pitch-black, firm coal; structure cuboidal; breaking with irregular, shining surfaces. No fibrous coal or pyrites apparent.

COMPOSITION OF THESE COALS OF JAMES GREEN'S BANK.
(Air Dried.)

Number in Report.	2834	2835	2836
Hygroscopic moisture	0.60 35.00 64.40	1.20 32.20 66.60	1.00
Totals	100.00	100 00	100.00
Total volatile matters	35.60 61 90 2 50	33.40 61.60 5 00	36.40 61.70 1.90
Totals	100.00	100 . 00	100.00
Percentage of sulphur	1 068 very dense. light grey-brown.	0.574 inflated. brownish-grey.	0.629 inflated. light salmon- colored.

These are remarkably good, pure coals, containing very small amounts of ash material, and small proportions of sulphur, more especially the latter two described. They could no doubt be used for the manufacture of coke, although they contain rather more volatile combustible matters than the most profitable coking coals.

No. 2837—Coal. (Sample 3a.) Fifteen miles above Pineville, on Muddy Branch of Clear creek, Bell county. Sample of the upper 34 inches of the bed. Collected by R. C. B. Thruston, July 15th, 1886.

A somewhat weathered sample of splint, or semi-cannel coal.

No. 2838—COAL from lower part, 43 inches in thickness, of the same bed as the next preceding. Collected by R. C. B. Thruston, July 15th, 1886.

A much soiled and weathered sample of cannel coal.

#### COMPOSITION OF THESE TWO COALS .- (Air Dried.)

Numbe	r i	n	Re	po	rt											2837	2838
Hygrometric moisture Volatile combustible matters		•					:	•			•					1.70 32.60	1 00 51.60
Coke																65 70	47.40
Totals	•	•		•			÷				٠.		·			100.00	100.00
Total volatile matters Fixed carbon in the coke	•		•	•		•							٠	•		34.30	52.60
Fixed carbon in the coke Ash	:	:: ::	•	:	•	•	•	•		:	•	•	•		:	62.30 3.40	,10.40 7.00
Totals			٠	٠				٠		٠	×	٠		•		100.00	100.00
Percentage of sulphur		٠,								•			٠			0.684	0.739
Character of the coke															;	spongy.	dense.
Color of the ash	•					•	•	•	•	•	٠	•	٠	٠	{	light grey- brown.	brown

These are both very good coals of their kind, and no doubt, when mined deeper in the bed, would leave less ash than these weathered samples. The cannel coal is very rich in volatile combustible matters. The upper layer would no doubt furnish good coke.

No. 2839—Coal. (Sample 4.) On James Bussell's land, on Cannon creek, Bell county. Collected by R. C. B. Thruston, July 20th, 1886. Two seams of coal, 25 and 23 inches thick severally, separated by 17 inches of shale, lie under a shale roof, on a clay floor.

A weathered and soiled sample.

#### COMPOSITION .- (Air Dried.)

Hygroscopic moisture Volatile combustible matters	$\begin{pmatrix} 0.80 \\ 33.90 \end{pmatrix}$ Total volatile matters 34.70
Light spongy coke	65.30 Carbon in the coke
	100.00
**	
Percentage of sulphur	1.508

This is a very good splint or semi-cannel coal, verging toward bituminous coal, containing quite a moderate proportion of ash material, and but little more than the average percentage of sulphur.

No. 2840—Coal. (Sample 9.) Feelan Risner's, on Browney's creek, Bell county. Sample collected by R. C. B. Thruston, September 22d, 1886.

A weathered sample.

No. 2841—COAL. (Sample 10.) From the same bed as the next preceding sample, there being only a 1-inch parting of shale between them.

#### COMPOSITION OF THESE TWO SAMPLES .- (Air Dried.)

Number	r i	n	R	ep	ort											2840	2841
Hygroscopic moisture			•		_											5.40	1.60
Volatile combustible matters																30.00	47.40
Coke																64.60	51.00
Totals	•	•	•	*	٠	ė	•	•	٠	•	*	•	٠	10		100.00	100.00
Total volatile matters				•										•0		35.40	49.00
Fixed carbon in the coke		•		•	•					•	•			•		60.50	47.70
Ash	٠	•	•	•	٠	٠	٠	٠	٠	٠			•	•	*	4.10	3.30
Totals			•	•			•	•	•	•	•			ě		100.00	100.00
Percentage of sulphur						٠										0.436	0.574
Character of the coke			•	•	•	•	•		•	•	•		•	•	{	Tent.	pulveru- lent.
Color of the ash		٠	٠	٠	•	÷	٠	٠	•	•	•	•	•	•	{	d'k salmon colored.	dark brown.

Both these samples had been somewhat altered by long weathering, the effects of which would be the reduction of the

volatile combustible matters and sulphur, with probably an increase of the ash percentage. The removal of the volatile ingredients caused the production of a pulverulent coke. Deeper in the bed, where the coal has not been altered by long weathering, the coke of No. 2840 would no doubt be spongy coherent. No. 2841 has yet the large percentage of volatile combustible matters which characterizes the cannel coals.

No. 2842—Coal. (Sample 5.) Head of Middle Fork of Williams' Branch of Yellow creek, Bell county. Collected by R. C. B. Thruston, July 24th, 1886.

This sample is from four several seams, which are, respectively, beginning at the top, 17, 12, 11, and 25 inches thick, divided by shale partings, which are, severally, 7, 2, and 2 inches thick. Another bed of coal, not included in the sample, lies beneath, under a thicker shale parting.

This mixed sample was generally pitch-black coal, of cuboidal structure, breaking with shining, irregular surfaces, containing very little coal of lamellar structure, very little fibrous coal, and no apparent pyrites.

#### COMPOSITION -(Air Dried.)

		Per cent.					Pe	r cent.
Hygroscopic moisture Volatile combustible matters	•	${1.60 \atop 33.20}$ Total volatile matters .		٠				34.80
Light spongy coke		65.20 { Carbon in the coke Light grey-brown ash	•	•	•	•	•	60,60 4,60
		100.00						100.00
Percentage of sulphur		0.876						48

This sample has the composition of a good coking coal.

No. 2843—Coal. (Sample 6.) Land of J. M. Robbins, on Crane creek of Yellow creek, Bell county. Sample by R. C. B. Thruston, from the upper 36-inch seam of the bed, September 21st, 1886.

A pure-looking, pitch-black coal; structure mostly cuboidal. Fracture irregular, with shining surfaces. Very little fibrous coal, and no pyrites apparent.

No. 2844-Coal. (Sample 7.) From the lower 23-inch seam

of the same bed as the next preceding sample, separated from that by a 3-inch shale parting.

A pure-looking, pitch-black coal: breaking with irregular, shining surfaces. Very little fibrous coal, and no pyrites apparent.

COMPOSITION	OF THE	SE COALS	-(Air Dried.)
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2	V u	ml	be	r i	n	Re	po	rt											2843	2844
Hygroscopic moisture					•														1.40	1.20
Volatile combustible m	att	er	s																32.30	34.90
Zoke					٠									•	•				66.30	63.90
Totals	•																×		100.00	100.00
Cotal volatile matters			•													•			33.70	36.10
Fixed carbon in the co																		•	63.50	62.10
Asb	•	•	•	•	٠	٠		•	•	•	•	•	•	٠	٠	٠	•	٠	2.80	1.80
Totals	•	٠		٠	٠	٠	٠	•	•	•	٠		÷	9.	(*)	•			100.00	100.00
Percentage of sulphur Character of the coke			334																0.670	0.529
Character of the coke			•					•				•	٠						dense.	spongy.
Color of the ash	•	•	•	•	•	٠	•	•	٠	•	•	•	•	•		•	•	}	lt. brown- grey.	salmon- colored.

Both these coals are remarkably pure and good, containing only very small proportions of ash material and sulphur. They would doubtless produce very good coke.

No. 2845—COAL. (Sample 8.) From John M. Marricle's bank, on Cubbog Branch of Browney's creek, Bell county. Collected by R. C. B. Thruston, September 22d, 1886. Coal, 39 inches: the upper 3 inches of bone coal; lowest 12 inches, soft coal; gas coal, 24 inches, between.

A pretty pure-looking sample, mainly of splint or semicannel coal. Portions breaking irregularly, with shining surfaces. But little fibrous coal or pyrites apparent.

No. 2846—Coal. (Sample 11.) On Richard Risner's land, on Hansee's creek, Bell county. Collected by R. C. B. Thruston, September 24th, 1886. Sample from the two layers, severally 16 and 53 inches thick, separated by a 1-inch shale parting, the roof being the soil. A much weathered sample.

No. 2847—Coal. (Sample 14.) On Andrew Marricle's land, on Cubbog creek, Bell county. Collected by R. C. B. Thruston, October 6th, 1886. Sample from the two layers, severally 21 and 25 inches thick, separated by a 1-inch parting, with shale roof and clay floor.

A weathered sample.

No. 2848—Coal. (Sample 15.) On Stone Coal Branch of Browney's creek. Sample from the upper four layers, severally of 23, 14, 2½, and 20 inches thickness, separated by thin shale partings; shale roof and clay floor. Another 3 inch coal and a 5-inch shale parting lie below these above the floor.

A weathered sample.

No. 2849—Coal. (Sample 16.) Near the head of Browney's creek, Bell county. Forty-four inches, including half an inch of shale 8 inches from the top. Collected by R. C. B. Thruston, October 11th, 1886.

A firm, pure-looking, pitch-black coal. Structure generally cuboidal. Fracture with irregular, shining surfaces. Little or no fibrous coal, and no pyrites apparent.

COMPOSITION OF THESE FIVE BELL COUNTY COALS .- (Air Dried.)

Number in Report.	2845	2846	2847	2848	2849
Hygroscopic moisture	0.80	2.20	4.10	2.60	1.40
Volatile combustible mat'rs. Coke	35.30 63.90	32.60 65.20	29.90 66.00	33.20 64.20	34.60 64.00
Totals	100.00	100.00	100.00	100.00	100.00
Total volatile matters	36 10	34.80	* (	35 80	36.00
Fixed carbon in the coke .  Ash	60 20 3 70	61.80 3.40	61.80 4.20	59.60 4.60	61.80
Totals	100 00	100.00	100.00	100.00	100 00
Per cent. of sulphur	0.425	0.601	0.789	0.931	0.821
Character of the coke . {	spongy.	pulveru- lent.	dense friable.	dense spongy.	dense spongy.
Color of the ash {	light grey- brown.	salmon- colored.	d'k salmon colored.	light brown.	d'k salmon colored.

These are all very good coals. The samples, Nos. 2846, 2847, and 2848, show one of the effects of weathering in their large proportions of moisture. They all would give very good coke, their ash percentage being all below a good average, No. 2849 giving an exceptionally low ash.

### BELL COUNTY COALS.—(Continued.)

No. 2850—Coal. On Four-mile creek of Cumberland river, Bell county. Geological position: Lower Sandy Branch seam. Bed 37 inches thick. Average sample collected by R. C. Ballard Thruston, July 1st, 1887.

A very pure-looking, semi-cannel coal. No pyrites apparent in the sample. Breaking in laminæ, generally with shining, irregular surfaces. Very little fibrous coal apparent.

No. 2851—Coal. (Sample 2.) One and a half miles up Four-mile creek of Cumberland river, Bell county. Geological position: 100 feet above Sand Branch coal. Collected July 1st, 1887, by R. C. B. Thruston.

Apparently a weathered sample of splint coal. This bed has four seams of coal, 4 to 28 inches thick severally; separated by three shale partings, which are severally from 4 to 17 inches thick.

- No. 2852—Coal. (Sample 3.) Four-mile creek of Cumber-land river. Geological position: upper Sandy Branch seam. Collected by R. C. B. Thruston, July 1st, 1887.

  A weathered sample of apparently splint coal.
- No. 2853—Coal. (Sample 4.) Four-mile creek of Cumber-land river. Collected by R. C. B. Thruston, July 1st, 1887. The bed, 745 feet above drainage, has two seams of coal, severally 17 and 21 inches thick, separated by a 1-inch shale parting.

Apparently a weathered sample of what seems to be a splint or semi-cannel coal.

No. 2854—Coal. (Sample 9.) On Shade Branch of Bennett's Fork of Yellow creek, Bell county. Geological position:

Hignite seam, 775 feet above drainage. Collected by R. C. B. Thruston, July 15th, 1887.

Coal generally breaking irregularly, with irregular, shining surfaces. Some portions imperfectly laminated. No pyrites apparent, and but very little fibrous coal.

No. 2855—Coal. (Sample 18.) Head of Steward's Branch, one and a half miles from Pineville, Bell county. Geological position: top of lower coal measures. Collected by R. C. B. Thruston, November 16th, 1887.

The bed contains four seams of coal, severally from 3 to 20½ inches thick, separated by two half-inch shale partings. This sample is from the upper two seams. The uppermost seam is so badly weathered as to resemble earth rather than coal.

No. 2856—Coal. (Sample 19.) Head of Steward's Branch, one and a half miles from Pineville, Bell county. Geological position: top of lower coal measures. Sample from the 10½ and 3-inch seams. Same bed as preceding sample. Collected by R. C. B. Thruston, November 16th, 1887.

COMPOSITION OF THESE BELL COUNTY COALS.

(Air Dried.)

Number in Report.	2860	2861	2862	2853	2864	2855	2856
Hygroscopic moisture Volatile combustible matters. Coke	2.00 34.60 63.40	4.40 29.50 66.10	4.60 28.80 66.60	7.60 28.94 63.46	1.44 32.56 66.00	6.70 30.10 63.20	0.80 29.76 69.94
Totals	100.00	100 00	100.00	100.00	100.00	100.00	100.00
Total volatile matters Fixed carbon in coke Ash	36.60 60.40 3.00	33.90 60 50 5 60	83.40 63.60 3.00	36.54 55.26 8.20	34.00 58.34 12.66	36.80 49.80 13.40	30.06 55.74 14.20
Totals	100 00	100 00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur Character of the coke Color of the ash	0.507 denso. light brown.	0 978 very dense. brown-grey.	. 0.506 pulvorulent. light brown.	0.603 pulverulent. light brown.	1.464 dense spongy. purplish-grey.	1,125 pulverulent. brownish-grey.	2.965 dense. light purplish- grey.

The so-called cannel coal, No. 2854, owes its dense structure to the large proportion of earthy matters which it contains, giving it a composition approximating that of bituminous shale. Nos. 2855 and 2856 also contain large proportions of ash material, and all three of these coals have quite large proportions of sulphur, more especially No. 2856.

The four other coals, viz: those from Four-mile creek, are remarkably pure, good coals, with ash percentages, severally, of from 3 to 8.20 per cent., and sulphur from 0.506 to 0.978 per cent. These four coals would probably yield good coke, notwithstanding samples Nos. 2852 and 2853 gave a pulverulent coke in our analysis; this fact being very probably owing to the weathered condition of the samples taken. The unweathered coal, from the interior of the bed, would very likely give a good, firm coke.

No. 2857—Coal. Three-quarters of a mile above the mouth of Kit Island Branch of Straight creek, six miles from Pineville. Bed 3 feet 9 inches thick. Geological position: Sandy Branch coal. Collected by R. C. B. Thruston, March 13th, 1888.

A jet-black, pure-looking coal, breaking generally irregularly, with irregular, shining surfaces. No pyrites, and but little fibrous coal apparent.

### COMPOSITION .- Air Dried.

		Per cent.			Pe	r cent.
Hygroscopic moisture Volatile combustible matters	:	1.40 $35.34$ Total volatile matters				36.74
Much inflated coke		63.26 Fixed carbon in the coke . Very light-brown ash	:	•	•	60.26 3.00
		100.00				100.00
Percentage of sulphur	• •	1.222			83	===

A remarkably good semi-cannel coal.

No 2857a—Limonite Iron Ore. (Sample No. 17.) On the north face of Pine Mountain, three and a half miles east of Pineville, Bell county. Oriskany ore. Shows about 6 to 8 inches. Collected by R. C. B. Thruston, November 15th, 1887.

GEOL. SUR.-8.

A dark-brown, irregularly cellular ore, mixed with some lighter brown ochreous ore.

### COMPOSITION .- Air Dried.

Iron peroxide .	-	2.0				- 20						12	02	 245			60.517 = 42.362 iron
Alumina	•	•	Ō			•				 •	•		100	 •	•	Ō	5.723
Lime and magne	281	8	Ca	rb	or	181	es										traces.
Phosphoric acid																	. 259
Siliceous residue																	
Water and loss					8			÷	٠							•	9.561
																1	100.000

### CLINTON COUNTY.

### Soils.

No. 2858—Virgin Soil. Hill-side limestone soil. Long's Vineyard, Cartwright's Post-office, Clinton county. Sample taken to the depth of 10 inches. Forest growth: walnut, white oak, sugar-tree, and poplar. On St. Louis carboniferous limestone formation. Collected by R. H. Loughridge.

Soil friable, of a light, dirty, brownish-grey color. All passed through the coarse sieve,\* except about 12.5 per cent. of small, irregular fragments of ferruginous sandstone or concretions. Its siliceous residue from digestion in acids all passed through the fine sieve,† except a very small proportion of fine white sand.

No. 2859—VIRGIN SUBSOIL. Hill-side limestone subsoil. Long's Orchard, Cartwright Post-office, Clinton county. Sample taken at the depth between 10 and 13 inches. Collected by R. H. Loughridge.

Slightly lighter-colored than the above surface soil; in pretty firm clods. All passed through the coarse sieve, except about 4 per cent. of irregular, small fragments of ferruginous sandstone or concretions. Its siliceous residue left only a small quantity of fine white sand on the fine sieve.

No. 2860-Soil. Mountain sandy soil. Sewell's Mountain,

<sup>\*</sup> The coarse sieve has about 64 meshes to the centimeter square.

<sup>†</sup> The fine sieve has about 1600 meshes to the centimeter square.

of Poplar Mountain range, near Cartwright's Post-office, Clinton county. Forest growth: chestnut, chestnut oak, black gum; some red and white oaks. Sample taken to the depth of 10 inches. Collected by R. H. Loughridge. Geological position: on lower coal measures.

Soil of a light buff-grey color; some friable clods. All passed through the coarse sieve, except one or two small fragments of ferruginous sandstone or concretions. Its siliceous residue all passed through the fine sieve, except about 8.3 per cent. of fine white sand.

No. 2861—VIRGIN SOIL, from one mile west of Cartwright's Post-office. Taken to the depth of 12 inches. Forest growth: chestnul and red oak. Said to be the poorest soil in Clinton county, excepting that of Poplar Mountain summit. Collected by R. H. Loughridge.

Soil of a light, dirty, buff-grey color; contains some friable clods. All passed through the coarse sieve, except about 4 per cent. of small fragments of ferruginous sandstone or concretions. Its siliceous residue left about 4.5 per cent. of fine white sand on the fine sieve.

No. 2862—Red Under-Clay, of the chestnut soil, from one mile west of Cartwright's Post-office, Clinton county. Sample taken from 18 to 24 inches below the surface. Collected by R. H. Loughridge.

Subsoil of a dull, orange-red color, in friable clods. When powdered, it is of a handsome orange tint. The coarse sieve removed from it about 8.9 per cent. of irregular small fragments of ferruginous sandstone. Its siliceous residue all passed through the fine sieve, except about 2.5 per cent. of fine white sand.

COMPOSITION OF THESE CLINTON COUNTY SOILS.
(Calculated Dried at 212° F.)

Numbers in Report.	2858	2859	2860	2861	2862
Organic and volatile mat'rs	3.511	2.998	2.985	2.618	5.483
Alumina	6.757	6.215	4.243	4.361	7.335 13.188
Lime carbonate	. 405	.305	.151	.051	trace.
Magnesia	.073	.055	.145	. 109	.240
Phosphoric acid (P2 O5) .	.146	.065	.097	.064	.267
Potash extracted by acids .	. 340	.537	.351	.247	.231
Soda extracted by acids .	.518	.247	.317	.256	.077
Water expelled at 400° F .	.846	.860	.514	.453	1.285
Sand and insoluble silicates.	87.366	88.706	90.784	91.928	71.047
Totals	99.962	99.988	99.587	100.087	99.153
Moisture expelled at 212° F.	1.368	1.625	0.705	0.700	2.625
Potash in siliceous residue .	. 394	.497	.417	.422	.171
Soda in siliceous residue .	not est.	.214	not est.	not est.	not est.
Fine white sand	small.	small.	8.300	4.532	2.567
Gravel and rock fragments.	12.500	4.269	2.361	4.030	9.940
Character of the soil	virgin.	v. subsoil.	sandy.	virgin.	under-clay.

With the exception of Nos. 2858 and 2862, phosphoric acid is deficient in these soils. more especially in Nos. 2859 and 2861. Carbonate of lime is quite deficient in Nos. 2861 and 2862. Organic and volatile matters are very small in proportion in all, especially in Nos. 2859, 2860, and 2861, and are only in small proportions in Nos. 2858 and 2862. In the latter soil, very probably, combined water of the iron oxide. driven off at red heat, increases the small sum. Alumina and iron oxide are in very large proportions in 2862, more especially the iron oxide, which gives to the soil its orange-red color. In 2860 and 2861 these ingredients are quite deficient. Potash was found in larger proportions in all than their gen eral character and large proportions of sand and insoluble silicates seemed to warrant. These insoluble materials, it is seen, are quite large in all but No. 2862, more especially in Nos. 2860 and 2861. Such sandy soils can only be made permanently productive by a judicious use of appropriate fertilizers.

### HARLAN COUNTY.

COALS.

No. 2863—Coal. Head of Luna's creek, about 1400 feet above the river at Big Stone Gap, Harlan county. Upper part of the bed 59 inches thick; lower part 15 inches; separated by a 1½-inch parting. Collected by A. R. Crandall, June 25th, 1887.

A pure-looking, pitch-black coal; generally irregularly laminated, with some little fibrous coal between the laminæ. Portions breaking irregularly, with shining, irregular surfaces. No pyrites apparent in the sample.

No. 2864—Cannel Coal. (Sample 6.) Eight miles from Mount Pleasant, at the head of Catteran's creek of Martin's Fork, Harlan county. Collected by R. C. B. Thruston, July, 1887. Sample from a 22-inch seam, in bed containing three seams, two of stone-coal, severally 18 and 6 inches thick, separated by a 2-inch shale parting; 120 feet above drainage.

A dull, grey-black cannel coal; irregularly laminated. No fibrous coal or pyrites apparent in sample.

No. 2865—Coal. (Sample No. 7.) At head of Catteran's creek of Martin's Fork, 155 feet above drainage; eight miles above Mount Pleasant, Harlan county. Collected by R. C. B. Thruston, July, 1887.

A bright, pitch-black coal; fracture irregularly cuboidal and irregular, with shining surfaces. No pyrites or fibrous coal apparent.

No. 2866—Coal. (Sample No. 8.) On John Bell Branch, head of Catteran's creek of Martin's Fork, nine miles from Mount Pleasant, Harlan county; 365 feet above drainage. Collected by R. C. B. Thruston, July, 1887, from two lower seams in a bed containing five several seams of coal, separated by shale partings.

Apparently a splint coal, somewhat weathered. Some little granular pyrites apparent in the sample.

No. 2867—Coal. (Sample 11.) From near the head of Catteran's creek of Martin's Fork of Cumberland river, Harlan

county. Collected by R. C. B. Thruston, August 2d, 1887. Sample from three upper seams of a bed containing several seams, some of which are separated by shale partings. A very badly weathered sample.

- No. 2868—Coal. (Cannel?) (Sample 12.) From the lower two seams of the same coal bed from which the next preceding sample was taken. Seams severally 12 and 16 inches thick, separated by a 9-inch shale parting. Collected by R. C. B. Thruston, August 8th, 1887. Resembles a cannel coal.
- No. 2869—Coal. (Sample 13.) On Trace Branch of Clover creek of Cumberland river, Harlan county. Collected by R. C. B. Thruston, September 2d, 1887.

  Sample mostly in a much crumbled condition.
- No. 2870—Coal. (Sample 14.) On Slick Rock Branch of Clover Fork of Cumberland river. Collected by R. C. B. Thruston, September 3d, 1887. Sample from the three lower seams of a bed containing four or five seams, separated by thin shale partings.

  Sample much crumbled and somewhat weathered.
- No. 2871—Coal. (Sample 21.) From near the mouth of Terry's Branch of Wallen's creek, Harlan county. Geological position: near the base of the lower coal measures. Collected by R. C. B. Thruston, November 17th, 1887. Received at laboratory January 26th, 1888.

A pure-looking, jet-black coal, breaking generally irregularly, with irregular, shining surfaces. No pyrites or fibrous coal apparent in the sample. Sample taken from the lowest seam, 23 inches thick, from a bed containing three different seams of coal.

No. 2872—Coal. (Cannel?) (Sample 20.) From two upper seams of the same coal bed as the next preceding. Seams severally 8 and 13 inches thick.

A dull-black coal, breaking imperfectly in laminæ. Fracture sometimes very flat conchoidal. No pyrites or fibrous coal apparent in the sample.

COMPOSITION OF THESE HARLAN COUNTY COALS.

(Air Dried.)

Number in Report.	2863	2864	2865	2866	2867	2868	2869	2870	2871	2872
Hygroscopic moisture Volatile combustible matters	31.40 65.60	1.66 42.80 55.54	1.86 30.74 67.40	1.74 33.32 64.94	2.50 31.90 66.60	0.80 87.30 61.90	1.40 82.00 66.60	3.60 29 40 67.00	0.90 34.30 64.80	0.90 30.10 69.00
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100 00
Total volatile matters	34.40 56.30 9.30	44.46 35.44 20.10	32.60 65.00 2.40	35.06 59.24 5.70	34.40 61.80 3.80	38.10 54.90 7.00	33.40 62.30 4.30	88.00 57.00 10.00	35.20 62.50 2.30	31.00 42.40 26.60
Totals	100.00	100.00	100.00	100 00	100.001	100 00	100.00	100.00	100.00	100.00
Percentage of sulphur	0.604 dense. lt, buff- grey.	0.549 pulver- ulent. lt. grey- brown.	0.498 spongy. light brown.	1.034 spongy. light brown.	0.438 dense. light brown.	1.925 dense spongy. purplish brown.	0.712 dense spongy. light brown.	0 622 pulver- ulent. light grey.	0.577 inflated. lt. grey- brown.	1.084 dense friable. It. pur'h grey.

The cannel coals. Nos. 2864 and 2872, are diminished in value by their 20.10 and 26.60 percentages of ash; otherwise they may be usefully employed as fuel. No. 2868 resembles a cannel coal, but gave only 37.30 per cent. of volatile combustible matters, and left a dense coke. Its ash is only 7.00 per cent., which is small for cannel coals, but its sulphur is nearly 2 per cent.

Nos. 2865, 2867, 2869, 2866, and 2871 are very good coals, their ash percentages being severally only 2.40, 3.80, 4.30, 5.70, and 2.30.

The ash percentages of Nos. 2863, 2864, 2870, and 2872, are much larger, being severally 9.30, 20.10, 10.00, and 26.60 per cent.

The weathered samples of coals always show to a disadvantage in their analyses, owing to loss of their volatile combustible matters and the admixture of earthy materials.

### HARLAN COUNTY COALS.—(Continued.)

No. 2873—Coal. (Sample 17.) Head of Puckett's creek, Harlan county. Sample from two upper seams, 41 and 5 inches severally, separated by a 2-inch shale parting; in a bed which contains nine seams of coal, separated by partings of shale, varying from half an inch to 18 inches in thickness. Collected by R. C. B. Thruston, October, 1886. (This coal is 80 feet above coal sample 18, near the top of the 10 feet 10 inch bed.)

Apparently a good splint coal, showing but little fibrous coal, and no apparent pyrites.

No. 2874—COAL. (Sample 18.) Head of Puckett's creek, Harlan county. Sample 80 feet below sample 17. Coal 44 inches thick. Clay beneath. Collected by R. C. B. Thruston, October 12th, 1886.

Mostly splint coal, with some softer coal of irregular fracture. No apparent pyrites, and but little fibrous coal.

No. 2875—Coal. (Sample 19.) Near head of Puckett's creek, Harlan county. Coal 54 inches thick; 665 feet

below that of sample 18. Collected by R. C. B. Thruston, October 12th, 1886.

A pure-looking coal; mostly splint, with some softer coal of irregular fracture, with shining surfaces.

COMPOSITION OF THESE PUCKETT CREEK COALS .- (Air Dried.

Number in Report.	2873	2874	2875
Hygroscopic moisture	4.00	2.40 34.20	1.80 34 20
Coke	65.00	63.40	64.00
Totals	100.00	100.00	100.00
Total volatile matters	35.00 56.00 9.00	36.60 60.60 2.80	36.00 60.10 3.90
Totals	100.00	100.00	100.00
Percentage of sulphur	1.027	0.684	0.917
Character of the coke gr	friable. rey-brown.	dense spongy. grey-brown.	dense spongy. brownish-grey.

The coals Nos. 2874 and 2875, very much alike in composition, are remarkably good splint or semi-cannel coals, which would produce coke of the best quality. No. 2873 contains more ash material than these, but its large proportion of moisture indicates that the sample was from the weathered outcrop of the bed, and it is probable that the coal deeper in the bank, where it has not undergone partial decomposition under the atmospheric agencies, will be found to be much more pure.

No. 2876—COAL. (Sample 20.) Land of James Howard, on Martin's Fork of Cumberland river, at base of the Cumberland mountains, Harlan county. Sixty-eight inches of solid coal. Sample collected by R. C. B. Thruston, October 14th, 1886.

A somewhat weathered sample of splint coal.

No. 2877—Coal. (Sample 21.) James Howard's, on Black mountain, Martin's Fork, Harlan county; 250 feet above

drainage. Sample collected by R. C. B. Thruston, Cctober 14th, 1886.

A pretty pure-looking sample of splint coal. Some portions of irregular fracture, with shining surfaces. Very little fibrous coal, and no pyrites apparent.

No. 2878—COAL. (Sample 22.) On Broad Branch of Martin's Fork, Harlan county. Coal 43 inches thick. Sample collected by R. C. B. Thruston, October 14th, 1886.

A pure-looking, pitch-black coal, breaking readily with irregular shining surfaces. No fibrous coal or pyrites apparent.

No. 2879—Coal. (Sample 25.) On Crumbie's Branch of Martin's Fork of Cumberland river, Harlan county. Sample collected by R. C. B. Thruston, October 16th, 1886. Two seams of coal, 36 and 5 inches thick severally, separated by an 11-inch slate parting.

A pure-looking, firm, pitch-black splint or semi-cannel coal. But little fibrous coal, and no pyrites apparent.

COMPOSITION OF THESE MARTIN'S FORK COALS .- (Air Dried.)

Number in Report.	2876	2877	2878	2879
Hygroscopic moisture	2 00	1.10	1.20	1.20
Volatile combustible matters.	33.40	32.90	33.00	35.10
Coke	- 64 60	66 00	65.80	63.70
Totals	100.00	100.00	100.00	100.00
Total volatile matters	35.40	34.00	34.20	36 30
Fixed carbon in the coke	61.60	63 40	61.80	57.10
Ash	3.00	2 60	4.00	6 60
Totals	100.00	100.00	100.00	100.00
Percentage of sulphur Character of the coke	0.643 very dense.	0.519 very dense.	0.628 dense spongy	1 906 dense.
Color of the ash {	light salmon- colored.		grey-brown.	grey-purple.

Nos. 2876, 2877, and 2878 are remarkably good coals, and could be profitably used in the manufacture of the best quality of coke. No. 2879 contains rather more ash and sulphur, and less fixed carbon, yet is also a very good coal of the semi-cannel variety.

### HARLAN COUNTY COALS.—(Continued.)

No. 2880—COAL. (Sample 23.) On land of G. W. Smith, on Crank's creek, Harlan county. Coal 43 inches thick, with sandstone roof, and floor of clay 2 inches thick, resting on sandstone. Collected by R. C. B. Thruston, October 15th, 1886.

A somewhat weathered sample.

No. 2881—COAL. (Sample 24.) On Crank's creek, Harlan county. Forty-two inches of solid coal, 400 feet above that of sample 23. Collected by R. C. B. Thruston, October 15th, 1886.

A bright, pitch-black coal, breaking generally with irregular shining surfaces. But little fibrous coal, and no pyrites apparent.

COMPOSITION OF THESE CRANK'S CREEK COALS .- (Air Dried.)

ı	Nu	ml	be	r i	n	Re	po	rt	•										2880	2881
Hygroscopic moisture. Volatile combustible n																			3.20	2.80
Volatile combustible u	at	tei	rs		•														33.60	32 00
Coke		٠	٠	•	•	•		•	•	٠	•	•	÷	*	S.	•	٠	•	63.20	65,20
Totals	*	٠	•	٠		÷	٠	٠		٠	٠		٠	×	•		•		100,00	100.00
Cotal volatile matters	·	٠		٠							•								36.80	34.80
Fixed carbon in the co	ke	•	•			•	٠			٠	•	*	•	•				•	56.80	62 80
Ash	•		•		•	•	(%)	•	•	•	•	٠	•	•	•	•	•	•	6.40	2.40
Totals						٠					•		•	្ន		•	•		100.00	100.00
Percentage of sulphur													Ç						1.782	0.354
Percentage of sulphur Character of the coke													•						very dense.	dense.
Color of the ash		٠								•				•		•		1	light brown.	light brown.

No. 2881 is a remarkably good, pure coal, which would yield very good coke. No. 2880, a weathered sample, probably does not fairly represent the character of the unweathered coal deeper in the bed; yet it is better than the general average of coals in quality, leaving only 6.40 per cent. of ash. Its considerable proportion of sulphur might be objectionable when it is applied to some purposes.

### HARLAN COUNTY COALS.—(Continued.)

No. 2882—Coal. (Sample A.) Head of Luna's creek of Poor Fork of Cumberland river, Harlan county. Geological position: upper coal measures. Total thickness of seam, 6 feet 6 inches, but only 5 feet sampled. Collected by R. C. B. Thruston, December 12th, 1887.

Sample somewhat weathered, and partly in a powdered condition. Some portions dull-black.

- No. 2883—Coal. (Sample B.) Head of Camp Fork of Luna's creek, Harlan county. Upper coal measures. Collected by R. C. B. Thruston, December 13th, 1887.

  Sample somewhat weathered; partly pulverized.
- No. 2884—COAL. (Sample C.) Head of Rockhouse Branch of Luna's creek, Harlan county. Upper coal measures. Collected by R. C. B. Thruston, December 18th, 1887.

A pure-looking coal, generally. Some portions imperfectly laminated, with some little granular pyrites and fibrous coal apparent; other portions breaking irregularly, with shining, irregular surfaces.

No. 2885—Coal. (Sample D.) Near the mouth of Luna's creek of Poor Fork of Cumberland river, Harlan county. Collected by R. C. B. Thruston, December 14th, 1887.

A pure-looking coal, mostly imperfectly laminated. Some surfaces brilliant and irregular; others dull, showing a little fine granular pyrites.

No. 2886—COAL. (Sample E.) Elkhorn bed, near the mouth of Luna's creek, Harlan county. Thirty feet above the last sample, and 270 feet above the bed of the creek. Collected by R. C. B. Thruston, December 14th, 1887.

A pure-looking coal, breaking generally irregularly, with shining surfaces. Very little fibrous coal or pyrites apparent.

No. 2887—Coal. Near the mouth of Luna's creek of Poor Fork of Cumberland river, Harlan county. Geological position: 25 feet above sample E. Bed 48 inches thick. Collected by R. C. B. Thruston, December 14th, 1887. A much weathered sample.

No. 2888—Coal. (Sample G.) Cold Iron Branch of Poor Fork of Cumberland river, Harlan county. Lower coal measures. Bed 42½ inches. Collected by R. C. B. Thruston, December 18th, 1887.

A good-looking coal. Mosly lamellar in structure. Portions breaking irregularly, with irregular, shining surfaces. Some little granular pyrites and fibrous coal apparent in the sample.

No. 2889—Coal. (Sample H.) Cold Iron Branch of Poor Fork of Cumberland river, Harlan county. Elkhorn bed of lower coal measures, 25 feet above sample G. Bed 55½ inches thick. Collected by R. C. B. Thruston, December 15th, 1887.

Generally a fine-looking coal. Breaking with irregular, shining surfaces. Some portions laminated, and more dull in appearance, with some little fine granular pyrites apparent.

No. 2890—Coal. On Turkey-pen Branch, 15 miles, on Poor Fork, above Harlan Court-house, Harlan county. Lower coal measures. Collected by R. C. B. Thruston, December 22d, 1887.

COMPOSITION OF THESE HARLAN COUNTY COALS. (Air Dried.)

Number in the Report.	2887	2883	2884	2885	2886	7887	2888	2880	2890
Hygroscopic moisture Volatile combustible matters Coke	1.90 28.90 69.20	8.20 28.06 68.74	2.10 82.90 65.00	38.80 64.00	5.70 82.90 61.40	80.50 80.50 60.50	1.80 81.60 66.60	2.40 81.70 65.90	2.80 81.60 66.20
Totals	100.00	100.00	100.00	100 00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	30.80 61.10 8.10	31.26 60.44 8.80	35.00 56.60 8.40	36.00 61.00 3.00	88.60 58.80 60 60 60	39.80 50.12 10.08	33.40 68.20 8.40	34.10 63.70 2.20	88.80 61.60 4.60
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Percentage of sulphur	0.486 spongy. light grey- brown.	0.541 dense friable. light brown.	0.774 dense. lt. brown'h grey.	0.552 dense spongy. very light brown.	0.425 light spongy. very light brown.	0.560 pulveru- lent. nearly white.	1.181 light spongy. light brown.	0.812 dense spongy. light brown.	0.560 spongy. it. brown'h grey.

These are, generally, very good coking coals, yielding from 60.20 per cent. in No. 2887, up to 69.20 per cent. in No. 2882. Their proportions of ash range from 2.20 per cent. in No. 2889 up to 10.08 in No. 2887, and 8.10, 8.30, and 8.40, severally, in Nos. 2882, 2883, and 2884. But 2887 was a much weathered sample, and the other three had also been somewhat altered by exposure, and may not, therefore, fairly represent the character of the interior of the beds. The sulphur is quite small, except in Nos. 2888 and 2884.

No. 2891—Iron Ore. (Limonite.) From 27 miles above Harlan Court-house, on Poor Fork of Cumberland river, Harlan county. Position: near base of lower coal measures. Collected by R. C. B. Thruston, December 2d, 1887.

### COMPOSITION .- (Air Dried.)

Iron peroxide .									٠		. 1	44.200 = 39.940 iron.
Iron peroxide . Alumina											. !	9.080
Lime carbonate											. 1	20.000
Magnesia carbon												
Phosphoric acid	(F	,	O.	)	•	•					. 1	2.726 = 1.173 phosphorus.
Sulphur	`.	•		٠.							. 1	
Silica											•	12.600
Water and loss				٠							.	7.156
											-	100.000

### KNOX COUNTY.

No. 2892—Coal. Two and a half feet thick; on Golden Fork of Poplar creek, Knox county. Collected by A. R. Crandall.

### COMPOSITION .- (Air Dried.)

			Pe	r cent.
Total volatile matters				37.60
Fixed carbon in the coke Light grey ash		:	•	56.46 5.94
				100.00
1	Light grey ash	Total volatile matters	Total volatile matters	Total volatile matters

A good, pure coal.

### LAUREL COUNTY.

No. 2893—Coal. S. E. Owsley's, Laurel county. Sample brought by A. R. Crandall, August, 1886.

### COMPOSITION .- (Air Dried.)

52-26 N 1990		Per cent.
Hygroscopic moisture	$\begin{pmatrix} 3.40 \\ 33.80 \end{pmatrix}$ Total volatile matters	. 37.20
Dense spongy coke	62.80 Fixed carbon in the coke	. 58 80 . 4.00
	100.00	100.00
		===
Percentage of sulphur	. 0.930	

A very good coal, which would doubtless make good coke.

No. 2894—COAL from near Lily, Laurel county. Bed 46 inches thick. Sample collected December 25th, 1887, by Dr. L. H. Blanton.

A bright, pure-looking coal, breaking generally irregularly, with irregular, shining surfaces. No pyrites or fibrous coal apparent in the sample.

### COMPOSITION .- (Air Dried.)

Per cent.	Per cent.
Hygroscopic moisture 2.70 Tollow Volatile combustible matters 34.60	al volatile matters 37.30
Spongy coke 62.70 { Fin	ted carbon in the coke 59 70 th grey ash 3.00
100.00	100.00
Percentage of sulphur 1.923	

A good semi-cannel coal, giving but a small proportion of ash. Its somewhat large proportion of sulphur might be a drawback in the use of its coke in the iron manufacture.

### LIVINGSTON COUNTY.

No. 2895-HYDRAULIC LIMESTONE from Smithland, Livingston county. Sent from Frankfort for analysis, January 9th, 1888.

A compact, very dark-grey limestone.

### COMPOSITION .- (Air Dried.)

								Per cent.
Lime								46.788
Magnesia								.782
Magnesia Alumina and	iron	oxide .						12.170 with a trace of phosphoric acid
Potash							•	.057
Soda								.039
Silica								9.000 including 0.8 of soluble silica.
Carbonic acid	and	moisture	1	•	•			31.164
								100.000

This has the composition of some hydraulic limestones; but it does not contain as much silica, alumina, and iron oxide as the harder-setting cements, and may not be as durable as these. It will require much care in its calcination, because when too much burnt it loses its property of hardening under water. In an experiment made by the writer, the powdered stone having been submitted to a red heat for two hours, then mixed with water to a paste, and, after standing for a quarter of an hour, put under water, hardened after a day or so; but another portion, which had been heated for a longer time, did not harden much under water in twenty-four hours.

### MARION COUNTY.

### MINERAL WATERS.

No. 2896—MINERAL WATER, from the well of Mr. Charles H. Burns, in the town of Lebanon, Marion county. The same water was struck about 45 feet from the base of the black slate, in the mud sandstone, just above the Hudson group; also, in the gas well now being bored near the Burns well. The water is in large quantity. In the gas well the water tastes strongly of sulphur.

Sample, collected by Mr. W. T. Knott, was sent in quart bottles, well corked and sealed. It had no smell of sulphur when received, June 19th, 1887.

No. 2897—Water, from Harrison's spring, about 2 miles east of Lebanon, Marion county; in the ash-colored shales lying just above the black slate. Sample collected by W. T. Knott, June 17th, 1887.

GEOL. SUR -9.

No. 2898—Water, from Johnston's well, about 2 miles east of Lebanon. This water comes from the black slate formation near its base. Sample sent by Mr. W. T. Knott, June 17th, 1887.

### COMPOSITION OF THESE MARION COUNTY MINERAL WATERS.

(In 1000. parts of the waters.)

Numbers.	2896	2897	2898
Carbonic acid gas } Hydrogen sulphide gas . }	small quants. not est.	none.	small quantities; not esti- mated.
Lime carbonate	0.107 .020 .023 12.767 1.062 .857 	0.650 .170 .132 	small quantity; not estimated. small quantity; not estimated. small quantity; not est. 1.760

The water of No. 2896 is a good saline chalybeate, with but a small proportion of hydrogen sulphide; that of 2897, containing only about half as much total saline matters, has a larger proportion of magnesium sulphate (Epsom salt), and also of calcium and sodium sulphates. It (No. 2897) is a good weak saline water. No. 2898 is a very weak saline water, containing a little sodium sulphide and hydrogen sulphide, which may give it a weak sulphurous taste when fresh from the well.

### MUHLENBERG COUNTY.

COKE.

No. 2899—Coke, made of coal from three-quarters of a mile northwest of Mud River Coal Mines, Mud River, Muhlen-

berg county, Kentucky. Land of William Willis, Jr. Bed 44 inches thick. Samples sent from Frankfort, October 8th, 1887.

There were two samples in the box, made from the upper and lower part of the bed, of pure-looking coke. Average portions of the two were mixed and analyzed.

### COMPOSITION .- (Air Dried.)

Moisture and	olitatov	m	att	er	3 6	exi	pel	100	d	at	re	d	he	at		٠	٠		•						0.20
Carbon in the	coke .		•	•		•	•	•	•			- 3			•	• 3		. 1							90.20
Purplish-grey	ash .	•	٠	•	٠	•	٠	٠	•	٠	•	•			•		٠	3	•				٠	•	9,60
																									100.00
																									====
Percentage of	sulphur										ু	٠,		٥	ु	ু	ु	:			-	ુ	0	90	1.26

A good average coke.

No. 2900—Cannel Coal. On land of G. F. McNary, near the western border of Muhlenberg county. Bed 22 inches thick, overlying coal No. 11. Sample collected by Charles Eaves, Greenville, Muhlenberg county.

Coal generally dull black; imperfectly laminated. Some portions breaking with imperfect, broad conchoidal fracture. Some surfaces shining, and one part iridescent. No fibrous coal, and only a few bright scales of pyrites apparent.

### COMPOSITION .- (Air Dried.)

		Per cent.			cent.
Hygroscopic moisture Volatile combustible matters	•	$\binom{2.66}{36.74}$ Total volatile matters			39.40
Spongy coke		60.60 Carbon in the coke Dark purplish-brown ash	•	*	42.20 18.40
		100.00			100.00
Percentage of sulphur		7,526			

This coal contains too much sulphur and ash materials to be profitable in competition with better neighboring samples.

### PIKE COUNTY.

IRON ORE.

No. 2901—Limonite Iron Ore. Collected by C. M. Parsons, Pike county, Kentucky. Sent from Frankfort January 3d, 1888.

A pretty dense ore, dark brown in the denser portions; brownish-yellow in the more friable parts.

### COMPOSITION .- (Air Dried.)

											1 30		
Iron peroxide . Alumina Phosphoric acid								٠			Per cent. 66.37 =	46.46	iron.
Alumina											12.27		
Phosphoric acid	(I	,	0	()							3.04 ==	1.319	phosphorus.
Lime carbonate	`.	:			,						.20		•
Magnesia											a trace.		
Magnesia Siliceous residue											2.60		
Sulphur											a trace.		
Sulphur Water and loss		•									15.52		
											100.00		

A rich iron ore, containing but little sulphur, and more than average phosphorus.

### SIMPSON COUNTY.

No. 2902—Sulphur Water. Sample sent by the Board of Council of the town of Franklin, for quantitative analysis. Sample in two half-gallon bottles; received June 16th, 1887. (Analysis paid for by the Board of Council.)

### COMPOSITION .- (In 1000. parts of the water.)

Carbonic acid gas .		•	•				•	•	•	1	amount not estimated.
Hydrogen sulphide	gas									j	
Calcium carbonate .											0.1518) Held in solution in the water
Magnesium carbonat	æ										by carbonic acid.
Iron carbonate		•			•						.0169) by carbonic acid.
Sodium chloride											.1700
*Sodium sulphide .											.0930
Iron sulphide											
Potassium sulphate											
Calcium sulphate .											
Magnesium sulphate											1.3060
Silica											.0078
Boracic acid											a trace; not estimated.
Organic matters											a trace; not estimated.
Total saline mat	ter	s i	n	10	00	p	ar	ts	of	1	2.5127
the water										1	2.0121

A good saline sulphur water, resembling Blue Lick water, but containing much less sodium chloride (common salt) than that. It also has much more magnesium sulphate (Epsom salt) than that celebrated mineral water, making it more aperient in its action.

<sup>•</sup> Probably more of this salt would be found in the water analyzed fresh from the well.

### WHITLEY COUNTY.

### COALS.

No. 2903—COAL. Upper two feet of Bennett's coal. (Overlying 7 inches of fire-clay.) On Harp's creek, Whitley county. Sample collected by A. R. Crandall.

No. 2904—Coal. Lower 3 feet of Bennett's coal, &c. Collected by A. R. Crandall.

### COMPOSITION .— (Air Dried.)

]	Number in Report.														2903	2904				
Hygroscopic moisture Volatile combustible r Coke	nat	te:	rs	:	:				•							:	:	:	2.46 34 14 63.40	1.60 34.00 64.40
Totals																			100.00	100.00
Total volatile matters Fixed carbon in the co	oke	:	:	•						:				•	•			:	36.60 57.80 5.60	35.60 56.80 7.60
Totals				•				•			·								100.00	100.00
Percentage of sulphur Character of the coke Color of the ash	•	:	•	•			•	•	•	•					•	:	:	. {	0.848 spongy. brownish- grey.	0 820 spongy. light grey.

They are both good coals, of the splint variety. The upper bed gives rather the purer sample.

### WHITLEY COUNTY COALS.—(Continued.)

No. 2907—Coal, from the Procter Company's coal mine, Indian creek, Whitley county. Entry No. 7, Room No. 12. Average sample from the whole face of the bed. Upper bench 21 inches thick; lower bench, 22 inches. Collected by A. R. Crandall, March 12th, 1888.

A pure-looking, jet-black coal; breaking generally irregularly, with irregular, shining surfaces. Imperfectly laminated. Very little fibrous coal, and no pyrites apparent in the sample.

No. 2908—COAL, from same mine. Entry No. 5, Room No. 10. Average of the lower 22-inch seam. Collected by A. R. Crandall, same date as the above sample. Resembles the preceding.

No. 2909—COAL, from same mine. Entry No. 5. Average sample of the upper bench. Collected as above stated. Resembles the preceding samples.

### COMPOSITION OF THESE COALS .- (Air Dried.)

Number in Report.	2907	2908	2909
Hygroscopic moisture	2.00	2.60	1.80
	33.70	33.20	35.00
	64.30	64.20	63.20
Totals	100.00	100.00	100,00
Total volatile matters	35.70	35.80	36.80
	61.90	62.20	60.60
	2.40	2.00	2.60
Totals	100.00	100.00	100.00
0.1	0.796	0.769	0.714
	ight spongy.	light spongy.	light spongy.
	very light	light	very light
	rownish-grey.	grey-brown.	brownish-grey.

These are remarkably good semi-cannel coals, containing remarkably small proportions of ash, and not an inordinate percentage of sulphur. They would yield very good coke if proper means were used in the coking to prevent it from being too light and spongy.

### TENNESSEE COALS.

No. 2905—COAL. (Sample 10.) Head of Little Coal Branch of Tackett's creek of Clear Fork of Cumberland river, Claiborne county, Tennessee. Collected by R. C. B. Thruston, July 15th, 1887. Geological position: Dean's coal. Sample from lower seam, 40 inches thick, of the bed, which contains three seams of coal.

A pure-looking coal, breaking generally with shining, irregular surfaces. Imperfectly laminated. No pyrites or fibrous coal apparent in the sample.

No. 2906—Coal. (Sample 11.) Mouth of Big Sugar Branch of Bennetl's Fork of Yellow creek, Claiborne county, Tennessee. Sample from a 19-inch middle seam, in a bed containing two other seams of coal, 34 and 7 inches thick, severally. Collected by R. C. B. Thruston, July 15th, 1887.

COMPOSITION OF THESE TENNESSEE COALS .- (Atr Dried.)

Number in Report.	2905	2906
Hygroscopic moisture	2.10 34.24	7.00 30.90
Coke	63.66	62.10
Totals	100.00	100.00
Total volatile matters	36.34 55.26	37.90 <b>58</b> .30
Fixed carbon in the coke	8.40	3.80
Totals	100.00	100.00
Percentage of sulphur	1.427	0.685
Character of the coke	dense.	pulveru- lent.
	light pur- plish-grey.	lt. brown- ish-grey.

No. 2906 appears to be a weathered sample.

TABLE I.-COALS.-(Air Dried.)

	V
Remarks.	Browning's cannel coal, Cumb'land river.  Myers' coal, Lower Br. of Yellow creek. On Caney Branch of Yellow creek. I. R. Barner's, Yellow creek. Myer's coal, Yellow creek. Myer's coal, Yellow creek. Ali. McTee's coal, Yellow creek. Dean seam, lower 36 inches, Gréasy cr. Dean seam, lower 36 inches, Greek. W. D. King's coal, Yellow creek. Berry Turner's, Long's creek. Gouch & Co., John Little's, Br. Ky. river. Green Taulbee's. L. H. Noble's, Leatherwood Br., cannel coal. Goff & Co., John Little Br.; cannel coal. I. T. Garrard's, east side Goose creek. I. T. Garrard's, east side Goose creek. I. M. Jones', Beech creek, lower part. I. T. Smith's, Tom's Br.; cannel layer. J. T. Smith's, Tom's Br.; cannel layer. Alvis Hubbard's, Katy's creek. I. T. Smith's, Tom's Br.; cannel layer. J. T. Smith's, Tom's Br.; cannel layer. Alvis Hubbard's, Katy's creek. I. T. Smith's, Tom's Br.; cannel layer. J. T. Smith's, Tom's Br.; cannel layer. J. T. Smith's, Tom's Br.; cannel layer. Alvis Hubbard's, Katy's creek. I. T. Smith's, Tom's Br.; cannel layer. J. T. Smith's, Tom's Br.; cannel layer. J. T. Shith's, Tom's Groev Creek. I. Jenkins' Fickles' Cove of Looney creek. J. Jenkins' Fickles' Cove of Looney creek. J. Jenkins' Fickles' Cove of Looney creek. Garner's Hollow. J. L. Cornett's.
Specific gravity	1.298 1.298 1.298 1.298 1.393 1.393 1.390 1.300
Color of the Ash.	Grey-brown Light grey-brown Light grey Light grey Light grey Light grey Light brown Light brown Light brown Light brown Light brownsh-grey Brownsh-grey Brownsh-grey Brownsh-grey Light brownish-grey Brownish Brownish
Character of the coke.	Light spongy Spongy Light spongy Light spongy Light spongy Pulverulent Dense Pulverulent Dense Pulverulent Dense Light spongy
Sulphur	2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2
Ash.	できょう まょう ようちょう ひょう ひゅう はっちゃ まっちゃ まっちゃ まっちゃ かっちゃ まっちゃ かっちゃ まっちゃ かっちゃ まっちゃ かっちゃ かっちゃ かっちゃ ない
Fixed carbon in the coke	8 8 5 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8
Total volatile mat-	+ XXX
Coke.	244284848484444848448844488448848484848
Volatile combus- tible matters	+ 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
Hygroscopic moisture	\$ 8 8 8 8 5 5 5 8 8 8 8 8 8 8 8 8 8 8 8
County.	Bell Bell Bell Bell Bell Bell Bell Bell
Number	8 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2

Garner's Spring, 15 m. ab. m. Poor Fork. Head Island Br. Poor Fork Cumb. river. Head Island Br. Poor Fork Cumb. river. Geo. Turner's, Clover Fork Cumb. river. Silas Woodson Kelly's, Yocum's creek. Silas Woodson Kelly's, Yocum's creek. Bitum. shale, Sharpe's cr. of Vocum's creek. Gool sample 16, same bed land J. Farley. Wright Winn's, Rt-hand Fk. Yocum's cr. Coal sample 16, same bed land J. Farley. Wright Winn's, Rt-hand Fk. Yocum's cr. Child's creek of Clover Fork.  I. Creech's, Mud Lick Fk. Seagrave's cr. Child's creek of Clover Fork.  Buck Branch Collover Fork.  Buck Branch Child's cr.; sample upper at in. coal. Head Child's cr.; sample of splint layer. Nar head Child's cr.; samp. shop coal. Head of Laurel Fork of Seagrave's creek. Low Gap Br. Child's cr.; samp. shop coal. Head of Laurel Fork of Seagrave's creek. Low Gap Br. Child's cr.; samp. shop coal. Head of Laurel Fork of Seagrave's creek. Low Gap Br. Child's cr.; samp. shop coal. Head of Laurel Fork of Seagrave's creek. Low Gap Br. Child's cr.; samp. shop coal. Head Jones', mear mouth Martin's Fork. Seagrave cr. of Clover Fork Cumb. river. Seagrave cr. of Clover Fork Cumb. river. Fugit creek of Clover Fork Cumb. river. From same bed, bitum. or splint coal.  From same bed, bitum. or splint. Coa
1.342 1.363 1.263
Very light brown Light brown Light brown Light brown Light grey Very light grey Light Brown Light buff Light buff Light buff Light buff Light buff Light brown Salmon Crey-brown Salmon Grey-brown Crey-brown Light brownish-grey Purplish-grey Purplish-grey Purplish-grey Purplish-grey Purplish-grey Light brownish-grey Light brown Light brown Light brown Light brown Light brown Light buff Light buff Light brown Light brownish-grey
Pulverulent
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Harban Ha
### 1999   1999

TABLE I.—(Continued.)—COALS.—(Air Dried.)

Remarks.	Jerry Ledington's cannel coal, Beech Fk. W. McFadden's, Big cr., Red Bird creek. Richard Collins, Hal's Fork, Big creek. Asper's bank, Middle Fork Ky. river. Silas Nantz', Oldham Br. Middle Fork. Rthand Br. Rthand Fork Collier's cr. Wm. Lewis', Roland's Br. of Poor Fork. Rockcastle Min. & L. Co., R'ckcastle cr. John Field's coal. Cannel coal, Middle Fork Ky. river. Wm. Boling's, Rush Br. Middle Fork. Fish-trap Br. of North Fork of Ky. river. T'. J Johnson's, Ebersole Br. Middle Fk. John Spencer's, Grapevine creek. Mouth of Guy's creck. John Spencer's, Grapevine creek. Mouth of Guy's creck. Joseph Campbell's, mouth Rock Lick Br. Alex. Combs', North Fork of Ky. river. Samuel Whittaker's, Willard creek. Joseph Campbell's, mouth Rock Lick Br. Ferts Gross', near Squabble creek. Barren Fork, nut coal. Possum Branch of Wolf creek. Fight Fork of Patterson creek. W. Jones' coal, Briar creek. W. Jones' coal, Briar creek. Richardson's coal, Briar creek. Mahan's coel. Mahan Station.	W. M. Mahan's coal, Mahan Station. Caddell's coal, Wolf creek. Thomas' cannel coal.
Specific gravity	1.332 1.385 1.385 1.322 1.502 1.334 1.334 1.334 1.335 1.335 1.335 1.335 1.355	1.311
Color of the Ash.	Grey-brown Light Inde-grey Light Inde-grey Light Balmon Light Berling Light Brownish-grey Light Brownish-grey Light Inde-grey Light Inde-grey Light Inde-grey Light Inde-grey Nearly white Light purplish-grey Very light grey Light brownish-grey	Light purplish-grey . Dark greyish-purple. Grey-brown
Character of the coke.	Dense Spongy Light spongy Dense spongy Dense spongy Dense spongy Dense spongy Frable Spongy Frable Light spongy Friable Friable Friable Friable Friable Friable Friable Friable Light spongy	Spongy
Sulphur	667 667 667 667 667 667 667 667 667 667	2,000
Ash.	1 m + p で 4 p p 5 m p p p p p p p p p p p p p p p p	8 2 2
Fixed carbon in the coke	\$\frac{4}{2}\frac{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac{2}{2}\frac	
Total volatile mat -	**************************************	2 8 2 2
Coke.	¥\$\$\\\ \angle \a	
Volatile combus- tible matters	\$ \$28 \$ \$40 \$ \$28 \$28 \$28 \$28 \$28 \$28 \$28 \$28 \$28 \$	3.5.3 5.52 5.53
Hygroscopic moisture	5 6 6 8 8 6 6 6 6 8 8 8 8 8 8 8 8 8 8 8	7.50
County.	Leslie Leslie Leslie Leslie Leslie Leslie Lescher Letcher Letcher Letcher Letcher Perry Pe	Whitley Whitley Whitley
Number	98889999999999999999999999999999999999	

1.263 Lower part of Creedmore coal. 1.341 B. P. Shelby's coal, upper 26 inches. 1.293 Wagner Siler's, on Mud creek. not est. J. S. Berry's, Jellico seam.
ey .   1.263   I . :   1.341   E . :   1.293   V rey.   not est.   J
Light purplish-grey . Salmon Light brown Light brownish-grey.
1.060 Light spongy670 Light spongy629 Light spongy.
1.721
1. 54 1. 721 Spongy 3.34 1. 060 Light 1. 60 . 670 Light 2. 70 . 629 Light
59.36 62.76 60.76
39.10 37.90 35.70 36.82
62.58
37.40 35.66 33.12 34.82
5.4.8.8
Whitley Whitley Whitley Whitley
2813 2813 2815 WWW

# WEST VIRGINIA COALS.—(Air Dried.)

Remarks.	Three miles above mouth of Preacher cr. Quinnimont mines. Fire creek. Stone Cliff. Stone Cliff. Sewell, on top of the conglomerate. Hawknest.
Specific gravity.	not est.
Color of the Ash.	Chocolate brown Light brownish-grey. Light buff-grey
Character of the coke.	Dense friable.  Much inflated. Light spongy. Light spongy. Light spongy. Light spongy. Auch inflated. Spongy.
Sulphur	0 711 4418 464 629
Ash.	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4
Fixed carbon in the coke .	22.7.7.7.2 2.7.7.7.7.2 2.6.8.4.8.9.8
Total volatile matters	3 2 2 2 2 2 2 3 3 3 3 3 3 3 3 3 3 3 3 3
Coke.	88.88.88 87.88.88 88.88 88.88 88.88 88.88
Volatile com- bustible mat- ters	% 8 8 8 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
Hygroscopic moisture	8 5 8 8 8 8
County.	Wise
Number	2822 2822 2822 2822 2822 2823 2823 2823

### APPENDIX.

TABLE I.-COMPOSITION OF COALS.-(Air Dried.)-Continued.

Translation and the second sec	Character of the coke.	6.
1.20 33.40 65.50 33.40 65.70 65.70 6	8.8	e ey.
2. 6	20.5	
5.40	2 2	: :
5.40 33.90 65.30 34.70 59.90 47.70 1.20 33.30 65.30 34.80 65.30 34	40.2	: :
5.40 3.00 47.40 35.40 47.40 37.30 65.30 49.80 47.70 1.20 37.30 65.30 34.80 35.30 36.	2.40	
47.40 47.40 47.40 47.40 47.40 47.40 47.70	.50	
2. 20 34.90 65.30 34.00 65.30 3.70 65.30 3.70 65.30 3.70 65.30 34.00 65.30 34.	5. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3.	
1.20 34.90 35.30 65.30 36.10 65.20 37.00 65.30 36.10 65.20 37.00 65.30 36.10 65.20 37.00 65.30 37.00 65.30 37.00 65.30 37.00 65.30 37.00 67.30 37.00 67.30 37.00 67.30 37.00 67.30 37.00 67.30 37.30 67.30 57.30 57.30 57.30 57.30	3 2	200
2.20 33.30 63.90 36.10 66.20 3.70 2.20 33.20 65.20 34.80 61.80 61.80 34.00 65.20 34.00 65.20 34.00 65.20 34.00 65.20 34.00 65.20 34.00 64.00 36.00 61.80 61.80 4.20 34.00 64.00 36.00 64.00 36.00 64.00 36.00 64.00 36.00 64.00 36.00 64.00 36.00 66.00 34.00 66.00 34.00 66.00 34.00 66.00 34.00 66.00 34.00 66.00 34.00 66.00 34.00 66.00 34.00 66.00 34.00 34.00 66.00 34.00 36.00 34.00 36.00 34.00 66.00 36.00 34.00 66.00 36.0	1.80	900
2.60 33.30 64.00 35.80 61.80 4.50 35.80 64.60 35.80 64.60 35.80 64.60 35.80 64.60 35.80 64.60 35.80 64.60 35.80 65.10 35.80 65.80 35.80 65.80 35.80 65.80 35.80 65.80 35.80 65.80 35.80 65.80 35.80 65.80 35.80 65	3.70	187
2.66 33.70 64.20 35.80 59.66 4.66 33.70 64.20 35.80 59.60 61.80 2.20 34.60 64.20 35.80 60.40 3.00 61.80 28.90 65.10 33.90 65.40 3.00 65.10 33.90 65.50 55.60 33.00 65.70 33.90 65.50 33.90 65.70 33.90 65.70 35.31 12.66 65.20 34.40 35.34 65.20 34.40 35.34 65.20 34.40 35.34 65.20 33.30 65.20 33.30 65.20 33.30 65.20 33.30 65.50 33.30 65.50 33.30 65.50 33.30 65.50 65.20 65.20 65.20 65.20 33.30 65.50 33.30 65.50 65.20	80.4	1
1.40 34.00 04.00 30.00 01.80 2.20 34.00 03.40 05.10 33.90 06.50 28.90 06.50 28.90 06.50 28.90 06.50 28.90 06.50 28.90 06.50 28.90 06.50 28.90 28	9:	957
34.00 33.00 33.00 33.00 33.00 34.00 33.00 33.00 34.00 33.00 34.00	80 2.20	100
4.66 28.89 66.69 33.49 65.76 28.94 65.46 36.54 35.46 36.50 33.40 65.70 30.10 65.80 34.00 55.34 12.66 50.36 34.00 55.34 12.66 35.34 13.40 55.50 34.40 56.36 35.34 13.40 55.60 34.40 56.36 35.44 50.36 35.34 20.10 55.50 34.40 56.36 35.44 20.10 55.50 34.40 56.36 35.44 20.10 55.50 34.40 35.44 20.10 55.50 35.34 55.50 55.50 35.40 35.60 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 35.30 55.50 55.50 35.30 55.50 55.50 35.30 55.50 5	<b>\$</b> 9	1
7.60 28.94 63.46 36.54 55.26 8.20 1.44 32.56 66.00 34.00 53.34 12.66 67.0 30.10 69.94 30.06 55.34 12.66 1.40 35.44 60.26 36.70 55.30 13.40 1.40 35.44 60.26 36.70 55.30 9.30 1.66 42.80 55.54 44.46 35.44 20.10 1.74 33.32 64.94 35.06 55.00 2.40 2.50 31.90 65.60 34.40 61.80 3.80 2.50 31.90 65.60 34.40 61.80 3.80	9.50	1000
1.44 32.50 00.00 34.00 55.34 12.00 0.30 39.70 0.30 39.70 0.30 30.10 0.30 39.70 0.30 39.70 0.30 31.40 05.60 34.40 56.30 9.30 13.40 17.66 42.80 55.54 44.46 35.44 20.10 17.86 30.74 67.40 32.60 65.00 2.40 17.8 33.32 64.4 35.00 55.90 2.40 17.8 33.32 64.4 35.00 55.90 57.00 38.10 55.90 57.00 38.10 55.90 7.00	8.30	1027
3.00 31.40 65.60 34.40 55.74 14.20 3.00 35.74 60.26 3.00 31.40 65.60 34.40 55.70 3.00 3.00 3.00 3.00 3.00 3.00 3.00 3	÷.0	
3.00 31.40 65.60 36.74 66.26 3.00 31.40 65.60 34.40 56.30 9.30 1.66 42.80 35.54 44.46 35.44 20.10 1.74 33.32 64.94 35.06 55.00 2.40 2.50 31.90 65.60 34.40 61.80 3.80 37.30 61.90 38.10 54.90 7.00	74 14.30	
3.00 31.40 05.00 34.40 56.30 9.30 1.66 42.86 35.54 44.46 35.44 20.10 1.74 33.32 64.94 35.06 55.00 2.40 1.74 33.32 64.94 35.06 55.00 2.40 1.74 33.32 64.90 33.00 54.90 7.00 1.70 1.70 1.70 1.70 1.70 1.70 1.7	3.00	
1.86 30.74 67.40 32.66 65.00 2.40 1.74 33.32 64.40 55.06 34.40 61.80 3.80 3.73 61.90 38.10 54.90 7.00	2 3	
2.50 31.90 65.60 34.40 61.80 3.80 3.80 37.30 61.90 38.10 54.90 7.00	2.50	
2.50 37.30 65.60 34.40 61.80 3.80 3.80 3.70 51.90 7.00	24 5.70	And the second
37.30 61.90 38.10 54.90 7.00	3.80	
	90.7	
1.40 32.00 00.00 33.40 02.30 4.30	30 4.30	
3.00 30.40 07.00 33.00 57.00 10.00	8 8	

Terry's Branch of Wallen's creek. Head of Puckett's creek. Head of Puckett's creek. Head of Puckett's creek. Head of Puckett's creek. Martin's Fork of Cumberland river. Back Mountain, Martin's Fork, C'umb'l'd river. Back Mountain, Martin's Fork, C'umb'l'd river. Broad Branch of Martin's Fork. On Crank's creek, Poor Fork. On Cank's creek, Poor Fork. Head of Luna's creek. Head of Luna's creek. Head of Luna's creek. Near mouth of Luna's creek. Near mouth of Luna's creek. Old Iron Branch of Poor Fork, Cumb'land river. Cold Iron Branch, Poor Fork, Cumb'land river. Cold Iron Branch, Poor Fork. Golden Fork of Poplar creek. Golden Fork of Poplar creek. S. W. Owsley's coal. Near Siley; bed 46 inches thick. Cannel coal (?) near western border of county. On Harp's creek: Bennett's coal; lower 3 feet. On Harp's creek: Bennett's coal; lower 3 feet. Procter Company coal mine, whole bed. Procter Company coal mine, whole bed. Procter Company coal mine, upper bench.
1.084 1.087 1.096 1.782 1.782 1.782 1.782 1.930 1.930 1.933 1.943
Light purplish-grey . Grey-brown . Brownish-grey . Light salmon-colored Grey-brown . Light brown . Light grey .
Priable  Dense spongy Dense spongy Very dense Very dense Dense spongy Dense Spongy Dense Sponse Light spongy Light spongy Dense spongy Light spongy Light spongy Light spongy Dense spongy Light spongy Spongy Spongy Light spongy Light spongy Light spongy Light spongy Light spongy
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88 6 8 8 5 8 8 8 8 8 8 8 5 5 6 6 6 6 6 6
Harlan Whitley Whitley Whitley Whitley Whitley
2874 22874 22874 22875 22875 22875 22881 22881 22884 22886 22886 22886 22886 22886 22886 22887 2

## TENNESSEE COALS.-(Air Dried.)

Remerks.	.685 Head of Little Coal Branch of Tackett's creek.		
Sulphur			
Color of the Ash.	Light purplish-grey . Light brownish-grey .		
Character of the coke.	8.40 Dense		
Ash.	!		
Carbon in the coke	58.30		
Total volatile matters	36.34		
Coke	Coke.		
Volatile com- bustible mat- ters			
Hygroscopic 28			
County.	Claiborne		
Number	3905		

TABLE II.-COKES.-(Air Dried.)

Number	County.	doisture ex- pelled at red heat	Carbon in the coke	Ash.	Sulphur	Color of the Ash.	Owner of the Coal.	fours of cok-	Remarks.
1	Ja.	8.0	8.8	5.30	1.48	Dark brown	F. Barnet	Made at	Birmingham, Alabama.
3500 H	Mall	.30	95.80	8	1.710	Dark brown	F. Barnet	Made at	iont, West Virginia
	3ell	S.	5.50	8	.959	Dark brown		Made at	pont, West Virginia.
1000	Bell	8.8	87.30	8.8	1.343	Reddish-brown	las Barnett	42 Made at Dirmingnam,	nam, Alabama.
	3ell	9.	80.30	10.40	1.500	Greyish-brown		Made at	nont.
30	3ell	1.30	88.90	8.6	1.346	Brownish-grey		Made at	
-	Rell	2.40	85.30	12.30	1.543	Chocolate-brown		Made at	ham, Alabama.
4-10	Sell	9:	8.8	8 9	.547	Light grey-brown	Jas. Myers	Made at Quinnimont,	
1000		200	20.50	2	5.0	Brown		Made	ham, Alabama
	3ell	.8	8	8	100	Light brown	Alf. McTee	Made at	
	3ell	2.	93.70	8.6	.450	Brownish-grey	Alf. McTee	Made at	ham, Alabama.
2/9	Bell	9	8:8	8	629	Brownish-grey	Σ,	Made at	
	bell	9.3	5.5	8 6	1.033	Light brown	W. D. King	20	ham, Alabama.
	3ell	8 8	20.54	8.8	401	Brown-grey.		_	nont.
0.0	Sell	8.1	8.8	6.20	8	Light brown	M. Dorton	Made at B	наш
	3ell	8.	92.70	6.40	1.975	Nearly white	Dean seam	24.0	ont.
	Sell	1.80	90.50	2.70	89.	Light brown-grey	Dean seam	Made at C	
/2830	Harlan	8	93.10	6.30	.546	Brownish-grey	Wallen's creek	Made at Q	West
2.7	Harlan	1.34	80.76	17.90	9.	Light grey	Cornett	Made at Q	West
	Harlan	0+.	93.00	8	1.008	Grey-brown	S. S. Branson	Made at C	Xes.
	Harlan	<u>%</u>	93.90	9.30	303	: : :	wallen s creek	Made at C	ž
	Knox	96.	95.90	3.30	240	:	Cond. Beart	٠ ا	ž X
	xoux	04.	93.70	2.90	.470	m-Rick	Sandy Dranen	wade at	wes.
	Vuox	. 50	00.06	S.	712	Drown	D. F. Ely	Made at O	Ž,
	who was a second	8.	8.3	0.40	240	Diowillali-grey	District Control	Made at	Į,
_	Jaurel	5.	00.00	0.0	510.	Drownish-grey	Wiley Land	×	nont, West Virginia.
_	etcher	.70	69.30	8	. 803	Light grey-brown	T C D	Made at	NO.
7	Whitey.	1.50	93.30	2.30	303	Drownish-grey	J. S. Derry S	72 Made at Cuinnimont.	nont, west Virginia.

1.260 Purplish grey . . . . . Made from coal three-quarters of a mile northwest of Mud River coal mines.

9.6

8.30

0.30

Muhlenberg. . . . .

APPENDIX.

TABLE II.-(Continued.)-WEST VIRGINIA COKES.-(Air Dried.)

Remarks.	Made at Quinnimont, March 27-24, 1886. Made at Quinnimont, in Solendorf Coppe oven. Made at Quinnimont, March 26th, 1886. Made at Quinnimont, March 16th, 1886. Made at Quinnimont, March 16th, 1886. Made at Quinnimont, March 16th, 1886. Made at Quinnimont, March 27th, 1886. Made at Quinnimont, March 27th, 1886.
Hours of cok-	# : # # # # # 7 7 7
Kind of Coal.	Hawknest Fire creek Fire creek Fire creek Fire creek Sewell Sewell
Color of the Ash.	Brownish-grey Light grey-brown Brownish-grey Light buff-grey Light reddish-brown Light grey-brown Light grey-brown Light brownish-grey Light brown-grey
Sulphur	0.491 17. 104. 1437 1582 1586 178
Ash.	% 3 4 4 4 8 8 8
Carbon in the coke	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2
Moisture ex- pelled at red heat	6 6 6 8 8 8 8 8 8
County.	Fayette Fayette Fayette Fayette Fayette Fayette Fayette Fayette Fayette
Number	84 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4

_
Dried.
(Air
ORES.
VIRON
TABLE

Remarks.	Clinton group ore, Captain W. G. Allen's property. Clinton group ore, from the Purvis lands. Clinton group ore, from land of William Warren, Rose run.		42.362 On N. face of Pine Mt., 31/2 miles E. of Pineville. (6-8 in.) Oriskany ore. 39.940 On Poor Fork of Cumberland river, 27 miles above Harlan Court-house. 46.46 Collected by C. M. Parsons
Per cent. of iron	33.341 36.001 40.999		42.362 39.940 46.46
Carbonic acid, water, &c.	1.143 10.863 .693 11.283 1.607 8.445		30.00
Water expelled at 212° F.	1.693		9.56r 7.156 15.520
Silica	7.160	PPENDIX.	23.940
Phosphoric acid (P <sub>2</sub> O <sub>5</sub> )	1,102	APPER	3.040
Magnesia car- bonate	9 974		es. 4. 338 trace.
Lime carbon-	16.560 13.080 15.160		20.000 .300
Alumina	5.468		5.723 9.080 12.270
Iron peroxide.	47.630 \$1.430 \$8.570		60.517 44.200 66.370
County.	Bath Bath		Bell
Number	2575		2857 <i>a</i> 2891 2901

-	
7	
2120	
at	
Dried	
-(Calculated	
SUBSOILS	
AND	
STIOS-	
Ξ	
TABLE	

Remarks.	Crawfishy soil, on Mayfield creek. Virgin Obion river bottom soil. Virgin Obion river bottom soil. Virgin obion river bottom soil. Virgin soil, Flatwood loam. Subsoil of next preceding. Virgin soil, Flatwood loam. Subsoil of next preceding. Virgin soil, dark loam. Subsoil of next preceding. Virgin dark bluff, loam soil. Subsoil of the next preceding. Virgin soil, sist bottom, Tennessee river. Virgin soil, sist bottom, Tennessee river. Virgin soil, next preceding. Virgin soil, paland. Subsoil of next preceding. Virgin soil, paland. Subsoil of next preceding. Virgin soil, post Oak Flatwoods. Subsoil of next preceding. Virgin soil, Post Oak Flatwoods. Subsoil of next preceding. Virgin soil, Post Oak Flatwoods. Subsoil of next preceding. Virgin soil, Black Jack Oak Barren. Subsoil of next preceding. Virgin soil, Dost Oak Flats, Clark's river. Virgin soil, Dost Oak Flats, Clark's river. Virgin soil, Dost Oak Flats, Clark's river. Virgin soil, Boat Oak Flats, Clark's river. Virgin soil, Boat Oak Flats, Clark's river. Virgin soil, Ioam soil, low ridge, Tenn. river. Virgin soil, Ratwood, on Harvey road. Subsoil of next preceding.
Fine sand	16.500 n. e. 44.650 44.650 3.150 3.150 3.150 1.800 1.800 1.800 1.800 1.800
Soda in sili- cious residue.	0.834 1.243 1.203
Potash in sili- cious residue.	24
Hygroscopic moisture	6 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
Sand and in- soluble sili- cates	888 88 88 88 88 88 88 88 88 88 88 88 88
Water expelled at 380-400° F	\$2.50 \$2.50
Soda	\$2.00 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
Potash	7. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.
Phosphoric acid (P <sub>2</sub> O <sub>5</sub> ).	8 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2
Magnesia	8 4 4 9 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6
Lime carbon-	1. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2.
Iron peroxide .	2 4 2 2 2 4 2 4 2 4 2 4 2 4 2 4 2 4 2 4
Alumina and manganese oxide	2.4. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4.
Organic and volatile mat-	\$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$
County.	Ballard Calloway Call
Number	\$555 \$555 \$555 \$555 \$555 \$555 \$555 \$55

ounty.	soil.	
Bottom soil, Clark's rivet. Oak and hickory flat lands. Subsoil of next preceding. Subsoil of next preceding. Subsoil of next preceding. Virgin soil of Red Oak Barr Subsoil of next preceding. Virgin soil, Post Oak Flatwo Subsoil of next preceding. Virgin, fine silty soil.	irgin soil, hill-side limestone soil irgin subsoil, hill-side limestone fountain sandy soil. Irgin, poor soil. ked under-clay.	
Sub		
n. c.	gravel 12.500 4.269 10.661 11.507	
2.50 2.50 2.50 2.50 2.50 2.50 2.50 2.50	9. 5. 5. 4 12. 9. 9. 9. 9. 9. 9. 9. 9. 9. 9. 9. 9. 9.	
1. 604 1. 213 1. 343 1. 671 1. 513 1. 252 1. 428 1. 446 1. 566	394 +497 +422 +711	
1.300 2.750 2.825 1.265 1.1655 1.500 1.500 1.500 1.500	1.368 1.025 2.705 2.025 2.025	
89.757 84.733 84.733 90.050 91.147 88.33 87.213 87.213 92.726	DIX. 87.366 88.706 90.784 91.928	
85. 95. 95. 95. 95. 95. 95. 95. 95. 95. 9	APPENDIX. 846   87.36   88.70   514   90.78   453   91.92   11.285   71.04	
. 11.9 . 12.9 . 13.8 . 14.8 . 14.5	NI 24.6. 57.5. 57.5. 57.5. 57.5.	
172 271 227 329 190 100 179 179	351	-
\$6.50 \$6.00	645 645 765 765	
186 137 137 161 161 161	. 255 255 255 255 255	
44. 54. 54. 54. 54. 54. 54. 54. 54. 54.	.405 .305 .151 .051 trace.	
1.764 1.681 1.880 1.880 1.880 3.086 3.441 1.533	.757 .215 .243 .361	
8 4 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	7.335	
44644444444444444444444444444444444444	5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.5.	
Marshall McCracken McCracken. McCracken. McCracken. McCracken. McCracken. McCracken. McCracken.	Clinton Clinto	The state of the s
GEOL. SUR	-10	

TABLE IV.-CLAYS.-(Air Dried.)

Remarks.	Sandy, tertlary clay; near Blandville, one-half mile northwest Clay, a miles east of Blandville. Clay, a miles east of Mayfield creek, 4 miles northeast of Millburn. Clay, a miles north of Wickliffe. Yellow ochreous clay, Wickliffe. Fire-clay, Wickliffe.	Decomposed chert, lower carboniferous siliceous group.  Tertiary, northwestern part of Calloway county.  White pipe-clay, below Quaternary gravel.  Tertiary, black pyritous, plastic clay. (Below 2640.)  Black plastic deriver, Cartiary.  Dark clay; Mahan's, 6 miles east of Murray.  Dark clay; Mahan's, 6 miles east of Murray.  Decomposing chert, Quaternary.  LaGrange Tertiary; Rufus Morris.  LaGrange Tertiary; a miles north of Lynnville.  Whitish pipe-clay, south of Quill Hill, Tertiary.  Boaz Station.  Clay from Howard's pottery, at Bell City.  Clay from Howard's pottery, at Bell City.  Gray's place, near Scale.  Gray's place, near Scale.  Banks of Tennessee river, at Highland.	Pipe-clay, on J. G. Pugh's place.  Decomposing chert, 4 miles west of Birmingham.  Mr. Jones', 3 miles south of Paducah.  Mr. John Mitchell's, 3 miles east of Lovelaceville.  Gypseous clay; Mr. Hough's place, 4 miles west of Paducah.  Blue micaceous clay, on Armstrong's place, 7 miles east of Paducah.
<b>5</b> /4	Sandy, tertlary clay; near Bland Clay, a miles east of Blandville. Olay, West Fork of Mayfield trollay, a miles north of Wickliffe. Yellow ochreous clay, Wickliffe. Fire-clay, Wickliffe.	Decomposed chert, lower carboniferon Tertiary, northwestern part of Callow White pipe-clay, below Quaternary gretiary, black pyritous, plastic clay. Black joint clay, Lignitic Tertiary. From Russell's pottery, 6 miles east Dark clay; Mahan's, 6 miles east of Black plastic clay, Tertlary; Rufus Decomposing chert, Quaternary. LaGrange Tertiary, 3 miles north of Whitish pipe-clay, south of Quill Hill Boaz Station. Clay from Howard's pottery, at Bell Fire-clay, bulf above Columbus. Be Burradell's, 5 miles north of Benton. Gray's place, near Scale. From a cistern on old Winton place Banks of Tennessee river, at Highla	Pipe-clay, on J. G. Decomposing cher. Mr. Jones, 3 miles. Mr. John Mitchell Gypseous clay; M. Blue micaceous classiue micaceous classiue micaceous classiue micaceous classiue micaceous classiue micaceous classius
Fine sand	53 - 54 - 55 59 - 59 - 59 59 - 59	3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3. 3	n. e. n. e. n. e. Small pr. 50. 580
Water and loss.	5.126 6.194 8.322 6.623	13. 620 13. 620 14. 77 4 14. 77 4 16. 181 17. 17 4 18. 18. 18 18. 18. 18. 18 18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	5. 255 1. 858 1. 879 7. 581 5. 688
Soda	0.283 .291 .207 .307	2274 2274 2274 2276 2273 2273 2273 2273 2273 2273	172 286 470 075 472 473
Potash	1.293 .247 .926 .714 n. e.	888 999 995 1138 1139 1139 1139 1139 1139 1139 1139	1.954 1.107 1.340 1.453
Magnesia	0.209 .101 .331 .137 .138	286 136 136 136 136 136 136 136 137 137 137 137 137 137 137 137 137 137	396
Lime	o. 269 trace. trace. trace.	.045 .773 .101 .101 .213 .011 .011 .134 trace. trace. trace. trace. trace. trace.	1325 1.743 1.743
Iron peroxide .	20.350	888 1 1688 1 1980 1 198	888 888 7.73 848 848 848 848 848 848 848 848 848 84
Alumina	16.580 20.800 14.830 22.830 15.760	28. 38. 38. 98. 98. 98. 98. 98. 98. 98. 98. 98. 9	24.960 24.960 22.73.730 22.040 17.540 16.540
Silica	74.840 75.540 63.840 74.840 73.240	8 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	62.920 93.700 59.500 67.580 69.220
County.	Ballard	Calloway Carloway Carloway Carloway Carloway Carloway Carloway Marses Marshall Marshall Marshall	Marshall Marshall McCracken McCracken McCracken McCracken
Number	2569 1 2569 1 2570 1 2571 1	265 265 265 265 265 265 265 265 265 265	2775 2777 2777 2779 2780

# GENERAL REMARKS

# APPLYING TO APPENDIX.

(See Page 98 to 135.)

The publication of my Chemical Report, No. 7, N. S., having been delayed, owing to want of funds, the occasion was taken to supplement it, on pages 98 to 135, with the report of the aralyses which had been made since Report No. 7 was forwarded to the Director of the Survey.

The Appendix contains the analyses of

- 71 Coals, mainly from Bell, Harlan and Whitley counties, Kentucky, with seven from West Virginia and two from Claiborne county, Tennessee.
  - 5 Soils from Clinton county, Kentucky.
  - 4 Mineral waters.
  - 3 Iron ores.
  - 9 Cokes (eight from West Virginia, and one from Muhlenberg county, Kentucky)
  - 1 Limestone, believed to be hydraulic.

In all, 93 samples.

The samples of coals from Bell, Harlan and Whitley counties were mostly collected by Mr. R. C. Ballard Thruston, and proved, on analysis, to be, with very few exceptions, very good coals, of the "semi-cannel," "splint," or "block" coals, which compare in composition very favorably with the very best coals in use, and especially are very good coking coals, many of them being better than several coals which are extensively used in other States for the manufacture of good coke.

Some few of these samples were taken from the unopened, much-weathered outcrops of their several beds, and consequently do not represent some of the best qualities of the coal of these beds as it exists deeper in the bank—yielding, for example, a pulverulent coke, when, doubtless, the un-

weathered coal would give a spongy or cellular coke—and presenting more ash material than belongs to the unaltered coal of the interior of the bed.

For comparison with these Kentucky coking coals, I have appended, in the following table of average compositions, the composition of a Connellsville, Pennsylvania, coal, which is used at a large number of coke ovens, from a recent analysis, reported by Joseph D. Weeks—"MINERAL RESOURCES OF THE UNITED STATES," calendar year 1885, page 99—in the following table of comparison of our Kentucky coals.

Excluding from the list of Bell and Harlan coals given in this Appendix, all samples of cannel coal, and all which left more than 7 per cent. of ash, as being unfit for coking, the result is as follows:

AVERAGE COMPOSITION OF THE COKING COALS OF BELL AND HARLAN COUNTIES.—(Air Dried.)

Percentages of	Fixed carbon.	Volatile matters.	Ash.	Sulphur.
In 19 Bell county coals	60 81 61 00	35.35 33.06	3 45 3.82	0.779 .823
Averages of these 38 Kentucky coals.	60.90	34.20	3.64	. 801
Average of Connellsville coal	60.30	31.38	7 24	1.090

No doubt these coals would generally yield coke of very superior quality.

The highest proportions of sulphur in these fifty-two Bell and Harlan county coals was found in Nos. 2879 and 2880 of Harlan, and 2839 of Bell, being severally 1.906, 1.782, and 1.508 per cent.

Of the Bell and Harlan county coals, the compositions of which are given in the first part of Report, probably not more than one-half are good coking coals.

# ADDENDA.

Since the preceding report was placed in the hands of the printer the following analyses were made, which, by advice of Mr. Procter, Director of the Survey, are added, as "Addenda."

# Bell County—(Continued.)

No. 2910—Coal. Kettle Island Branch of Straight creek, Cumberland river, half a mile from the mouth of the creek. Land of Abe Lock. Geological position: 1st Coal? Average sample from the 46-inch face. Collected by A. R. Crandall, July 24, 1888.

A pure-looking, pitch-black coal, breaking, generally, imperfectly cuboidal and irregular, with shining irregular surfaces. Very little fibrous coal or pyrites apparent.

#### COMPOSITION .- (Air Dried.)

	Per cent.	1	Per cent.
Hygroscopic moisture	$\left\{\begin{array}{c} 0.20\\38\ 40\end{array}\right\}$ Total voiatile matters	v	. 38.60
Much inflated coke	61.40 { Fixed carbon in the coke Light brownish-grey ash	: :	. 57.80 3.60
	100.00		100.00
Percentage of sulphur	. 1.757		

A very good coal, containing rather more volatile combustible matters than the very best coking coal, which would make good coke, which would contain rather more than the best average of sulphur.

No. 2910a—Coal, in the approach to Cumberland Gap, Bell county. Geological position: inter-conglomerate. Bed varying in thickness from two feet four inches to two feet seven inches. Collected by John R. Procter, August, 1888.

A pure-looking semi-cannel or splint coal.

# COMPOSITION .— (Air Dried.)

**			Per cen	t.		P	er cent.
Volatile combustible matters		:	. 33.20	Total volatile matters			. 34.20
Spongy coke	•	٠	. 65.80	Fixed carbon in the coke Light reddish-brown ash	•	•	·63.60 • 2.20
			100.00				100 00

A remarkably pure, good coal, which would make very good coke.

# BRECKINRIDGE COUNTY.

No. 2911—Coal sent by A. Inglis, Breckinridge Cannel Coal Company, Cloverport, Kentucky, August 17, 1888.

# COMPOSITION .— (Air Dried.)

	Per cent.	Per cent.
Volatile combustible matters	$\begin{array}{c} 6.30 \\ 40.50 \end{array}$ Total combustible matters	46 80
Dense spongy coke	53.20 { Fixed carbon in the coke Light greyish-brown ash	47.88 5.32
	100 00	100.00
Percentage of sulphur	. 2.320	

A very good semi-cannel coal. It contains, however, rather large proportion of sulphur.

No. 2912—A, B, C and D. "BITUMINOUS SANDSTONES." (Sand saturated with petroleum.)

Sample A.—Ten miles south of Grayson Springs, Grayson county. Base of the conglomerate. Collected by Edward Orton, August, 1888.

Sample D.—From same locality. Collected by Edward Orton. Ten feet thick. Base of the conglomerate.

Sample B.—From near Elizabethtown, Hardin county. From base of the conglomerate. Collected by Moritz Fischer.

Sample C.—From land of John Richards, Carter county. Tough, pasty, black masses of sand penetrated by petro-

leum, some of which, extracted by gasoline and ether, appeared of the ordinary consistence of rather dense rock oil.

Proportions of petroleum as determined by combustion:

In sample A=9.40 per cent. In sample B=8.75 per cent. In sample C=8.50 per cent. In sample D=8.00 per cent.

## CALDWELL COUNTY.

No. 2913—Iron Ore, limonite, from "Stone Farm," two and a half miles north of Princeton, Caldwell county. Geological horizon: St. Louis. Sample collected by C. C. Genung.

A hard, generally dark brown, dense limonite, with very little ochreous ore. Only the iron, silica and phosphorus determined, in the air-dried ore, which were as follow:

Percentage of iron, =54.510. Percentage of silica, = 7 600. Percentage of phosphorus = .279.

A very good, rich iron ore.

#### CHRISTIAN COUNTY.

No. 2914—MINERAL WATER. From a bored well on the farm of J. S. McCarley, one and a half miles from Hopkinsville, on the L. & N. R. R. Called "Blue Lick Water" or "Green Lick." This water was struck at the depth of one hundred feet, and petroleum was found in small quantities in several of the strata passed through.

The water is said to be "perfectly clear" when first taken out of the well, but it soon becomes turbid, deposits a sediment of a dark color, and of a bituminous character.

It contains 1.07 per cent. of solid matters, mostly saline, with some sulphurous and nitrogenous organic matters; also much hydrogen sulphide gas, with a little carbonic acid gas.

Its saline contents are, namely: Lime and magnesia sulphates; sodium chloride (common salt); lime and magnesia carbonates, and traces of iron and alkaline carbonates.

## CRITTENDEN COUNTY.

No. 2915-MINERAL WATER, from "Crittenden County Sulphur Spring," four miles west of Marion.

This water contains a notable quantity of hydrogen sulphide gas, with some carbonic acid gas. One thousand grains of the water left only 0.44 of a grain of saline matters on evaporation at 212° F.

These consist mainly of sodium chloride and magnesium sulphate, with small traces of iron, lime, potash and lithium salts.

It is a pure but weak "white sulphur" water, which might, no doubt, be advantageously used as a medicinal agent.

# JEFFERSON COUNTY.

No. 2916—MINERAL WATER, from depth of 1,900 feet in the well, bored for natural gas, on the property of S. E. Edmunds, on the west side of Third street, between Weissinger and Magnolia streets, Louisville, named "St. Patrick's Well." The water is called "magnetic water."

The well flows two gallons per minute, out of very hard rock, as white as marble. Combustible gas flows out with the water, enough to supply several burners in the structure where the water is dispensed.

Qualitative analysis showed the presence of hydrogen sulphide, a trace of hydrogen carbide, and carbonic acid gases.

It contains, in solution, a considerable proportion of sodium chloride (common salt); also chlorides and sulphates of calcium and magnesium, notable traces of lithium and potassium salts, and some little of acid carbonates of lime and magnesia held in solution by the carbonic acid.

One hundred cubic centimeters of the water, evaporated to dryness at the temperature of 212°, left 1.829 grammes of these saline matters.

As the cork of the jug containing the water had been tampered with and broken in its transit from the well, the gases of the water had mainly escaped before it reached the laboratory. The magnetic properties of this water, like those of the once celebrated magnetic springs of Michigan, belong to the iron tubing, which is polarized by the earth's magnetism, and not to the water.

# KNOX COUNTY.

No. 2917—Coal, from near Gray's Station, North Jellico-Mines, Knox county. An average sample from the whole face of the 45-inch bed, taken 65 feet in the entry. Collected by A. R. Crandall, August 2d, 1888.

A bright, pure-looking, pitch-black, soft coal, breaking large cuboidal and irregular, with irregular, shining surfaces. Very little mineral charcoal or pyrites apparent.

## COMPOSITION .- (Air Dried.)

	Per cent.		P	er cent.
Hygroscopic moisture	. 1.20 Total volatile matters	  •		. 36.60 . 60.00 . 3.40
	100 00			100.00
D	1.049			

# Percentage of sulphur . . . . . . 1.043

# LAUREL COUNTY.

No. 2918—Coal, from near Lily. Bed six feet thick. Sample collected by Dr. L. H. Blanton, April, 1888.

A pure-looking, jet-black coal, imperfectly laminated, breaking, generally, with irregular shining surfaces. Bird's-eye structure in some pieces. A little fibrous coal apparent.

# COMPOSITION .- (Air Dried.)

Per cent.			er cent.
$\left\{\begin{array}{c} 2.40 \\ 35.50 \end{array}\right\}$ Total volatile matters			. 37.90
$62.10 \begin{cases} \text{Fixed carbon in the coke} \\ \text{Nearly white ash} \\ \dots \\ \dots \end{cases}$	:	:	. 3.80
100.00			100.00
	$2.40 \atop 35.50$ Total volatile matters 62.10 { Fixed carbon in the coke Nearly white ash	2.40 35.50 Total volatile matters	2.40 Total volatile matters

Percentage of sulphur . . . . . 0.961

A remarkably good and pure "semi-cannel" or "splint" coal, which would make very good coke.

#### LOGAN COUNTY.

No. 2919—MINERAL WATER, from a "sulphur well," on the premises of the "Logan County Female College," Russell-ville, Kentucky.

On qualitative analysis, this water was found to contain: Carbonic acid gas with a trace of hydrogen sulphide; small quantities of lime and magnesia acid carbonates; traces of potash and soda carbonates, and small proportions of sulphates and chlorides of these bases.

The total amount of saline matters in 100 cubic centimeters of this water, dried at 212° F., was only 0.044 gramme. It is quite a pure, soft, weak sulphur water.

No. 2919a—MINERAL WATER, from a well 163 feet deep, Geo. R. Beall's, near Russellville, Logan county. Geological position; Lower St. Louis.

By qualitative analysis, it was found to contain hydrogen sulphide and carbonic acid gases.

On evaporation it left of saline matters, dried at 212° F., 2.69 parts to the 1000. of the water. These were found to consist of chlorides of sodium, potassium, with a trace of lithium; the sodium chloride (common salt) being in largest proportion; also small proportions of magnesia and lime sulphates and carbonates.

It may be characterized as a weak white sulphur water.

## McLEAN COUNTY.

No. 2920—MINERAL WATER, from well of B. B. Shacklett, Sacramento, McLean county. Collected by Hon. R. S. Triplett.

Water slightly turbid; very slightly tinted yellowish. Has a somewhat fetid odor in the jug.

One litre of this water, evaporated to dryness at 212° F., left 5.624 grammes of saline matters=(0.562 per cent). This was slightly colored with organic matter, and when heated to redness left some charcoal from the organic matter.

Qualitative analysis showed that the saline ingredients were: Magnesia and lime carbonates with a trace of iron; magnesia sulphate (the largest ingredient); lime sulphate, and sodium and potassium chlorides.

The presence of organic matters would very probably render this water unsafe as a common beverage, or for medicinal purposes.

# MUHLENBERG COUNTY.

No. 2921—Coke, made at the Earlington ovens, Hopkins county, from Coal No. 11, not including 11 inches of the top. Pond river, Muhlenberg county. Coal 6 feet 9 inches thick. Sample collected by R. H. Arkenburgh, Jr. Received May 30, 1888.

No. 2922—Coke, made at Earlington ovens from Coal No. 9, on Pond river. Coal 5 feet 4 inches thick. Collected by R. H. Arkenburgh, Jr. Received May 30, 1888.

# COMPOSITION OF THESE COKES .- (Air Dried.)

											No. 2921.	No. 2922.
Hygroscopic moisture			_							-	0.30	= 0.20 per cent.
Volatile combustible matters										. !	.30	= .30 per cent
Carbon in the coke										.	92.60	85 80 per cent
Hygroscopic moisture Volatile combustible matters Carbon in the coke					•				•		6.80	13.70 per cent
											100.00	100.00 per cent
Percentage of sulphur	•	٠	•	·		ः	•	•	÷		1.896	1.647 per cent
Color of the ash	•			*	•		•	. 10	•	}	light grey- brown.	light grey- brown.

No. 2922 contains more than the usual average of ash material. The other coke is quite pure and good.

# OHIO COUNTY.

No. 2923—Coal. George Bettamy's mine, near South Panther creek, Ohio county. Bed three feet two inches thick. Sample collected by Moritz Fischer. Received June 27, 1888.

Some portions irregular cuboidal, breaking with shining irregular surfaces; other portions thin lamellated, with thin coatings of mineral charcoal and fine granular pyrites. Some little shining pyrites apparent.

No. 2924—Coal. Justin Simpson's mine, near Ætnaville, Ohio county, Kentucky. Bed three feet ten inches thick. Sample collected by Moritz Fischer. Resembles the preceding sample.

#### COMPOSITION OF THESE OHIO COUNTY COALS.

	Air l	DRIED.	CALCULATE 212	D DRIED AT
	No. 2923.	No. 2924.	No. 2923.	No. 2924.
Hygroscopic moisture Volatile combustible matters	7.60 37.80 54.60	10.00 31.40 58.60	40.91 59.09	34.89 65.11
Totals	100.00	100.00	100.00	100.00
Total volatile matters Fixed carbon in the coke Ash ,	45.40 45.00 9.69	41.40 53.10 5.50	48.70	59.00 6.11
Totals				
Character of the coke	spongy. light pur- plish grey.	dense spongy nearly white.		
Percentage of sulphur	2.334	0.934	2.527	1.040

No. 2925—Iron Ore. Land of G. W. Taylor, on west bank of Rough creek, two and a half miles north of Hartford, Ohio county. Sample collected by John R. Procter, August 26, 1888.

A nodular, dark-blue carbonate ore, superficially oxidated on some parts of the sample. Partially analyzed, the following results were obtained:

# AIR DRIED.

Ferric oxide (par	rtl	y	fr	on	a 1	the	e f	er	roi	18	ca	rb	on	at	e)				. 49.360 =	=34.55 iron
Alumina																		•	. 5.713	
Lime carbonate .	. 5								28										. 10.300	
Phosphoric acid											•								. 1.867	
Siliceous residue	(n	no	st	ly	8i	lic	a)											•	. 14.400	

Magnesia carbonate, carbonic acid, water, etc., were not esti-

No. 2926—Coal. Johnson coal, near Fordville, Ohio county. Bed three feet three inches thick. Sample collected by John R. Procter, August 26, 1888.

Breaks into thin laminæ, which are coated, more or less, with mineral charcoal, with some granular pyrites.

No. 2927—COAL. J. C. Gentry's, three and a half miles south of Fordsville, Ohio county. Collected by John R. Procter, August 25, 1888.

Breaks, generally, in thin laminæ, mostly with irregular, shining surfaces; some covered with fine mineral charcoal. No bright pyrites apparent.

# COMPOSITION OF THESE OHIO COUNTY COALS .- (Air Dried.)

														No 2926.	No. 2927.
Hygroscopic moisture Volatile combustible matters					្									6.00	6.10
Volatile combustible matters														35 20	40.30
Coke			•	•	•		•	٠				•		58.80	53.60
Totals	•	٠		ė	ř		•	٠	•					100.00	100.00
Total volatile matters Fixed carbon in the coke		٠							•		•			41.20 45.70	46.40
Ash	:	:	:	:	:		:	•	:	:	:	:		13 10	49.50
Totals		•	•	•		•		•	٠	•		•		100 00	100.00
Percentage of sulphur					•	•					•	•		1.812	2.115
Character of the coke			•	•	•		•	•			٠	•	}	dense spongy.	dense spongy.
Color of the ash	٠	•	•	•	•			•	•	•	•	•	}	nearly white.	light grey- ish-brown.

No. 2927 is a remarkably good semi-cannel or splint coal, which would make good coke, which, however, would contain a rather larger proportion of sulphur. No. 2926 contains a rather large proportion of ash material.

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# GEOLOGICAL SURVEY OF KENTUCKY,

JNO. R. PROCTER, DIRECTOR.

# CHEMICAL REPORT

OF THE

# COALS, COKES, MINERAL WATERS,

IRON ORES, LIMESTONES, ETC., ETC.,

# OF KENTUCKY.

BY ROBERT PETER, M. D., ETC.,

CHEMIST TO THE SURVEY.

THE EIGHTH CHEMICAL REPORT IN THE NEW SERIES, AND THE ELEVENTH SINCE THE BEGINNING OF THE SURVEY.

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# INTRODUCTORY LETTER.

CHEMICAL LABORATORY,
KENTUCKY GEOLOGICAL SURVEY,
LEXINGTON, KY., January 22d, 1890.

JOHN R. PROCTER,

Director of Geological Survey of Kentucky:

With pleasure I herewith transmit to you, for publication, my report of the chemical work done in this laboratory since my last report.

Very respectfully,

ROBERT PETER, M. D., ETC., Chemist to the Kentucky Geological Survey. This page in the original text is blank.

# CHEMICAL REPORT.

Of the 82 samples of Coals submitted to analysis, 9 were from Bell county; 9 from Jackson county; 6 from Laurel county; 9 from Lawrence county; 1 each from Martin and Owsley counties; 22 from Pike county; 7 from Pulaski county; 3 from Rockcastle county; 12 from Whitley county, and 3 from Tennessee.

The Pike county coals analyzed showed the greatest number with the *smallest average* proportions of *volatile combustible* matters; which ranged from 26.00 per cent. in No. 2999, up to 28.13 per cent. in No. 2994.

The *highest* proportions of volatile combustible matters in the *cannel coals*, range from 40.50 per cent. in No. 2969, of Lawrence county, up to 52.34 per cent. in No. 2928, of Bell county, and 54.80 per cent. in No. 2983 of Owsley county. The lowest proportion is 24.70 per cent. in No. 2962 from Laurel county.

The proportions of carbon in the cokes, from 26.66 per cent. in No. 2928, Bell county (a cannel coal with 20.40 per cent. of ash), range up to 60.00 per cent. in No. 2989, and as high as 65.80 per cent. in No. 3000; both Pike county coals.

Of the 22 samples of coal from Pike county analyzed, no less than ten contained fixed carbon in the coke, in proportions varying from 60.00 per cent. in No. 2987, up to 65.70 per cent. in No. 3002.

The highest percentages of carbon in the coke of the Bell county coals were 62.50, 62.80 and 63.60; in numbers, 2936, 2934 and 2935, severally.

The ash percentages range from 1.40 per cent, in No. 2934 of Bell county, and as low as 0.90 per cent in No. 2954, of Laurel county, up to 10.00 in No. 2970, of Lawrence county, and as high as 37.54 per cent. in No. 2962, of Laurel county.

There being eleven of the coals examined, several of which are cannel coals, the ash percentages range between 10.00 and 37.54 per cent. The remainder of the 82 coals analyzed, gave ash percentages ranging from 0.90 to 9.40 per cent.; 34 of the 82 having ash percentages, ranging from 0.90 to 6.00 per cent. only.

The percentages of *sulphur* range from 0.370 per cent., in No. 2928, of Bell county, up to as high as 6.450 and 4.714, in Nos. 2967 and 2960, of Lawrence county, and 4.670 per cent. in No. 2962, of Laurel county. These, however, are exceptional cases; in the whole 82 coals the sulphur percentages are in smaller proportions than 1 per cent.; ranging from 0.370 per cent. upward.

In regard to the coking quality of these coals, it is probable that samples Nos. 2931, 2932, 2933, 2934 and 2935, of *Bell county*, would make good coke. Their ash percentages being, severally, 6.40, 6.40, 5.70, 1.40, and 2.40. Of course, those which leave the smallest proportion of ash are the best.

Of the Jackson county coals, only Nos. 2949, 2954, 2955 and 2956, would, most probably, make good coke; containing, severally, 6.40, 8.20, 6.30, 3.10 and 4.64 per cent. of ash. An ash percentage in the coal of 8.20 would probably be too high for coke of best quality. The larger ash percentages in the other samples which gave "spongy coke," diminish proportionally their value as fuel.

The 6 Laurel county coals give three which may yield good cokes, viz: Nos. 2959, 2960 and 2961, the ash percentages of which are severally 0.90, 2.90 and 5.80. No. 2963 is a coking coal, but it leaves 7.60 per cent. of ash, which, of course, would make a larger percentage in the coke.

Of the 9 coals from Lawrence county, only three would probably make coke of the first quality, viz: Nos. 2965, 2971 (bis) and 2967, which give, severally, 4.20, 4.40 and 6.00 per cent. of ash. The other coking coals of this county reported, viz: Nos. 2964, 2966, 2969 and 2968, are reduced in value by their larger proportions of ash, severally amounting to 8.20, 14.00, 14.20 and 16.00 per cent.

Seven of the Pike county coals analyzed promise to yield good cokes, viz: Nos. 2986, 2988, 2989, 2991, 2992 2995 and

2998; the ash percentages of which coals are, severally, 6.40, 6.40, 6.00, 5.40, 6.20, 5.80 and 4.60.

The cokes of the other coking coals of this county, reported above, are diminished in value by their larger proportions of ash, and the large percentages of *sulphur* in coals Nos. 2986, 2091, and 2992, amounting to 1.703, 2.527 and 2.280 per cent., severally, might diminish the value, for some uses, of the cokes made from them.

The seven Pulaski county coals gave only three which promise good cokes, viz: Nos. 3009, 3012 and 3013, which leave severally 8.00, 9.40 and 5.60 percentages of ash. Only one of these, viz: No. 3013 would give coke of the best quality, the large ash of the other two diminishing their value.

The three Rockcastle samples promise only one which would give good coke, viz: No. 3018, the ash of which coal is only 4.40 per cent., and the sulphur only 0.494 per cent.

Among the twelve Whitley county coals analyzed, eight promise to give good cokes, viz; Nos. 3022, 3023, 3024, 3026, 3027, 3028, 3031 and 3032, which, severally, left only 3.70, 1.40, 1.30, 2.80, 3.40, 4.00, 4.80 and 5.14 per cents. of ashes. The percentages of *sulphur* in these coals being, severally, 0.453, 0.494, 0.714, 0.796, 0.961, 0.522, 1.428 and 0.659.

Whitley county appears to be especially favored in coking coals.

#### BELL COUNTY.

#### COALS.

No. 2928—Cannel Coal, from a mine opposite Pineville, belonging to the "Four-mile Coal and Land Company."

A dull black, exceedingly tough cannel coal or bituminous shale, imperfectly laminated, in some parts breaking with imperfect conchoidal fracture. No pyrites or fibrous coal apparent. Sample sent from Frankfort February 23, 1889.

No. 2929—Coal: on Caney Fork of Left Fork of Straight creek. A 42-inch coal without parting. Average sample collected by R. C. B. Thruston, August 20, 1888.

No. 2930—Coal: on right hand Fork of Straight creek, twoand-a-half miles above the mouth. Geological position,

- upper splint bed. Bed 60 inches thick. Average sample taken from the entire bed, except a 6-inch parting, by R. C. B. Thruston, August 20, 1888.
- No. 2931—COAL: average sample from the 48-inch coal in the bed of Straight creek, dipping 70 degrees, near the mouth of "Kettle Island Branch." Collected by R. C. B. Thruston, August 22, 1888.
- No. 2932—Coal: "Cockerell's coal," on Cannon creek, one mile above the railroad. Average sample of the 2 feet 8 inches face of the bed. "Lower Barner Seam." Collected by A. R. Crandall, May 26, 1889.
- No. 2933—Coal: "Cockerell's coal, Barner seam," on Cannon creek. Average sample from the 2 feet 10 inches face of the bed. Collected by A. R. Crandall, May, 1889.
- No. 2934—COAL: "Cockerell's coal," on Cannon creek. Average sample from the 37-inch face of the bed under the sandstone ledge, 500 feet above the creek. Collected by A. R. Crandall, May 31, 1889.
- No. 2935—Coal on Bennett's Fork, in Fork Ridge. Average sample of the 48-inch face, driven in 75 yards. Collected by A. R. Crandall, December 17, 1889.
- No. 2936—Coal on "Bennett's Fork," near Tennessee line, Mingo mountain. Average sample of the bed. Coal 64 inches, clay-parting 6 inches. Collected by A. R. Crandall, December 17, 1889.

SITION OF THE ABOVE DESCRIBED BELL COUNTY COALS .- (Air Dried.)

	9998	2929	2930	2931	2932	2933	2984	2985	2936
Number in Keport.							1		
Hyperoscopic moisture	0.60	1.20	2.20	2.06	1.00	1.20	33.80	1.20	3.46 30.24
Volatile combustible matters	52.34 47.06	65.80	63.10	62.40	62.50	62.50	64.20	99	66.80
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	52.94 26.66	35.00	36.90	37.60 56.00	37.50 56.10 6.40	37.50 56.80 5.70	35.80 62.80 1.40	34.00 63.60 2.40	33.70 62.50 3.80
Ash	20.40	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
		_!!		970	0 103	0 935	0.677	0.439	0.756
Percentage of sulphur	pulver-		Dense friable.	light spongy.	spongy.		90	light spongy.	
Color of the ash	grey-	light brown.	reddish brown.	dark lilac:	purplish grey.	grey.	brown'h.	reddish.	reddish.

These Bell county coals are all valuable; even the cannel coal, No. 2928, with its small proportion of carbon in its pulverulent coke—only 26.66 per cent.—and its large quantity of ash—20.40 per cent.—may be profitably used in the production of gas, or as fuel, because of its large proportion of volatile combustible matters, which equal 52.34 per cent. of its weight, and which, besides giving out more light in burning, gives very much more heat than an equal weight of fixed carbon. Its proportion of sulphur, also, is quite small.

The coals, numbers 2029, 2931, 2932, 2933, 2934 and 2935, all appear to be coking coals. The best and most valuable of these are Nos. 2929, 2934 and 2935, which yield, severally, 61.80, 62.80 and 63.60 per cent. of fixed carbon (stated in the order in which they are named), contained in 65.00, 64.20 and 66.00 per cent. of spongy cokes, and contain only 0.549, 0.577 and 0.439 per cent., severally, of sulphur.

The other coking coals described above, viz: numbers 2931, 2932 and 2933, containing, severally, 3.845, 2.193 and 2.335 per cent. of sulphur, would be less valuable, especially in the iron industry, than the others mentioned above, but yet would be very good fuel for most purposes; all leaving but a moderate proportion of ash in burning.

No. 2934 leaves but a remarkably small proportion of ash on burning, viz: only 1.40 per cent. The ashes of all these coals, except that of the cannel coal, are below the general average from good coals, varying from 1.40 per cent. in No. 2934, to 6. 40 per cent. in Nos. 2931 and 2932. The coals, Nos. 2930 and 2936, which yielded only *friable cokes* in the crucible, are quite pure and valuable coals for all ordinary purposes, and may, possibly, under the pressure of the coking-oven, yield good coke.

# BELL COUNTY COKE.

No. 2937—Coke. Seventy-two-hour coke, made from the run of the mine.

Sample from Pine Mountain Iron Company, Pineville, Bell county. Received April 24, 1889.

#### COMPOSITION.

Hygroscopic moisture 0.40 Volatile matters at red heat 30	Volatile matters 0.70
Coke	Carbon in the coke 94.90 Grey-brown ash 4.40
100.00	100 00
Percentages of sulphur 0.492	

Quite a good, pure coke.

# BELL COUNTY IRON ORES.

No. 2938—Oriskany Iron Ore, on Straight creek. Average sample from the 6½ feet exposure. Collected by R. C. B. Thruston, August 23, 1888.

A cellular limonite, with much ochreous incrustation.

No. 2939—"Lock's Iron Ore." Kettle Island Branch of Straight creek. Bed 3 feet 7½ inches thick. Average sample from the whole face of the bed collected by R. C. B. Thruston, August 22, 1888.

A limonite ore encrusted with ochreous material; some lumps with a little iron carbonate in the interior.

# COMPOSITION OF THESE BELL COUNTY IRON ORES .- (Air-Dried.)

						Nι	ım	be	rs										ī	2938		2939
			-		-		-		50	,	-					-	 	-	-!			
Iron peroxide			÷	়					Ŋ.		8		90				200	23		a 60.023		71.120
Alumina								٠								30		្ន	. !	5 148	1	3.680
Lime carbonate																			i	trace.		1.900
Magnesia																				.180	200	2.160
Phosphoric acid																			. 1	b .639	d	3.11
Sulphur																0.7			1	.412	1	1 09
Silica							- 60		3.0	2.0	0.4	•	*			000			. !	19.500	1	10 00
Water, etc., and	le	988						350	*		2			**	*	**		*		14.098	. }	16.93
																			-	100.00		

<sup>(</sup>a.) Equal to 42.164 per cent. of iron.

Rich iron ores-large phosphorus in No. 2939.

<sup>(</sup>c.) Equal to 49.784 per cent. of iron.

<sup>(</sup>b.) Equal to 0.279 per cent. phosphorus.

<sup>(</sup>d.) Equal to 1.340 per cent. phosphorus.

# BRECKINRIDGE COUNTY.

No. 2940—"BITUMINOUS SANDSTONE." Breaks of Rough river, Lost Run, 6 miles east of Hardinsburg. Bed 5 to 7 feet thick. Sample collected by Mr. J. C. Fawcett, January, 1889.

No. 2941—"BITUMINOUS SANDSTONE," collected by Mr. C. W. Moorman, of Cloverport, who states that it is 3 to 30 feet in thickness, says he has 700 acres of it. The sample is from its outcrop. Analyzed by incineration, etc.

COMPOSITION OF THESE BITUMINOUS SANDSTONES .- (Air Dried.)

				um															2940	2941
— Semiflu <b>F</b> ine sil	id bitumen iceous sand	(or	Petro	leur	n)											•			7.96 91.04	9.25
																			1.00	
				1 1 a a a a a a				_		-									Petroleum.	Fine sand
Sec Car	ter county	for	analy	sis	of	a	no	th	er	bi	itu	ım	in	ou	ıs s	ar	ıd,	-	3.56	96.44
A nother	omposition r from For	of v	rhich chene	is TT	tab	,	٠		•	٠	•	٠	•	•	٠	*	•	ſ	6.37	90.4

It appears, on analysis, that what seems to be a compact, and, sometimes, a tough sandstone resolves itself, on analysis, by burning, or by the action of bitumen solvents, into petroleum and fine sand, which, in some samples, is beautifully pure and white. Whether thick petroleum could be used with advantage on macadamized or sandy streets and roads, is well worthy of trial; it would, at least, prevent the penetration of water, which is so very injurious. If the experiment is made, the street or road should be perfectly dry and the petroleum should be heated.

#### CALDWELL COUNTY.

No. 2942—MARL. Somewhat indurated. From land of Mr. E. M. Stephens, 3 miles east of Princeton, Caldwell county, Ky. Bed 4 feet thick.

Mostly in soft, friable laminæ, of a light-grey color, nearly white, with some ferruginous stains. Quite plastic with water when pulverized. Fused before the blow-pipe. Did not prove to be "hydraulic" on experiment.

# COMPOSITION OF THIS MARL.—(Air Dried.)

								1111				521			2				2000		Per cent
Silica																	00400				37.800
Alumina, with trace	of i	LOI	1 0	X	de											6					13,321
Lime		16.65									2				50						21.168
Magnesia				Ŷ					÷												1.440
Phosphoric acid					•		70.0	-		14		23			:	23	2223				1.279
Potash										82			٠			1			·		2.391
Soda	70 83	77.2	23	٠		135			8			23		12	0			_	-	-	2.234
Soda	and	d i	oss	i.	Ŷ								•	٠	•				ì	:	20.367
																					100.000

The potash, phosphoric acid, lime, etc., contained in this marl, may recommend it as a top-dressing to a poor soil. It might probably be also used in some forms of terra cotta or cheap pottery.

CARTER COUNTY.

No. 2943—BITUMINOUS SAND, from head of Tygert's creek. A tough, black solid, which, when rubbed up in a mortar, became soft and somewhat plastic. Digested in gasoline, etc., it gave up, of thick petroleum, 3.56 per cent., leaving pure white sand, 96.44 per cent.

The sand appeared to be pure enough to be employed in the manufacture of the finest white glass.

# DAVIESS COUNTY.

No. 2944—"Cannel Shale," from Owensboro, Daviess county, sent by R. S. Bevier to Mr. Procter, October 1, 1888.

A dense black shale, with impressions of small bivalve shells and some bright pyrites.

# COMPOSITION .- (Air Dried.)

	Per cent.	392 • • CARREL (200/0020) - 22					r cent.
Hygroscopic moisture Volatile combustible matters	2.70	Total volatile matters		•			14.50
Volatile combustible matters :		Carbon in the residue					13.10
Pulverulent residue	. 85.50	Grey-brown residue .	•		•	•	72.40
	100.00						100.00
							-

Of very little value for fuel.

No. 2945—MINERAL WATER from Bell's farm, 12 miles south of Owensboro, on the line of the Owensboro and Falls of Rough Creek Railroad. Sample collected by Captain R. S. Triplett. Received July 15, 1889.

The water gave a slightly acid reaction, and had deposited a small amount of ferruginous sediment in the jug in which it was brought.

# COMPOSITION IN 1,000 PARTS.

Iron sulphate		25			*3		×			•0									•			200			0.031
Alumina Sulphate						*												0.00						. 1	.928
Lime suipnate																								. 1	. 507
Potash sulphate .					*						+		S 8		.0										.024
Soda sulphate													· 3											. 1	.019
Sodium chloride .																						•			.019
Magnesium chlorid	le									2			. 1		20				-	20	2	10	0.5		.057
Silica		•				•	•				•			•		٠	•	•							.028
Total saline m	at	ter	s i	n	1,0	000	p	ar	s	of	tl	he	wa	ite	r									.	1.613

The principal active ingredients of this water are the sulphates of alumina and lime. It is slightly chalybeate and astringent

#### FAYETTE COUNTY.

The great durable fertility of the region of the limestone of of the Lower Silurian formation, in which Fayette and its neighboring counties are located, is owing greatly to the phosphoric acid and other fertilizing materials which exist in notable proportions in this limestone. As early as 1848, the attention of the present writer was attracted to this subject, and, in the Albany Cultivator of April, 1849, he published the result of two analyses of this limestone taken from a well in Lexington, with remarks on the great value of this ready source of these essential elements of fertility.

At that time, however, the methods of chemical analysis were by no means perfect, and the proportion of phosphoric acid in the limestone analyzed was, no doubt, greatly underrated at 1.35 per cent. In more recent times, many more analyses have been made by him of different layers of our limestone from different localities in Fayette county, and it has been found that, while this whole limestone formation contains phosphoric acid, it varies in proportion from about 1.40 per cent. up to more than 31. per cent.; and that the richer layers of the rock have, nowhere been found, as yet, of more than one foot in thickness, as is the case with the remarkable layer on the Newtown turnpike, described in Chemical Analyses of Kentucky Geological Survey, Vol. A., page 245.

The results of a few more analyses of the blue limestone of Fayette county, for the determination of the phosphoric acid, are given below:

No. 2946—B. Blue Limestone. Structure fine crystalline, granular; used on the turnpike, just beyond Lexington, on the Newtown road, opposite the brick-yard.

Proportion of phosphoric acid found, 7.935 per cent.

No. 2946—C. Blue limestone, on the same road, from farm of Mr. Cromwell, about 3 miles from Lexington.

Proportion of *phosphoric acid* found, 2.982 per cent.

No. 2946—D. Fossiliferous blue limestone, the softer, marly layer found between the harder layers. Winton farm, 7 miles from Lexington, on the same road.

Proportion of phosphoric acid, only 3.136 per cent.

An analysis of another thin layer on the side of South Limestone street, Lexington, near the Agricultural Experiment Station, made in the laboratory of the station, gave proportion of phosphoric acid, 29.73 per cent.

In the several thin layers found in this limestone on South Limestone street, examined at the Agricultural Station, the phosphoric acid was found to vary from 29.73 (d) per cent., as above stated, equal to 64.93 per cent. of phosphate of lime, down to 3.84 per cent., equal to 8.83 per cent. of phosphate of lime. A layer in the limestone at the city quarry was found to yield only 3.30 per cent. of phosphoric acid, equal to 7.21 per cent. of phosphate of lime.

The great body of the limestone, no doubt, contains a still smaller percentage.

#### FLOYD COUNTY.

No. 2947—Limonite Iron Ore, from Cane Branch of Beaver creek, on farm of Jefferson Robinson, Floyd county. Sample collected by C. N. Brown, September 5, 1888.

A reddish-brown limonite.

# COMPOSITION .- (Air Dried.)

										Per cent.	
Iron peroxide .	़				়					55.526	Equal 38,868 per cent of iron.
Alumina										10.568	
Lime carbonate										12.100	
Magnesia carbon	at	0				-				2.497	Mark Company and the same
Phosphoric acid				-			35.			2.046	Equal 0.893 per cent. of phosphorus
Silica				0			35	0.000		5.000	
Silica	*		•	8		•	•		٠	12.263	Ì
										100.000	

# JACKSON COUNTY.

- No. 2948—"CANNEL COAL," on Deer Fork of Horse Lick creek, Jackson county. Bed 33 inches thick, at the top of the conglomerate rock. Average sample from the whole face of the bed near the outcrop. Some little clay adhering to the sample. Collected by A. R. Crandall, June, 1889. A dull-looking cannel coal.
- No. 2949—Coal. "Jack Carpenter's coal," Horse Lick creek Jackson county, near the top of the conglomerate limestone. Average sample of the 36-inch face, near the outcrop. Collected by A. R. Crandall and Sullivan, June 25, 1889.
- No. 2950—COAL. Wells'. Sand Lick creek, Jackson county. Average sample from the whole 36-inch face, without parting, near the outcrop. Collected by G. M. Sullivan, June 28, 1889.
- No. 2951—COAL, on Mace Branch of Big Clover creek, Jackson County, 45 feet above the Subcarboniferous limestone. Average sample from the upper and lower benches. Collected by G. M. Sullivan, June 28, 1889.

Upper bench, 2 feet 7 inches; clay parting, 9\frac{3}{4} inches; lower bench, 2 feet 4 inches.

- No. 2952—COAL. Pete McDaniel's, on Joe's Branch of Raccoon creek, Jackson county. Average sample from the 3 feet face at the outcrop. Collected by A. R. Crandall, June 28, 1889.
- No. 2953—Coal of Jake Gabbard's bank, Indian creek, 3 miles below McKee, Jackson county. Geological position 95 feet above the Subcarboniferous limestone. Average sample. Collected by G. M. Sullivan, July 8, 1889
- No. 2954—Coal on Scott Dickson's land, Bell's Branc\_ of Indian creek, 1½ miles from McKee, Jackson county. Geological position 120 feet above the Subcarboniferous limestone. Average sample. Collected by G. M. Sullivan, August 14, 1889. Bed 36½ inches.
- No. 2955—Coal from "Turkey-foot Bank," owned by Captain Wilson, Turkey-foot Branch of War Fork of Station Camp, Jackson county. *Geological position* 85 feet above the Subcarboniferous limestone. Average sample. Collected by G. M. Sullivan, August 15, 1889.
- No. 2856—Coal from the "Shelton Brock" bed, near Collinsworth; 30 inches thick; on Rock Lick Branch, a tributary of Big Clover creek. Average sample. Collected by G. M. Sullivan, October 18, 1889.

GEOL. SUR.-2

COMPOSITION OF THESE JACKSON COUNTY COALS.—(Air Dried.)

Number in the Report.	2948	2949	2950	2951	2962	2963	2964	2965	2956
Hygroscopic moisture Volatile combustible matters	1.80 42.40 55.80	83.20 63.20	3.20 32.80 64.00	3.20 32.10 64.70	1.70 87.60 60 70	4.60 27.30 68 10	32.40 32.60 64.00	4.40 36.60 59 00	1.70 35.10 63.20
Totals	100.00	100.00	100.00	100.00	100 00	100.00	100.00	100 00	100.00
Total volutile matters	44.20 46.60 9.20	36.80 56.80 6.40	36.00 55.80 8.20	35.30 51.70 13.00	39.30 49.10 11.60	31.90 68.10 10.00	36.00 57.70 6.30	41.00 55.90 8.10	36.80 58.56 4.64
Totals	100.00	100.00	100.00	100 00	100.00	100.00	100 00	100.00	100.00
Character of the coke	dense spongy. very ltgrey. 0.645	spongy. It. bro'sh- grey. 0 494	spongy. grey. 0.782	dense spongy. durk- grey. 0.480	spongy. grey- brown. 2.307	dense friable. light grey. 0.522	dense spengy. purpl h- brown. 1.648	light spongy. lt.bro'sh- groy. 1.098	spongy. brown'h grey. 0.824

The only coal entitled to the name of cannel coal, among these samples, is No. 2948, which gave 42.40 per cent. of volatile combustible matters; the other samples yielding only from 27.30 per cent. in No. 2953, up to 36.60, No. 2955. This coal, No. 2948, leaving only 9.20 per cent. of ash, and a dense spongy coke, is purer than most other cannel coals.

No. 2955 gave only 3.10 per cent of ash, the lowest of ali the samples; the proportions running up to 10.00, 11.60 and 13.00, in Nos. 2953, 2952 and 2951. All except Nos. 2948, 2951, 2952 and 2953, would make very good coke; those named, as well as No. 2949, containing an objectionable large quantity of ash.

# JEFFERSON COUNTY.

No. 2957 (2) LIMESTONE from a bed in the canal at Louisville. Sample sent by Wm. Skeene & Co., July, 1889.

COMPOSITION, as received in powdered condition in a barrel.

Lime carbonate	•	٠			•		19			3.4	13		40	343			77.29
Lime carbonate	•3										9.						7.87
Lime sulphate	*0		Ç.								8.						.99
Phosphoric acid																	.38
Alumina and iron oxi	de					•											.33
Potash and soda																	tr. only
Silicious residue																	4.90
Alumina and iron oxi Potash and soda Silicious residue Moisture and loss .																	8.23
																25.200	
																	100.00

Quite a pure, somewhat magnesian limestone, which would make very pure white lime on calcination, but would hardly pay as a fertilizer, except on soils very deficient of lime.

# JOHNSON COUNTY.

#### COKE.

No. 2957 (3). Coke, from coal of Greasy creek. Sample sent by J. H. Northrop, August 9, 1889.

#### COMPOSITION.

Hygroscopic moisture 3.66 Voletile combustible matters	Total volatile matters 4.40
Coke	Carbon in the coke
100.00	100.00
Percentage of sulphur 0.570	6

# LAUREL COUNTY.

No. 2958—"Semi-Cannel Coal," from a 36-inch seam, located 1½ miles from Lily Station on the Louisville & Nashville railroad, Laurel county. The seam is 5 feet below water level. An average sample of the whole bed. Sent July 17, 1889.

Rather a dull-black, semi-cannel coal, breaking in laminæ, with some mineral carbon between, but no apparent pyrites; some little ferruginous incrustations on external surfaces.

No. 2959—COAL; from mine at Lily, on the Knoxville Branch of the L. & N. R. R., owned and worked by the Lily Mining & Manufacturing Company. Total thickness of the bed 44 inches. Sample sent by Rev. L. H. Blanton.

A bright, pure-looking coal, presenting, in some parts, "bird's-eye" structure; no mineral carbon apparent, but a little bright pyrites.

- No. 2960—Coal; From Lily, Laurel county. Average sample from two rooms of the 44-inch face of the bed, including a 2-inch "bone-coal," 8 inches from the top. Collected by A. R. Crandall, June 6, 1889.
- No. 2961—COAL; at "Winding-stair Gap," Laurel county. Average sample from the 38-inch face, driven in about 10 feet. Collected by A. R. Crandall, July 31, 1889.
- No. 2962—COAL; on "Indian Camp Branch," a tributary of Craig creek, 2 miles above Vox Post-office, Laurel county, on John Brewer's land; known as "Brewer's cannel coal," 32 inches thick. Average sample collected by G. M. Sullivan, October 8, 1889.
- No. 2963—COAL; 52 inches thick, on W. F. Bray's land, known as Bray's coal-bank, on Craig creek, near Vox Post-office, Laurel county. Average sample collected by G. M. Sullivan, October 7, 1889.
- No. 2963 (bis)—Carbonate Iron Ore; from Laurel river, near Lily. Sample by A. R. Crandall, June, 1889.

Found, by titration to contain 32.766 per cent. of iron in its 86.919 per cent. of carbonate, etc.

COMPOSITION OF THESE LAUREL COUNTY COALS.—(Air-dried.)

Number in Report.	2968	2969	2960	2961	2962	2963
Hygroscopic moisture Volatile combustible matters Coke	1.20 88.00 60.80	1.80 87.60 60.60	37.00 61.00	1.80 36.40 61.80	1.80 24.70 74.00	1.70 38.50 64.80
Totals	100 00	100.00	100.00	100,00	100.00	100 00
Total volatile matters	39.20 46.60 14.20	39.40 59.70 0.90	39.00 58.10 2.90	38.20 56.00 58.00	26.00 36.46 37.54	36 20 57.20 7.60
Totals	100.00	100.00	100.00	100.00	100.00	100.00
Character of the coke	dense spongy. lilac-grey. 0.797	dense spongy. white.	light spongy. grey. 1.085	light spongy. light pur- plish-grey. 1 650	pulverulent. grey-brown. 4.670	spongy. brownish grey. 1.785

Of these coals, it is probable the following may be used in the manufacture of coke, named in the order of their relative value for this use, viz: Nos. 2959, 2960, 2961 and 2963. The best of them is No. 2959, which gave 60.60 per cent. of dense spongy coke, containing 59.70 of carbon, and a very remarkably small proportion of white ash, viz: only 0.90 per cent. Nos. 2960, 2961 and 2963 containing rather more ash and sulphur, would, no doubt, yield good coke for all ordinary purposes. Nos. 2958 and 2962—containing, severally, 14.20 and 37.54 per cent. of ash material—while they may be used as ordinary fuel are, of course, proportionately less valuable than the purer coals.

#### LAWRENCE COUNTY.

COALS.

No. 2964—COAL; head of Straight Fork of Donathan creek. Average sample from the middle 15 inches. Collected by A. R. Crandall, August 27, 1888.

Irregularly laminated; some mineral charcoal on the laminæ, with a little of fine-grained pyrites.

No. 2965—Coal from the same bed as the next preceding sample. Average sample by Mr. Crandall, from the upper 27 inches.

Resembles the preceding sample; somewhat more laminated.

No. 2966—COAL from the same bed as preceding. Average sample of the lower 17 inches. By Mr. Crandall.

Resembles the two preceding samples, except that some portions break into thin lamellæ, resembling bituminous shale.

No. 2967—COAL; a lump from the bottom 6 inches. Collected by A. R. Crandall, August 27, 1888.

Fracture generally irregular, with irregular, shining surfaces; some portions laminated, with dull surfaces coated with mineral charcoal and fine grained pyrites.

No. 2968—COAL; on Jacob Peters' land, on Three-mile creek, Lawrence county. Average sample from the top 37 inches. Collected by A. R. Crandall, August 27, 1888.

- No. 2969—COAL; average sample from the bottom 13 inches of the bed next above described.
- No. 2970—COAL; "Big Coal," on Griffith's creek, Lawrence county. Sample collected by John R. Procter, August 29, 1888.

A lump with ferruginous coating on one edge. Breaks into irregular laminæ, which are coated irregularly with fibrous coal and fine granular pyrites.

No. 2971—"Cannel Coal" (White House). Mouth of White House, Lawrence county. Average sample from the whole bed, from 2 pits 200 feet in. Collected by A. R. Crandall.

A pure-looking cannel coal; no fibrous coal or pyrites apparent. Fracture imperfectly conchoidal.

COMPOSITION OF THESE LAWRENCE COUNTY COALS,—(Air-dried.)

Number in Report.	2964	2965	5966	2967	2968	2969	2970	2971
Hygroscopic moisture. Volatile combustible matters	3.00 36.40 60.60	6.00 32.40 61.60	36.00 61.00	39.00 58.00	2.20 38 80 64.00	1.80 40.50 57.70	4.00 39.40 56.60	1.20 41.80 57.00
Totals	100.00	100.00	100.00	100.00	100 00	100.00	100.00	100.00
Total volatile matters	39.40 52.40 8.20	38.40 57.40 4.20	39.00 47.00 14.00	42.00 6.00 6.00	36.00 48.00 16.00	42 30 43.50 14 20	43.40 46.60 10.00	48.00 46.00 11.00
Totals	100.00	100.00	100 00	100.00	100.00	100.00	100.00	100.00
Character of the coke	dense spongy. reddish grey. 0.961	dense spongy. reddish grey. 0.494	spongy. purplish gray. 4.714	light spongy. purplish grey. 6.450	light spongy. dk. pur'h grey. 1.922	spongy. dk. pur'h grey. 1.373	dense spongy. It. pur'h grey. 0.988	friable. brownish grey. 0,961

\*See end of Lawrence county for analysis of another coal

Remarkable differences are shown in the composition of the four samples of different layers in the same bed of coal; Nos. 2964-5-6-7—for example; the *Hygroscopic moisture* was found to be as high as 6 per cent. in the coal of the upper layer, No. 2965, and only 3. per cent in the 3 other layers. The volatile combustible matters, 32.40 per cent. in this upper layer, was 39. per cent. in the bottom 6 inches, No. 2967. The coke, 58 per cent. in this bottom layer, was 61.60 per cent in the upper 27 inches, No. 2965.

The carbon in the coke, 57.40 per cent. in the upper 27 inches, was only 47. per cent. in the lower 17 inches, No. 2966.

The proportions of ash in these different layers of the bed varied still more, being only 4.20 per cent. in the upper 17 inches (No. 2965), and as high as 14.00 per cent. in the lower 17 inches (No. 2966).

Sulphur was found in the proportions of 0.494 per cent. in the upper 27 inches in No. 2965, and as high as 4.714 and 6.450 in Nos. 2966 and 2967, the lower layers.

The coals of each layer examined yielded a spongy coke in the crucible. The largest ash proportions are in Nos. 2970, 2971, 2966, 2969 and 2968, which gave from 10. to 16. per cent. of ash on burning. While their cokes might be available for many purposes, they could not compete in the iron industry with cokes which contained less earthy material; more especially as these cokes contain also large proportions of sulphur.

For ordinary fuel, of course, these coals are all available, but their real value is always to be discounted by the quantity of ashes they leave behind in burning.\*

# LAWRENCE COUNTY.

#### IRON ORES.

No. 2972—Limonite Iron Ore, from a bed at See Gap. A dense, dark-brown ore, coated and mixed with ochreous ore. Collected by A. R. Crandall.

No. 2973—Iron Ore; Yellow Kidney, on Straight creek, fork of Three-mile creek, Lawrence county. Sample collected by A. R. Crandall, August 27, 1888.

<sup>\*</sup>See end of the following iron ores for another Lawrence county coal.

- No. 2974—Iron Ore; Limonite, irregularly cellular, with a little ochreous ore. Head of Three-mile creek, Lawrence county. Collected by A. R. Crandall, August 28, 1888.
- No. 2975—Iron Ore; Limonite, from same locality as the next preceding; similar ore. Collected by A. R. Crandall.
- No. 2976—IRON ORE; part of a nodule of dense Limonite, with soft ochreous material in the interior. Wallbridge's Station, on Chattaroi Railroad, Lawrence county.
- No. 2977—Iron Ore; brown Limonite ore. Peck's Station, Griffith's creek, Lawrence county. Collected by A. R. Crandall, August 28, 1888.
- No. 2978—IBON ORE; head of Griffith's creek. A dense, dark-reddish-brown Limonite ore. Collected by A. R. Crandall, August 27, 1888.
- No. 2979—Iron Ore; north of Louisa, Lawrence county.
  Said to be three feet thick. Sample sent from Frankfort by
  Mr. Procter, received September 7, 1888.
  A very hard oolitic ore.

COMPOSITION OF THESE LAWRENCE COUNTY IRON ORES. - (Air-Dried.)

	9800	0000	9074	997.6	2976	2977	2978	2979
Number in Report.	23.62	0 167	1107					
Iron peroxide	45.476	62.469	54 185	63.813	67.029 6.999	64.329	15.151	7.771
Alumina	trace.	trace.	trace.	trace.	trace.	trace.		89.
Magnesia carbonate	689.	.639	18,000	12,400	25.400	3.326 9.400	3.800	13.600
Silicious residue	9.98	13.860	11.426	14.384	10.904	16.009		16.179
Water and loss	100.000	100.000	100.000	100.000	100.000	100.000	100.000	100.000
	900	906 906	97 930	44.819	39.920	45.030	38 154	42.550
Percentage of iron	0.279	0.279	.223	.167	.223	.927	.167	101.

The metallic iron contained in these ores varies from 45.030 per cent. in No. 2977, down to 31.832 per cent. in No. 2972. The value of the first-named ore is much affected by its large proportion of phosphoric acid, equal to 3.325 per cent. The other ores contain this ingredient in proportions from only 0.167 in Nos. 2975, 2978 and 2979, up to 0.279 per cent. in Nos. 2972 and 2973. All of these ores are valuable.

No. 2971 bis—Coal from Whiteboro creek, sent by J. H. Northrop, of Louisa, Ky.

# COMPOSITION .- (Air-Dried.)

	Per	r Cent.					1	Per	Cent.
Hygroscopic moisture Volatile combustible matters .		4.00	Total volatile matte	rs .				•	37.80
Light spongy coke		62.20	Carbon in the coke Light-grey ash	::	:	:	:		57.80 4.40
		100.00							100.00
Percentage of sulphur		0.878							

A good coking coal.

# LOGAN COUNTY.

No. 2980—WHITE LIMESTONE (St. Louis Limestone), from Russellville, Logan county. Sample collected by D. W. Burgher. Received from Frankfort September 16, 1889. A light-grey, nearly white, fine-oolitic limestone.

# COMPOSITION .- (Air-Dried.)

											900				_	1		 				Per Cent.
Lime carbonate .																				٠		95.600
Magnesia carbonate	е																		়	្ន		95,600 .832
Alumina, iron oxid	e s	an	d	pb	109	pl	101	ric	80	eid												1.200
Silicious residue .				٠.		٠.														•		1 200
Magnesia carbonate Alumina, iron oxid Silicious residue . Moisture and loss													•								•	1.168
																						100.000

Quite pure limestone, which would yield very white fine lime for all ordinary uses.

# MARTIN COUNTY.

No. 2981—Coal from Wolf creek. Geological position: Lower coal. Average sample from the whole 36-inch face. Collected by A. R. Crandall, June 12, 1889.

# COMPOSITION .- (Air-Dried)

Hygroscopic moisture Volatile combustible matters	. 2.20	Total volatile matter	s.	e:				35.00
Spongy coke	. 65 00	Carbon in the coke Light-grey ash .			:	•	•	48.00 17.00
	100.00							100.00
Percentage of sulphur	. 1.813							

Its large proportions of ash and sulphur would reduce its value for the manufacture of coke, etc.

#### OHIO COUNTY.

No. 2982—MINERAL WATER, from the Worrell farm, near the line of the Falls of Rough Creek Railroad. Sample obtained by R. S. Triplett. Received from Frankfort July 23, 1889.

# COMPOSITION in 1,000 parts of water.

Ferrous sulphate Aluminum sulphate					1			÷	٠				7	$\mathbf{x}$	×					•					0.122
Aluminum sulphate																								. ]	.248
Calcium sulphate																									. 590
Magnesium sulphate																									.687
Calcium sulphate Magnesium sulphate Potassium sulphate .																									019
Sodium sulphate	•																								.017
Sodium chloride																								•	019
Silica	٠	٠	•	٠	٠	٠	*	٠	٠	٠	•	•	•	٠		٠	•	•	٠	•	•	•	•		.036
In 1,000 parts													*		×		0.00						1.00		1.738

A saline, astringent, chalybeate water.
[See Appendix for an Ohio county coke.]

#### OWSLEY COUNTY.

No. 2983—Coal; on Dick Reynolds' land, South Fork of Kentucky river. Average sample from 20½-inch bed, by G. M. Sullivan, July 27, 1889.

## COMPOSITION .- (Air-Dried.)

Hygroscopic moisture Volatile combustible matters	:	1.20	Total volatile matter	rs				•	•	56.00
Dense coke		44.00	Carbon in the coke Light-brown ash .	:	•	•	:	:	:	85.20 8.80
		100.00								100.00
Percentage of sulphur	•	0.577								

This appears to be a good cannel coal.

#### PIKE COUNTY.

#### IRON ORES.

No. 2984—IRON ORE, from the farm of J. Harvey Leslie, on Caney branch of John's creek. Sample collected by A. R. Crandall, August 1, 1889.

A yellowish brown Limonite ore.

No. 2985—Iron Ore; on the Hawk Fuller branch of Buffalo creek, on Spencer Boyce's land. Bed 28 inches thick; near the top of the hill. Sample collected by C. N. Brown, August 4, 1888.

A dark-reddish-brown Limonite ore, containing a little organic matter.

# COMPOSITION OF THESE PIKE COUNTY IRON ORES .- (Air-Dried.)

		N	uı	nt	er	s	in	R	ер	or	t.							2984	2985
Iron peroxide													٠					(a) 58.835	(c) 70.423
Alumina																		16.627	9.588
Lime carbonate .																		1.200	.500
Magnesia carbons	te																	.605	. 227
Phosphoric acid .																		(b) 2.178	(d) 1.279
Silicious residue .											•							 4.600	1.200
Lime carbonate. Magnesia carbons Phosphoric acid. Silicious residue. Water, etc., and l	088		•		•			•			•	•	•	•	٠	٠	•	15.955	16.883
																		100.000	100.000

<sup>(</sup>a) Equal to 41.184 per cent. of iron.

<sup>(</sup>c) Equal to 49.296 per cent. of iron.

<sup>(</sup>b) Equal to 1.492 per cent. of phosphorus.

<sup>(</sup>d) Equal to 0.868 per cent. of phosphorus.

# PIKE COUNTY COALS.

No. 2986—COAL; A. P. Bevins', on Bend branch of John's creek, Pike county. Average sample from the whole face—upper part 22 inches, lower part 27 inches. Collected by A. R. Crandall.

A bright-looking coal, breaking generally irregularly, with irregular shining surfaces, some small portions laminated, with very little mineral charcoal, and no pyrites apparent.

No. 2987—Coal: 6 feet coal, on the ridge between Caney branch and Little Bushy branch of John's creek. Coal about 625 to 650 feet above the creek. Average sample collected by C. N. Brown, August 23, 1888.

Coal mostly laminated; some mineral charcoal on the laminæ; no bright pyrites apparent.

No. 2988—COAL; on Pond creek. Average sample collected from the whole 62 inches of the face, by A. R. Crandall, August 8, 1889.

No regular parting in the bed.

- No. 2989—Coal; on Tug Fork. Average sample from the whole face of the bed. Collected by A. R. Crandall, August 9, 1889. Coal, 2 feet; shale, 15 inches; coal, 43 inches.
- No. 2990—Coal; on Turkey creek. Average sample of the whole 23-inch bed—cannel coal—badly weathered near the outcrop. Collected by A. R. Crandall, August 9, 1889.
- No. 2991—Coal; on Tug Fork, 1 mile above the mouth of Pond creek. Average sample of the 45-inch bed, excluding a 1-inch parting, 14 inches from the bottom. Collected by A. R. Crandall, August 9, 1889.

Ten inches of coal above 8 inches clay shale not included.

No. 2992—Coal; on Tug Fork, two miles above White Post. Average sample from the 6-feet bed (some little clay adhering) taken from near the mouth of the entry, by A. R. Crandall, August 11, 1889.

Three small partings in the bed.

COMPOSITION OF THESE SEVEN PIRE COUNTY COALS.—(Air-Dried.)

Number in Report.	2986	2987	2988	2989	2890	2981	2882
Hygroscopic moisture Volatile combustible matters Coke	34.60 63.80	28.80 68.20	1.40 27.20 71.40	32.00 66.00	2.10 44.70 53.20	1.20 38.40 60.40	1.40 86.60 62.00
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	36.20 57.40 6.40	31.80 57.20 11.00	28.60 65.00 6.40	34.00 60.00 6.00	46.80 45.50 7.70	39 60 55.00 5.40	38.00 55.80 6.20
Totals	100.00	100.00	100.00	100.00	100 00	100 00	100.00
Color of the ash	spongy. purplish gray. 1.703	donse friable. light grey. 0.590	spongy. very light grey. 0.494	spongy. light grey. 0.577	dense. brownish red. 1.606	light spongy. grey. 2.527	spongy. grey. 2.280

# PIKE COUNTY COALS—Continued.

- No. 2993—COAL; on Read branch of Big creek. Average sample of the 70-inch face, excluding a 2-inch parting near the base, and a ½-inch parting 3 feet 4 inches from the top. Collected by A. R. Crandall, August 8, 1889. (Some clay adhering.)
- No. 2994—COAL; Cane Hollow. Average sample from the 32-inch bed. (Splint and Block coal, 1½ inches; Bone coal at the middle.) Collected by A. R. Crandall, August 7, 1889.
- No. 2995—Coal; on Big creek. Average sample from the 37-inch bed, excluding two 1-inch partings. Collected by A. R. Crandall, August 7, 1889.
- No. 2996—Coal; in "Coal Hollow," on Big Creek. Average sample from the 35-inch face. Collected by A. R. Crandall, August 7, 1889.
- No. 2997—Cannel; Coal on "Big creek." Average sample from the 32-inch bed, excluding a parting 8 inches from the bottom, by A. R. Crandall, August 7, 1889.
- No. 2998—COAL; Shadel Pauley's coal, on Big creek. Average sample from the 4 feet bed. Collected by A. R. Crandall, August 8, 1889.
- No. 2999—Coal; at mouth of Peter creek. Average sample from the 62-inch bed, at the outcrop. Collected by A. R. Crandall, November 20, 1889. Some clay adhering to the sample.

GEOL. SUR .- 3.

COMPOSITION OF THESE SEVEN PIKE COUNTY COALS.-(Air-Dried.)

Number in Report.	2993	2994	2995	2096	2002	2998	2388
Hygroscopic moisture Volatile combustible mutters	1.20 33.20 65.60	1.60 28.30 70 10	2.60 34.80 63.10	1.60 32.80 65.60	1.40 42.00 56 60	1.60 35.80 62.60	68.80
Totals	100.00	100.00	100.00	100 00	100.00	100.00	100.00
Total volatile matters	34.40 61.00 4.60	29.90 55.90 14.20	36.90 57.30 5.80	34.40 56.40 9.20	43.40 46.40 10.20	87.40 58.00 4.60	31.20 69.70 9.10
Totals	100.00	100.00	100.00	100.00	100.00	100 00	100 00
Character of the coke	pulveru- lent. lt.brown'h grey. 0.631	dense. light grey. 0.631	spongy. grey brown. 0.604	dense spongy. light grey. 1.538	dense. lt.brown'b grey. 1.098	light spongy. brownish grey. 0.769	pulveru- lent, very light grey. 0.412

# PIKE COUNTY COALS—Continued.

- No. 3000—Coal; James H. Coleman's coal, head of Beech Fork of Peter creek. Average sample from the 63-inch bed. Collected by A. R. Crandall, November 17, 1889.
- No. 3001—Coal; Davis' coal, on Peter branch of Blackberry creek. Average sample of the 55-inch bed, by A. R. Crandall, November 20, 1889.
- Splint coal, and probably a little shaley, 8 inches from the top, near the outcrop.
- No. 3002—COAL; Thompson and Hatfield's, on Blackberry creek. Average sample from the lower five feet at the outcrop. Collected by A. R. Crandall, November 20, 1889. Some little clay adhering.
- No. 3003—Coal; at mouth of Holly Fork of John's creek. Harrison Field's coal. Average sample from the 6-feet bed, from the weathered face. Collected by A. R. Crandall, November 19, 1889.
- No. 3004—COAL; Elias Hatfield's, on Blackberry creek. Average sample of the 58-inch bed. Collected by A. R. Crandall. (A separate sample of the 6 inches near the top, marked B, was taken for test of ash, which left as much as 28.30 per cent. of ash.)
- No. 3005—COAL; Andrew Collins' coal, on Stinking branch of John's creek. Average sample from the lower 50 inches. Collected by A. R. Crandall, November 18, 1889.
- No. 3006—Coal; Dotson's coal, from near Jamboree postoffice, on Peter creek. Average sample from the 46-inch bed, including 2 inches of bony splint coal. Collected by A. R. Crandall, November 19, 1889.
- No. 3007—Coal; from Coon branch of Tug river, near the mouth of Blackberry creek. Average sample of the 39-inch outcrop. Collected by A. R. Crandall, November 20, 1889. Some clay adhering.

COMPOSITION OF THESE EIGHT PIKE COUNTY COALS.—(Air-Dried.)

Number in report.	3000	8001	3002	8008	3004	9000	3006	3007
Hygroscopic moisture	1.80 26.20 72.00	1.40 27.40 71.20	26.90 26.80 70.80	1.60 27.94 70.46	2.60 28.16 69.24	1.46 81.84 67.20	0.80 28.40 70.80	1.70 80.80 68.00
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	28.00 65.80 6.20	28 80 63.74 7.46	29.70 65.70 4.60	29.54 64.26 6.20	30.76 65.24 4.00	82.80 56.80 10.40	29.20 60.80 10.00	82.00 61.54 6.46
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Character of the coke	friablo. brownish grey. 0.549	denso spongy. brownish grey. 0.649	friable. reddish. grey. 0 494	dense friable. grey. 0.450	dense friable. roddish groy. 0 500	spongy. grey. 0.727	light spongy. dark grey. 0.719	dense friable. reddish grey. 0.521

In the above-described 22 coals from Pike county, the *Hygroscopic moisture* is found to vary from only 0.80 per cent. in No. 3006, up to 5.20 per cent. in No. 2999. This difference is probably partly owing to varying atmospheric conditions when the analyses were made. To get the real proportions of the other essential ingredients of the coals, the calculation should be made on the coals dried at a fixed temperature, say at 212° F., a calculation easily to be made of the above reported results.

Of these 22 coals, eleven, at least, are, very probably, good coking coals, as may be seen in the following summary of their essential ingredients, viz:

		1	Nι	ım	ıbe	er	in	R	ер	or	t.						Coke—per cent.	Carbon—per et.	Ash-per cent .	Sulphur-por
No. 2986				٠							٠	: : : : : : : : : : : : : : : : : : :				.	63.80	57.40	6.40	1.703
No. 2988																. i	71.40	65.40	6.40	. 494
No. 2989																.	66.00	60.00	6.00	. 570
No. 2991																	60 40	55.00	5 40	2 527
No. 2992															•	.	62 00	55.80	6.20	2 280
No. 2994											•						70.10	61.00	14.20	.631
No. 2995																.	63 10	57.30	5.80	. 604
No. 2996																.	65.60	56 40	9.20	1.538
No. 2998																.	62.60	58.00	4.60	. 769
No. 3001													39		2		71,20	63.70	7.46	.549
No. 3005			201		0				20	20			· ·		20		67.20	56 80	10.40	. 727
No. 3006												ů.			:		70 80	60.80	10 00	.719

It can be seen in this table that the *coke* obtained from these *coals* ranges from 60.40 per cent. up to 71.40 per cent. The *carbon* contained in these cokes is from 55. up to 65.40 per cent. The *ash* is from only 4.60 per cent., up to 14.20, in a single sample; and the *sulphur* from 0.494 per cent. up to 2.527.

In the cokes the percentages of ash and sulphur would, of course, be larger than is calculated in the coals.

It may be interesting to compare with these coals, the composition of some of the best coking coals of Pennsylvania as given in the special report (Second Geological Survey of Pennsylvania)—the Coke Report of F. Platt, 1875—as follows, page 120:

# ANALYSES OF THESE TYPES OF THE BEST COKING COALS OF PENNSYLVANIA.

													Connellsville (Pitts- burg seam)—per cent	Bennington (Miller's)—per cent	Broad Top (Barnet)	Coals (Miller's)
Fixed carbon .				:0		٠	٠						59.62	68.50	74.65	71.12 7.50
Ash												. 1	8.23	8.00	7.50	7.50
Volatile matters				•								. !	31 36	22.38	16.00	19.68
Sulphur	1 12	: 22	•	•	•	•	٠		•	•	2		. 784	1.12	1.85	1.70

# PULASKI COUNTY.

#### COALS.

- No. 3008—Coals; on Addison branch of Cumberland river. Average sample from the 54-inch face near the outcrop. Hard coal. Collected by A. R. Crandall. (Some clay adhering.)
- No. 3009—Coal; Rayburn's coal, Slipup mine. Average sample of the 40-inch face, collected by A. R. Crandall, July 30, 1889.
- No. 3010—Coals; on Big Lick creek. Average sample from the 31-inch face. Splint coal, the upper 4 inches cannel coal. Collected by A. R. Crandall. (Some clay adhering.)
- No. 3011—Coal; on Bear creek, semi-cannel or bone coal.

  Average sample of 4 inches at the top of the 42-inch bed.

  Collected by A. R. Crandall, August 1, 1889.
- No. 3012—COAL; on Bear creek. Average sample from the 42-inch face of the bed, exclusive of the semi-cannel at top (the next preceding sample). Collected by A. R. Crandall, August 1, 1889.
- No. 3013—Coal; from Happy Hollow. Average sample from the 37 to 39-inch bed, in room 60 feet in the main entry. Collected by A. R. Crandall, July 23, 1889.
- No. 3014—Coal; Edwards' mine, Cumberland river, 1½ miles below the mouth of Rockcastle river. Sample from the whole 30-inch bed, in the old entry. Collected by A. R. Crandall, July 30, 1889.

COMPOSITION OF THESE PULASKI COUNTY COALS.—(Air-Dried.)

Number in Report.	3008	3009	3010	3011	3012	3013	3014
Hygroscopic moisture Volatile combustible matters	2.40 35.60 62.00	2.00 84.80 63.20	2.80 33.80 63.40	1.00 34.30 64.70	2.40 34.00 63.60	1.40 36.20 62.40	3.40 84.40 62.20
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	38.00 54.40 7.60	36.80 55.20 8.00	36.60 53.60 9.80	35.30 44.30 20.40	36.40 54.20 9.40	37.60 56 80 5.60	87.80 57.20 5.00
Totals	100.00	100.00	100.00	100.00	100.00	100 00	100.00
Charactor of the coke	dense. brownish grey. 1.700	spongy. brownish grey. 1.550	dense friable. light grey. 2.580	dense. red.	spongy. grey. 0.467	spongy. nearly white. 1.043	dense. light grey. 0.687

Probably only 3 of these Pulaski county coals would yield very good coke, viz: Nos. 3009, 3012 and 3013, the ash percentages of which, severally, are 8.00, 9.40 and 5.60; the others, except No. 3014, which gives only 5.00 per cent., give from 7.60 to 9.80 per cent., one single sample, No. 3011, leaving as much as 20.40 per cent. of ash, which diminishes proportionately its value as fuel. In all except Nos. 3012 and 3014, which have only 0.467 and 0.687 per cent, severally, the proportions of sulphur are quite large, varying from 1.043 per cent. in No. 3013, up to 1.700 per cent. in No. 3008.

# ROCKCASTLE COUNTY.

#### COALS.

- No. 3016—COAL; from Coffee Branch, of Crooked creek. Average sample of 36-inch bed. Collected by G. M. Sullivan, May 31, 1889.
- No. 3017—COAL; "Spiery Coal," from near Livingston. Bed 84 feet above the Subcarboniferous limestone. Average sample from the face of the whole 33-inch bed. Collected by G. M. Sullivan, July 17, 1889.
- No. 3018—COAL; on Tarlton Clark's land, known as Clark's bank, Brush creek, 4½ miles from Brush Creek Station on the K. C. railroad. Geological position: 85 feet above the Subcarboniferous limestone. Average sample collected by G. M. Sullivan, from the 52-feet face, August 6, 1889.

#### COMPOSITION OF THESE ROCKCASTLE COALS.

Nur	n be	r	in	Re	po	ort									1	3016	3017	3018
	**			**						-		-						
Hygroscopic moisture					.:											3.80	4.20	4.40
Volatile combustible n	att	er	9													31.00	33.70	31,00
Coke				•	•	•	•				•				•	65.20	62.10	64.60
Totals		٠	•	•	÷	•		٠	•	•	Ö	je.		•		100.00	100.00	100,00
Total volatile matters																34.80	37.90	35.40
Carbon in the coke .					•		•	•		•	•					56.20	55.90	60.20
Ash					•	•	•	•	•	•	٠	•	•		•	9.00	6.20	4.40
Totals				٠	٠	•	•	•	•	•	•			•	•	100.00	100.00	100.00
Character of the coke						•	•									dense.	dense friable.	spongy.
Color of the ash																very lt.		light
																grey.	grey.	grey.
Percentage of Sulphur	37 3*						•		*		2			*3		0.412	0.968	0.494

No. 3018 is a remarkably pure, good coal, which would, no doubt, make very good coke. The others are not quite as valuable, as seen in their larger proportions of ash, yet are good fuel. The large proportion of moisture is, very probably, partly an accident of the season.

(It is easy to remove this irregular cause of variation by calculating the percentage in the samples, excluding the Hygrometric moisture, which has been expelled at the temperature of about 212° F.)

# WARREN COUNTY.

#### MINERAL WATERS.

No. 3019—Sulphur Mineral Water; from a well of Mr. D. E. Mercer, at Rich Pond. Well 105 feet deep, the first 50 feet in clay, the rest in the Saint Louis Limestone. Sample received from Frankfort, in a gallon-jug imperfectly sealed, October 9, 1888.

COMPOSITION; in 1000 parts of the water.

Hydrogen sulphide ga	6	( [	en	ai	ni	ng	. (	•					411	Cubic centimeters
Lime carbonate													0.3000	Hold in solution b
Magnesia carbonate.													.0667	Held in solution be carbonic acid.
Iron carbonate													.0920	carbonic acid.
Sodium chloride													6.5840	
Potassium chloride .													.0346	
Magnesium chloride													.6555	
Magnesia sulphate.													.1263	
Sodium, sulphate and													traces.	not estimated.
Silica													.0160	
Not estimated and loss													.0529	
Saline matters in 1000	p	ar	ts	0	f	the	, v	vat	ter	0.0	٠		7.9280	

A saline sulphur water, slightly chalybeate, perfectly wholesome.

No. 3020—MINERAL WATER; from a well bored 175 feet deep, on the public square, Bowling Green. Sample sent by Col. M. H. Crump. Received June 12, 1889.

#### COMPOSITION; in 1000 parts of the water.

Iron carbonate				়				:	٠	়	ુ				•	0.0030 )
Lime carbonate				•	•		•	•			ा. ः•ः	•	-	•		0.0030 Held in solution by
Magnesia carbonate																.0091) carbonic acid.
Sodium chloride .		•	•													.0397
Potassium sulphate			•											•		.0139
Sodium sulphate .																.2199
Lime sulphate				32												.0977
Magnesia sulphate							2									.2280
Silica						٠				:•						.0040
Saline matters in 10	00	p	ar	ts	of	t	he	w	at	er				٠	34	0.8393

A weak saline and very weak chalybeate water, wholesome for all ordinary purposes, but somewhat "hard." Will deposit most of its carbonate of lime on boiling.

# WARREN COUNTY.

#### LIMESTONE.

No. 3021—LIMESTONE; a block sent from the McLellan Stone Company's quarry, Rockfield, Warren county, Kentucky, by Mr. James Brown. Received November 21, 1888.

A compact, fine oolitic, light grey limestone, showing no fossils. Some few thin dark lines, not more than a twentieth of an inch thick, run horizontally through the block. These dark laminations were objected to by the architect.

# COMPOSITION (Air-dried.)

Carbonate of lime	•		•	•	•	•	٠	•			94,000
Carbonate of magnesia.						•		•	•	•	2.116
Iron peroxide, alumina,											
Silica											
Water, etc., and loss											.984

It was found that the stone did not cleave on the dark seams, the material of which is of a somewhat ferruginous composition, and probably would be as durable as the rest of the rock, and might even get harder by weathering.

# WHITLEY COUNTY.

COALS.

- No. 3022—Coal; near Pleasant View, on the Louisville & Nashville Railroad. Outcrop of a 31-inch bed. Sample sent from Frankfort by A. R. Crandall, February 7, 1889. "Will it coke?"
  - A bright, pure-looking coal.
- No. 3023—COAL; a selected sample of Jellico coal, lower portion of the bed. Procter mine, near Red Ash P. O. Brought by A. R. Crandall, March 15, 1889.
- A bright, pure-looking coal; bird's-eye structure in part; no fibrous coal or pyrites apparent.
- No. 3024—Coal; from same bed; upper portion contains some little fibrous coal.
- No. 3025—COAL; H. C. King's, near Mahan Station, Wall's creek. Average sample from the whole 42-inch bed. Collected by A. R. Crandall, May 2, 1889.
- No. 3026—Coal; Truman's coal, on Cane creek, near Jellico. Average sample of the whole bed: coal, 34 inches; clay shale, 1 inch; coal, 4 inches. Collected by A. R. Crandall, May 8, 1889.
- No. 3027—COAL; from Coal Bank Branch, Carr's Fork of Watts' creek, 3 miles from Rockhold. Average sample from the 48-inch face, without parting. (One inch of bone coal 2 feet from the top.) Collected by A. R. Crandall, May 9, 1889.
- No. 3028—Coal; from near "Pleasant View." Average sample from the whole 40-inch face, near the outcrop. Collected by A. R. Crandall.
- No. 3029—COAL; head of Cane creek. Average sample from the whole face, 3 feet 9 inches thick, including a 3-inch parting, 8 inches from the bottom. The lower part is splint coal; the lower, third and the upper benches are also. Collected by A. R. Crandall, July 17, 1889.

No. 3030—Coal; Worley's Branch of South Fork of Cumberland river; lower opening 63 inches. Twelve inches of the bottom is semi-cannel coal. Average sample from the outcrop of the whole face. Collected by A. R. Crandall, October 28, 1889.

A somewhat weathered sample. (Some clay adhering.)

- No. 3031—Coal; Jellico coal from near Williamsburg. Average sample from the whole face; 31 to 33 inches, 75 feet in the entry. Collected by A. R. Crandall. October 5, 1889.
- No. 3032—Coal; Splint coal, from near Brummitt's station on the Louisville & Nashville Railroad, near the Knoxville Branch. (Coal belonging to Mrs. Mary E. Varnon, of Stanford, Ky.) Sample sent from Frankfort, received January 6, 1890.

Some mineral carbon and fine-grained pyrites between the laminæ.

No 3032 (bis.)—"BIRD'S-EYE CANNEL COAL;" Halsey opening. Average sample of the cannel part (lower), 2 feet 4 inches; 6 inches of common coal above, under shale roof with layers of sandstone. The whole of the cannel coal not shown. Collected by A. R. Crandall, March 15, 1889.

Pretty bright "bird's-eye" coal in parts; portions more dull cannel. Some little bright scales of pyrites apparent.

COMPOSITION OF THESE WHITLEY COUNTY COALS.—(Air-dried.)

Number in Report.	3022	3023	3024	3025	3026	3027	3028	3029	3030	3031	3037	3032(bis)
Hygroscopic moisture Volutile combustible matters. Coke	n. e. n e. 59.46	2.40 38.10 64.50	2.00 36.40 61.60	4.60 32.80 62.60	1.60 37.20 61.20	1.80 87.00 61.20	2.40 34.80 62.80	4.80 82.70 63.00	5.40 33.40 61.20	2.00 84.76 63.24	30.80 86.80	1.00 48.00 56.00
Totals		100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Total volatile matters	40.54 55.76 3.70	85.50 63.10 1.40	38.40 60.30 1.30	37.40 59.00 3.60	38.80 58.40 2.80	38.80 57.80 3.40	37.20 58.80 4.00	37.00 56.00 7.00	38.80 58.06 3.14	36.76 58.44 4.80	33.40 61.46 5.14	44.00 48.80 7.20
Totals	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Character of the coke Color of the ash	very lt. spongy. grey brown. 0.453	dense spongy. It. grey brown. 0.494	dense spongy. It. grey brown. 0.714	dense. spongy.  It. bro'sh it. bro'sh grey.  Grey. grey.  0.742 0.796	spongy. lt. bro'sh. grey. 0.796	spongy. lt. grey. 0.961	dense spongy. light grey. 0.522	dense friable. bro'sh grey. 0.906	pulver- ulent. It. bro'sh grey. 0.446	light spongy. lt. bro'sh grey. 1.428	spongy. light grey. 0.659	very dense. greyish brown.

With the exception of No. 3022 and No. 3032 (bis) in which the ash is 7.00 and 7.20 per cent., severally, these coals are quite pure from earthy material; the percentages of ash falling as low as 1.30 per cent. in No. 3024, and reaching 4.80 per cent. only, in the next highest of the whole, No. 3031.

It is probable that good coke could be made from at least seven of the number, viz: Nos. 3022-3-4-6-7-31 and 3032.

All of them are good fuel, No. 3029 and 3032 bis being somewhat less valuable than the others, in the proportion of their larger ash percentages.

# WHITLEY COUNTY COKE.

No. 3033—COKE; made of slack coal, from the Jellico vein, Kensee mines. Sample collected by M. E. Thruston, October, 1888.

#### COMPOSITION.

	Per cent. Per	cent.
Hygroscopic moisture Volatile combustible matters	0.10 octal volatile matters	0.70
Dry coke	99.30 Carbon in the coke	4.60
	100.00	0,00
Percentage of sulphur	0.961	

The percentage of ash is rather large; otherwise this is quite good coke.

#### WHITLEY COUNTY.

#### CLAYS.

No. 3034—CLAY; from a bed 5 feet thick, on Indian creek. Sample collected by A. R. Crandall, March, 1889.

A friable, indurated clay of a buff-grey color. Refractory before the blow-pipe.

No. 3035—CLAY; "Jellico clay," Whitley county and Tennessee. A white, indurated clay. Refractory before the blow-pipe.

#### CHEMICAL REPORT.

#### COMPOSITION OF THESE WHITLEY COUNTY CLAYS.

					1	₹u	m	be	r i	n	Re	po	ort	•								3034	3035
Silica (of whi Alumina (wit	ch	3.	70	p	er	ce	nt	i	s f	ine	2 8	an	d)									59.10	84.7
Alumina (wit	h a	. 1	itt	lė	OX	id	e	of	ir	on	)											29.76	11.4
Lime `											٠.											trace.	trace
Magnesia																						.72	.6
Potash																-						3.86	1.5
Soda																2			23	-		trace.	.0
Water and los	88				·											•	•			ì		6.56	1.5
																					1	100,00	100.00

Both are good fire clays. The large proportion of silica in No. 3035 would make it quite refractory.

# MINERAL WATERS FROM WHITLEY COUNTY.

- No. 3036—MINERAL WATER; from a spring one mile south of Woodbine, Whitley county. Sample sent by Dr. A. Gatliff, at the request of A. R. Crandall, May, 1889.
- No. 3037—MINERAL WATER; from a well 1 mile south of Woodbine. Sample sent by Dr. A. Gatliff. Received May 20, 1889.

# COMPOSITION OF THESE MINERAL WATERS.

In 1000 parts of the water.

			N	u	ոե	ær	· i	n 1	Re	po	rl	•										3036	3037
Iron carbonate																						0.003	trace
Magnesia carbonat	6							332													. 1	.016	.01
Magnesia carbonat Lime carbonate .														Ç						•	٠.	trace	.03
Sodium chloride.								•														.044	.08
Sodium chloride . Potash sulphate . Magnesia sulphate Alumina and silice	•	•	•			٠	•	•					•			•	٠	•	•	•		.049	.02
Magnesia sulphate	٠.	•	•	*	٠			•	•	•	•	•	•	•	•		•	•	•			1. 1	
Alumina and silice	١.	٠	٠		•	*	•	•	•	•	•	•		•	٠			٠	•			traces.	.09
In 1000 parts of th																							0.46

No. 3036 is quite a weak chalybeate water, but both contain so small a proportion of saline matters that they scarcely deserve the name of "Mineral Waters."

#### TENNESSEE.

No. 3038A—COAL (Jellico); from East Tennessee Coal Company's mine, main entry, rooms Nos. 9 and 10. The lower part 30 inches thick. "Bone coal," 1 inch, running out in some places. Upper bench 14 inches. Sample collected by A. R. Crandall, March 15, 1889.

No.3038B-Sample of "Bone Coal."

No. 3039—Coal—"Mingo Mountain." Coal, 23 inches; shale, 2½ to 3 inches; coal, 44 inches; clay, ——. Sample collected by A. R. Crandall, April 17, 1889.

# COMPOSITION OF THESE TENNESSEE COALS .- (Air-Dried.)

Nu	ml	bei	r i	n	Re	po	ort										3038 A	3038 в	3039
Hygroscopic moisture	•																2.00	n. e.	4.40
Hygroscopic moisture Volatile combustible n	18	tte	78														36,00	n. e.	32.80
Coke																	62.00	n. e.	62.80
Totals						•			•					•	٠	•	100.00		100.00
Total volatile matters																	38.00	n. e.	37.20
Carbon in the coke .		٠															60.00	n. e.	56.60
Ash	•	٠	•	•	٠	•	•	٠	٠	•		•	•	•			2.00	14.70	6.20
Totals					•	•		•:	•	•	•	•	٠	*		•	100.00		100.00
Character of the coke			•	•	•	•				•	•				×	33	dense spongy.		III abic.
Color of the ash										٠						្	ltgrey		
Percentage of sulphur																	0.686	0.686	

Both, Nos. 3038 and 3039, are good coals. The bone coal excluded.

(See Whitley county No. 3035 for analysis of a clay common to Tennessee and Kentucky.)

Remarks.	Cannel coal; mine opposite Pineville.	On Right hand fork of Straight creek.	On Straight cr., near m. of Kettle Isl'd br.	Cockeroll's coal on Cannon creek.	Cockerell's coal Cannon cr., bed under san st.	On Bennett's fork of Red river.	On Bennett's fork near Tennessee line.	Cannel coal, Deer fork of Horse Lick creek.	Jack Carpenter's coal. Horse Lick creek.	Or Moss Branch of Rig Clover creek	Poto McDaniel's on Joes' br. of Raccoon cr.	Of Jake Gubbard's bank, Indian creek.	On S. Dickson's land. Bell's br. of Indian cr.	Capt. Wilson's "Turkey-foot bk." on Tfoot	br. of War Fk. of Station Camp.	"Shelton Brock" bed near Collinsworth.	Mine of Lily on Russellville br. L. & N.R. R.	From Lily	On " Winding Stair " Gan.	On Indian Camp hrInc. Brewer's land.	On W F Brav's land, on Craigs creek.	Head of Straight fork of Donathan creek.	Samo bed: unner 27 inches.	Same bed; lower 17 inches.	
Sulphur, per cent	0.370	1.126	3.845	2.193	0.577	0.439	0.766	0.646	0.494	0.182	90.400	0.59	9.00		_	0.824	0.00	1.010	1.000			0.00	494	4.714	
Color of the	170,700,700	rd'h-brown	dark-lilac.	pur'h-grey	pur'h-grey	It -reddish.	ltreddish.	v'ry lt.grey	ltbrgrey	grey.	ak -grey.	grey-or.	IIKIII-KIEY.	dn.spongy. pr ii-prowii	lt. spongy. Itbrgrey	br'sh-grey.	mac-grey.	Wille	grey.	it pur grey.	grey-or wil	or n-grey.	red ii grey.	red in grey.	Jun 11-Rich
Character of coke	pulveru'nt.	spongy.	lt. spongy	spongy.	spongy.	spongy.	re. spongy.	dn.spongy.	spongy.	spongy.	dn.spongy.	spongy.	dn. Irikole.	an.spongy.	lt. spongy.	spongy.	dn. spongy.	dn. spongy.	it. spongy.	it. spongy.	pulver lent	spongy.	dn. spongy.	dn. spongy	spongy.
Ash		8.20 74		6.40	5.70	9.40	9.5	9.70	6.40	8.20	13.00	3.8	30.00	08.9	3.10	4.64	14.20	9.6	2.50	20.5	37.04	9.6	S .	3.5	13.6
Carbon in coke	26.66	61.80	26.93	56.10	36.85	20.00	8.69	46.60	26.80	66.80	61.70	49.10	58.10	27.10	55.90	58.56	46.60	69.70	58.10	99	36.46	67.20	62.40	67.40	3.75
Total volatile.			37.60												41.00	36.80		•							
Coke			62.40												69.00	63.20								9.6	
Volatile com- bustible			35.54												36.60									32.40	•
Hygroscopic moisture .	0 0	200	8 52	8	1.20	8	1.20	3.40	8 8	8.20	8.20	1.70	4.60	3 40	4.40	1.70	1.20							9.00	
County	Bell.	:	::	:	:	:	: :	Looken		:	:	:	:	3	:	:	Laurel.	:	:	:	:	:	Lawr'ne	:	:
Number	128	920	2980	932	933	186	9:35	936	676	096	961	962	453	964	955	926	896	696	096	196	3965	5963	2964	2962	5960

TABLE I.-Continued

O	VIII VIII VIII VIII VIII VIII VIII VII
Remarks.	Same bed; a lump from the bottom. On J. Peters' land, Three mile cr., top 87 in. On Jacob Peters' land, bottom 13 inches. "Big Coal," Griffith's creek. Cannel coal, White House, m. White House. Whiteboro creek. Wolf creek. Wolf creek.  Jick Reynolds' land, S. fork of Ky. river. Wolf creek.  On Tug fork.  On Tug fork.  On Tug fork.  On Tug fork, 1 mile above Pond creek. Tug fork, 2 miles above White Post. On Read branch of Big creek. In Cane Hollow. On Rig creek. In Cane Hollow. On Big creek. In Cane Hollow. On Big creek. In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  In Cal Hollow. On Big creek.  Andrew Collins' coal, Biackberry creek.  Head of Beech fork of Peter creek.  Thompson & Hatfield's coal, Blackberry creek.  Heids' coal, m. of Holly fork, John's cr. Elias Hatfield's coal, Blackberry creek.  Andrew Collins' coal, Stinking br, John's cr. Bodson's coal, near Jamborce P. O., Peter cr.
Sulphur, per cent	6.450 1.922 1.373 1.373 961 961 1.703 1.703 1.606 1.60
Color of the	pur'h grey.  hro'h grey. light grey. light grey. light grey. light grey. light grey. rry.lt.grey. light grey. grey. grey. grey. light grey. light grey. light grey. light grey. light grey. libr'h grey. libr'h grey. libr'h grey. li.br'h grey.
Character of the coke	lt. spongy. lt. spongy. dn. spongy. dn. spongy. friable. lt. spongy. dense. spongy. dense. spongy. dense. lt. spongy. dense. lt. spongy. friable. dn. spongy. friable. dn. spongy. friable. dn. spongy. lt. spongy.
Ash	88888989999999999999999999999999999999
Carbon in the coke	284444748877788487778888888888888888888
Total volatile.	238444488888848884448888888888888888888
Coke	75.25.86 75.25.
Volatile com- bustible	28.28.28.28.28.28.28.28.28.28.28.28.28.2
Hygroscopic moisture	888888888888888888888888888888888888
County	Law nce.  Martin. Owsley. Pike.
Number	2967 2968 2968 2971 2981 2988 2988 2988 2988 2988 2988 2998 299

												Cl	H	CM	[I	CA	L		RI	EP	,01	RT
Coon br. of Tug r. near m. of Blackberry cr. Addison's branch of Cumberland river.	Rayburn's coal, Slipup mine.	On Big Lick creek.	On bear creek, 4 inches of lower 49 inches	on bear creek, sample of lower 12 mones.	From " nappy montow.	Care beauth of Chooked creek	College branch of Crown of Livingston	Spiery coal, iron near Livingston.	CHRIS DRINK, Drush Cl., 17 in. 110in Deathon.	Dear Fleas, View, L. C. IV. IV. IV. Will is cone.	Proceed mine near theu Asia 1. O. Della sain.	From Procter mine, upper part.	H. C. King s, near manan station.	Truman's coal, Cane creek, hear Jenico.	Coal Bank Dr. Carrs lork of Wates cleen.	From near rieusant view.	Trom nead of Carley creek.	Worley Dr. of D. lork of Culliberiand 11751.	Jeilico creek near Williamsburg.	Distriction of the contract of	DIEG s-eye cannel, marsey opening.	
-	1.550	N .	1.203	0.467	1.048	0.687	214.	906.	484	453	494	0.714	742	•	•	•	3	440	1.428	000	1.20/	
dn.friable.	spongy. bro'h grey.	du friable. light grey.	dense, red.		-		dense. vr. It. grey.	dn.friable, it.br n grey	spongy. light grey.	vr.lt.spngy greybrown	dn.spongy, it.grey bro.	dn.spongy. It grey bro.	dense. It.grey bro.	spongy. It.br'h grey	spongy. light grey.	dn.spongy. light grey.	dn. friable. br h grey.	pulver lent it.br h grey	It. spongy. It. br'h grey	spongy. Inght grey.	very dense, grey'n bro.	
	8.8	8.6	20.40	9.40	9.60	8.9	8.6	9.70	4.40	8.70	1.40	1.30	3.60	2.80	3.40	8.5	3.5	3.14	8.80	6.14	7.20	
61.54	55.20	53.60	44.30	54.20	26.80	67.20	56.20	55.90	60.20	55.76	63.10	60.30	20.09	58.40	24.80	~	_	_			_	
88.80	36.80	86.60	85.30	36.40	37.60	37.80	34.80	37.90	36.40	40.54	35.50	38.40	37.40	38.80	88.80	87.20	87.08	38.80 88.80	36.76	33.40	44.00	
68.00	68.20	63.40	64.70	63.60	62.40	62.20	65.20	62.10	64.60	59.46	64.50	61.60	62.60	61.20	61.20	62.80	63 00	61.20	68.24	99.99	26.00	
30.30	34.80	33.80	34.30	34.00	86.20	34.40	31.00	88.70	31.00	п. е.	33.10	36.40	32.80	87.20	37.00	34.80	32.70	33.40	34.76	30.80	43.00	
1.70	2.00	2.80	8.	2.40	1.40	8.40	3.80	4.20	4.40	n. e.	2.40	2.00	4.60	1.60	1.80	2.40	4.30	6.40	2.00	2.60	8.	
Pike.	,,,	:	:	:	:	:	Rock'tle.	:	:	Whitley.		:	:	:	:	:	:	:	;	,,	;	
3007	3008	3010	3011	3012	8013	3014	3016	3017	3018	3022	8053	3024	8025	8026	3027	3028	3059	3030	3031	3035	bis.	

# TENNESSEE COALS.—(Air-Dried.)

2.00 dn. spongy. lt. grey br686 Jellico coal, East Ten. Coal Co. mine686 Jellico coal, sample of the Bone Coal6.20 dn. friable. lt.br'h grey .824 Mingo Mountain.
Jellico coal, East T. Jellico coal, sample Mingo Mountain.
.686 .686 .824
lt. grey br. lt.br'h grey
dn. spongy.
14.70
80.00 n. e. 56.60
38.00 n. e. 37.00
62.00 n. e. 62.80
36.00 n. e. 32.80
2.08 4.40
3038n 3038n

PENNSYLVANIA COKING COALS-(Air-Dried)

Remarks.	
Per cent. of sulphur.	.784 1.12 1.86 1.70
Color of the ash	
Character of the coke	
Ash	8.23 7.50 7.50
Carbon in the coke.	59.62 68.50 74.65 71.12
Total volatile mat- ters	
Coke	
Volatile combustible matters	8888
Hygroscopic moist- ure	31.16.19.61
County	
Number	

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Remarks.	Made from the "run of the mine." Made of slack coal, Jellico vein, Kensee mines. From coal of Greasy creek.
Hours of cok-	22
Owner of the coke	grey-brown. Pine Mt.Ir.C. grey-brown. light-grey.
Color of the	grey-brown. grey-brown. grey. light-grey.
Sulphur ,	0.492 .961 .576 1.648
Ash	4.40 14.70 7.00 19.60
Carbon	99.30 84.60 88.60 77.40
Moisture at red heat	0.70 0.70 4.40 1.70
County	Bell. Whitley Johnson. Obio.
Number	2937 3083 2957(3 3040

TABLE III.-IRON ORES-(Air-Dried.) (Limonite.)

Remarks.	Oriskany ore, Straight creek.	Lock's iron ore, Kettle Island branch. From Cane branch of Beaver creek.	Bed at See Gap.	Straight creek fork of Three-mile creek.	Head of Three-mile creek.	Same locality.	Walbridge's station on Chattarol rallroad.	Peck's station, Griffith's creek.	Head of Griffith's creek.	North of Louisa.	J. H. Leslie's, Cane branch of John's cr.	On Hawk Fuller br. of Buffalo creek, on Spencer Boyce's land.	
Sulphur	0.412	1.098		1	No.	t e	est	im	at	ed			
Phosphorus .	0.279	1.340	.279	.279	. 553	.167	553	.927	. 167	.167	1.492	898.	
Iron	42.163	49.784 38.868	31,832	36.735	87.930	44.319	39.920	45.030	38.154	45.550	41.184	49.596	
Moisture and loss	19.500 19.500 42.163	932	9.985	13.860	11.426		10.30	16.009	9.308	16.179	15.955	16.823	
Silicious residue	19.500	16. 5.000	31.000	28.800	16.00	12.400	25.400	9.400	3.800	13.600	4.600	1.200	
Phosphoric acid	0.639	3.110	639							.383	2.178	1.279	
Magnesia car- bonate	0.180	2.160	.378	trace.	454	.076	.156	.151	2.052	.681	.605	.227	
Lime carbon-	trace.	1.900	trace.	:	:	:	:	:	14.800	9.	1.200	.500	
Alumina	5,148	13,680	9,525				6.63				6.627	9.538	
Iron peroxide.	60 023	71.120									58.835		
County	Bell.	Flord	Lawnce	:	:	:		:	,	;	Pike.	3	
Number	2938	2989	2972	2973	2974	2975	2976	2977	2978	2979	2984	2985	

CARBONATE IRON ORE.

3015. Laurel. Sample from near Lily, on Laurel river, containing 32.766 per cent. of iron.

TABLE IV.-LIMESTONES-(Air-Dried.)

	OHEMICAL
	Phosphoric acid in various thin layers in the Blue Limestone. The highest proportion yet found is about 31 per cent.  From bed in the cannel. St. L. Limestone from Russellville.
County	Fayette.
Moisture and loss	8.235 1.168
Silicious residue.	n. e
Soda	n. c. ;; ;; ;; tr.only. n. e. n. e.
Potash	n. e. ;; ;; ;; tr.only. n. e. n. e.
Phosphoric acid.	7.935 2.982 8.136 29.73 8.80 0.383 n. e.
Iron peroxide .	0.380 0.700
Alumina	0.380 0.700
Magnesia carbonate	n. e.
Lime carbonate.	77.290n 95.600
Number	2746B 2746C 2746D 2746E 2746F 2957 2957 8021

(a.) Also 0.992 per cent. of lime sulphute.

# APPENDIX.

# OHIO COUNTY-Continued.

No. 3040—Coke; sent by the Hon. R. S. Triplett, received January 24, 1890. Made of coal No. 1, B. stratum 54 inches. thin shale parting 4 inches, and upper coal 15 inches. From line of the O. F. G. R. R., Ohio county. Coke made on the ground. Block coal.

#### COMPOSITION .- (Air-Dried.)

	Per cent.				er cent.
Hygroscopic moisture Volatile combustible matters . Coke	1.70 Total volatile matters. 1.30 Carbon in the coke 97.00 Light-grey ash	:		•	. 3,00 . 77,40 . 19,60
	100.00				100.00

Percentage of sulphur . . . . . 1.648

The large percentage of ash and rather more than a good average of sulphur lessen the value of this coke.

# THE CONDITIONS OF SULPHUR IN COALS.

It has been believed for some time past that a portion of the sulphur in coals is in a free or uncombined condition, while another portion is in some unknown form of organic combination. It is a well-known fact that vegetable and other organic substances, when undergoing decomposition in water containing sulphates of iron, lime or magnesia, etc., may reduce the salts to sulphides, with production of Hydrogen sulphide in the case of lime and magnesia sulphates; conditions which may have been present during the formation of coal.

As hydrogen sulphide (H<sub>2</sub> S) is decomposed by contact with the air; the sulphur being deposited; it can easily be seen that coals may frequently contain free sulphur, or some form of sulphuretted organic compound or mineral charcoal from this cause.

In 1873-4, the present writer analyzed, for free sulphur, a fibrous coal or mineral charcoal from Muhlenburg county, Ky., by treating it with purified carbon bi-sulphide, and found in it a notable quantity, also sulphate of lime. (See reports of Geological Survey of Kentucky, New Series, vol. 1, lower paging 287.)

Subsequently he learned that Professor T. G. Wormley, early in 1870, then chemist to the Geological Survey of Ohio, had ascertained that the proportion of sulphur found in coals was frequently greater than sufficed to combine with the iron present in them as bi-sulphide, a fact which has been verified in many cases by other chemists, and he asked the question, "does sulphur combine in any form with the carbon of the coal?" (Report of the Ohio Geological Survey, vol. 1, page 360.)

The process above described, for separation of sulphur, was used by the present writer in six samples of coal No. 7, from Willard, Carter county. Collected by Mr. Geo. Gibbs, at the instance of Prof. James P. Kimball, Ph. B. and Mr. H. W. Bates, in 1879, and in each case uncombined sulphur was obtained by him, varying in quantity from 0.012 to 0.060 per cent.

In addition to this *free sulphur* found in some coals, a certain amount of sulphur may exist in them in chemical combination with the coal itself.

On boiling portions of the samples of Carter county coals above mentioned, in a watery solution of carbonate of soda, to ascertain the amount of sulphuric acid present in them, the filtered decoction thus obtained was highly colored, in some samples, from the presence of organic matter, especially that from a greatly weathered sample from Buzzards' Roost Entry, Willard.

To ascertain the nature of this organic matter, twenty grammes of this coal, finely pulverized, air-dried, were boiled in a solution of twenty grammes of carbonate of soda, and the dark-brown decoction thus obtained was filtered and acidu-

lated with Hydrochloric acid, which threw the organic matter down as a dark-brown precipitate, which, when dried at 212° F., weighed 0.15 of a gramme, equal to 0.70 per cent. of the air-dried coal, which quantity might, probably, have been somewhat increased by further digestion of the coal in the alkaline solution.

This organic or bituminous matter, resembling humus in some of its properties, brownish-black when dry, was only slightly soluble in strong alcohol; insoluble in ether, and bisulphide of carbon, and burned without leaving a notable quantity of ash.

Examined for its proportion of sulphur by oxidation in a mixture of nitric acid and Chlorate of Potash, etc., and estimating it as barium sulphate, it was found to contain 0.0146 per cent. of its weight of sulphur. It is highly probable that in the unweathered coal, the combined sulphur is in much larger proportion, as it is now well-known that, when exposed to the air, the combined sulphur oxidates, and may be washed out by water as sulphuric acid and soluble sulphates. This formation of acid, under these circumstances, may be observed on the "Ebonite" plates used in some forms of electrical machines, the surfaces of which, on exposure to the air, become covered with acid moisture, caused by the oxidation of the sulphur of the highly vulcanized caoutchouc of the plates by the oxygen and moisture of the atmosphere.

Caontchouc, which is a compound of carbon and hydrogen (C<sub>4</sub> H<sub>7</sub>) readily combines with sulphur, which greatly modifies its properties, as is exemplified in the numerous forms of what are denominated vulcanized rubber, in which it is in proportions of 2 to 10 per cent. in the softer varieties, and in the harder varieties the sulphur is not only in chemical combination with the caoutchouc, but also a large proportion is held, as it were in mixture, in that substance, up to 30 per cent. in some cases. As is well known, this combined sulphur gives toughness, durability and hardness to the native hydro-carbon caoutchouc, according to the quantity present.

The presence of a certain quantity of sulphur in some cannel coals, may be the cause of their remarkable toughness; thus the celebrated Breckinridge coal of Hancock county, remarka-

ble for its toughness, was found, on analysis (see Chemical Report in Volume 2, O. S., page 212 of Kentucky Geological Survey Reports), to contain as much as 2.476 per cent. of sulphur. Other coals, such as tough so-called "Block coal," may possibly owe their toughness to the same cause.

A remarkable fact, reported by Mons. Th. Sidot (Comptes Rendus, LXX., page 605, March, 1870\*), illustrates the influence of sulphur on the tenacity of charcoal or carbonaceous substances, which, when heated in an atmosphere of carbon di-sulphide, gained in weight and tenacity by combination with sulphur, as follows:

Placing a stick of wood in a porcelain tube, displacing the air by the vapor of carbon di-sulphide, and then gradually heating it to redness, keeping up this heat for an hour, the wood, on cooling, was found to have changed into a coal very different from ordinary charcoal, even when very soft wood or cork was used. Especially was it highly sonorous, resembling in this respect the most sonorous metals, such as steel, silver, aluminum, etc., or glass.

Subsequently, he heated in this manner, a small oak-wood bell, which thus became as sonorous as a metal bell of the same dimensions. The coal obtained by this process differed from common charcoal in its elasticity, and its great conductability of heat and electricity, so that he recommended it for the carbon cylinders of Bunsen's battery. Moreover, he found that it far excelled the gas retort carbon in the intensity of light which it emitted under the influence of an electric current, and, therefore, would be admirably suited for electric lighting; and he states that he obtained the same result with flax, hemp, cotton, paper and silk; thus, to some extent, anticipating and improving on Edison in his use of paper charcoal for this purpose.

This peculiar charcoal, prepared from wood, had a metallic lustre on its surface, was more dense than wood charcoal, and did not notably absorb gases. It acted like a metal when heated, becoming gradually white hot without taking fire in one spot like charcoal, and cooled quickly when removed from the fire, because of its good heat conducting power.

<sup>©</sup> Copied into Polytechnic Journal, Dingler, CXCVI-472, and in CXCV-34 7.

This change of charcoal into a dense coal, was not owing entirely to the absorption of sulphur, but most probably, also to the chemical union of hydrogen with the organic matter during the process of charring; for M. Sidot found that the same result could be obtained by charring the material in an atmosphere of wood alcohol vapor, or of a hydro-carbon. This is rendered the more probable, he states, by the circumstance, that when the wood is heated in a porcelain tube, in a current of the vapor of methylic alcohol the interior of the tube becomes lined with light, shining coal filaments.

These experiments of Mons. Sidot, as well as the vulcanization of caoutchouc, illustrate the tendency of sulphur to unite with compounds of hydrogen and carbon, as do also the mineral called Tasmanite, and numerous organic compounds. The singular mineral, Tasmanite, so-called from the locality, Tasmania, in which it is found, a brownish-red mineral, in translucent scales, is composed of carbon, 79.3 per cent; hydrogen 10.4 per cent., oxygen 4.9 per cent, and sulphur 5.3 per cent.

Some of the essential constituents of animals and vegetables have somewhat analogous compositions; thus, the dried albumen of the white of eggs, consists of carbon 54.3 per cent., hydrogen 7.1 per cent., oxygen 21.0 per cent., nitrogen 15.8 per cent. and sulphur nearly 2.0 per cent.; and the albumen of animal blood, the fibrine of their muscles, the curd of their milk, have similar compositions.

No animal could exist without the essential sulphur in the organic constituents of its fluids and solids, and these seem to be derived from the vegetables which primarily or secondarily constitute their food, in all of which sulphur is found in organic composition with hydro-carbons, nitrogenous or non-nitrogenous; hence, it is not wonderful that it is to be found in organic combination in coals, which originate mainly from the partial decomposition of vegetable bodies, during which decomposition sulphur may also be derived from other sources above referred to.

"A sample of cretaceous anthracite brought from Peru, yielded by analysis, according to Dr. Percy,\* 10.35 per cent

<sup>\*</sup> Percy on Fuel. Appendix, page 567.

of sulphur, and only 3.75 per cent. of ash. Of this excessive proportion of sulphur, only 3.52 per cent. was evolved on heating the coal for an hour without access of air at an intense white heat. (!) The ash consisted principally of silicate of alumina, and contained only 0.018 per cent. of sulphur. The retention, by the carbonaceous matter, of so large a proportion of sulphur in excess of any possible combination in the ash points, as suggested by Dr. Percy, to the combination of the larger part of the sulphur in this coal with its organic constituents. Another example, leading to the same inference, is given by Dr. Percy.

"A miocene coking coal from New Zealand, yielded 2.50 per cent. of a white ash, and proved to be free from any sulphate soluble in hydrochloric acid; hence, the suggestion that the sulphur may exist in the coal in the same state in which it exists in albumen, fibrin, hair, etc." \*

These facts throw much light on the generally large proportion of the sulphur of coals which is retained in the cokes made from them, even when their ash material shows no equivalent amount of iron, lime or magnesia.

Sulphur in coals may exist in several different conditions:

- 1. It undoubtedly exists in them in a free or uncombined condition, especially in unweathered coals.
- 2. It is found in coals combined with iron as sulphide or bi-sulphide—different forms of pyrites.
- 3. It is present in organic combination with the hydro-carbon of coals.
- 4. It may be present, especially in weathered coals, in the form of free sulphuric acid or of sulphates of iron, lime or magnesia.

These facts were measurably verified in the analyses made in 1879 by the present writer, of the samples of coals from Carter county, Ky., referred to above; samples which were carefully collected from the coal No. 7, Coalton coal, near Willard, by Mr. Gibbs, with special reference to this investigation.

The following table of the results of the analyses of four

<sup>\*</sup> Prof. Jas. P. Kimball, "Relations of Sulphur in Coal and Coke," etc., 1879.

samples, two of highly weathered, and two of the unweathered coal of the interior of the mines, illustrate the effects of the exposure to the atmospheric agencies to which the first named samples had been subjected.

COMPOSITION, equally air-dried, of the four Carter County Coals.

												of two unweath- ered samples	of two weathered
Hygroscopic moisture Volatile combustible matters		20		•				20			,	5.98	7.08
Volatile combustible matters											.	34.92	33,05
Fixed carbon in the coke .												53.82	53,42
Ash											.	5,28	6 45
Total sulphur in the coal		•									 . [	2.241	1,669
Total sulphur in the coke .		•						*1				1.336	1,036
Sulphur lost in coking Free sulphur extracted by car									*		. 1	.874	.633
Free sulphur extracted by car	·be	n	bi	-81	ulı	oh	ide				. 1	.036	.022

It can be seen in this table that, while the relative proportions of volatile combustible matters seem to have been only slightly reduced, and the fixed carbon in the coke but little altered by the long process of weathering; the proportion of ash is notably increased, an effect which may be partly owing to the formation of earthy sulphates in the coal, partly to the removal by oxidation of the organic matter of the coal. But the most marked change is to be seen in the proportion of sulphur, which appears to have been diminished in all its forms by the weathering process, partly, no doubt, by its slow oxidation by the atmospheric agencies, and its escape in the form of soluble sulphates washed out by water, and partly as volatile sulphurous acid.

Sulphur in the form of sulphates of iron, alumina, lime and magnesia is found in coals, more especially in coals which have been long exposed to atmospheric agencies—air and water—the results of the oxidation of the sulphur and sulphides contained in them.

To ascertain the presence and relative proportions of these

sulphur compounds in the weathered and unweathered samples of the coals from Willard, Carter county, collected by Mr. Geo. Gibbs at the instance of Prof. Kimball and Mr. H. W. Bates, the following process was adopted: twenty grammes each of the several finely powdered samples were digested for 48 hours in pure distilled water, kept at the temperature of about 200° F. during the daytime. The solution was filtered and the coal on the filter thoroughly washed with distilled water.

The watery solutions, each divided into two portions, were analyzed, the one to ascertain the amount of sulphuric anhydride, the other for the several bases. The coals on the filters transferred to beakers, were digested for four hours in a weak solution of hydrochloric acid, to dissolve out any basic sulphate of iron or sulphate of lime, etc.; and the amount of sulphuric acid ascertained by barium chloride, etc. The results of these examinations are given in the following table, in percentages of the coals.

## COMPOSITION.

		NUMBER	AND CHARA	CTER OF TE	NUMBER AND CHARACTER OF THE SAMPLES.	
	2.	5.	1.	4.	3.	6.
	Unweathered	Unweathered	Weathered	Weathered	Unweathered	Unweathered
Sulphuric anhydride in the watery solution	0.195	0.069	0.266	0.470	0.139	0.073
Sulphuric anhydride in the weak acid solution	020	.044	.092	.034	.058	.028
Percentage of sulphur equal to the total sulphuric anhydride	.103	.041	.143	.201	620.	.040
Iron sulphate in the watery solution	.243	.074	.465	.026	.227	.103
Lime sulphate in the watery solution	.039	.025	.032	999.	.032	.026
Magnesia sulphate in the watery solution	015	.007	.013	660.	600.	900
Alumina sulphate in the watery solution			a trace;	not estima ted	ted.	

The practical significance of this table is in the relative proportions of sulphur found in the weathered and unweathered coals.

The unweathered coals, 2 and 5, gave, severally, 0.103 and 0.041 per cent. of sulphur; the mean of the two being 0.072 per cent; while the weathered samples, 1 and 4, gave 0.143 and 0.201 per cent.; the mean being 0.172 per cent.

Moreover, the other two unweathered samples, Nos. 3 and 6, gave only 0.079 and 0.040 per cent. of sulphur severally, in their sulphates, their mean proportion being only 0.0595 per cent.

All going to show, that a portion of the sulphur of the coals had been oxidated into sulphuric anhydride in the weathering process, with the production of sulphates, which, being more or less soluble, could be washed out by water, or might remain in the coal as more or less soluble salts.

Not only are sulphates of iron, alumina, lime and magnesia found in weathered coals, but also in the beds of clay found associated with the coal-beds, causing a saline efflorescence on drying, and affecting the quality of the material. Hence, workers in clays find it advantageous to wash them in preparation for some uses. It is recommended, that in the analysis of coals, for the determination of their sulphur, the pulverized coal should be first digested in a weak mixture of hydrochloric acid and water, to remove and estimate the sulphur contained in the form of acid or in the sulphates.

The organic or ultimate analyses of the two samples of the unweathered Carter county coal, Nos. 2 and 5, made at the time stated, was as follows:

COMPOSITION, Dried at 212° F.

										No. 2	No. 5.
	-									Per cent	Per cent
Carbon				28				2		73.842	72.853
Hydrogen								20	.	4.946	4.814
Sulphur		•		23		8				2.262	2.523
Oxygen and nitrogen, by difference										13.413	13.370
Ash			•		•	7.0	•			5.537	6.440
										100,000	100,000

The considerable proportion of sulphur which may remain in cokes, even when they have been subjected to a very high temperature in the oven, may, according to the facts and analyses above given, be partly in the condition of organic combination with the hydro-carbon of the coke, and partly in combination with iron, lime and magnesia, as partly decomposed iron sulphide and lime and magnesia sulphates. In the smelting of iron, the lime and magnesia sulphates are believed to exert no injurious influence on the iron, they being removed in the slag without undergoing decomposition, but it is probable that sulphur may be transferred to the iron both from the organic sulphurous compound and the partly decomposed iron sulphide and sulphate present.

GEOL. SUR .- 5.

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